

Phase transition

in finite systems

Francesca GULMINELLI and Philippe CHOMAZ

■ Finite systems

- ◆ Zeroes of the partition sum
- ◆ Bimodal event distributions
- ◆ Negative heat capacity
- ◆ Abnormal fluctuations

■ Experimental results

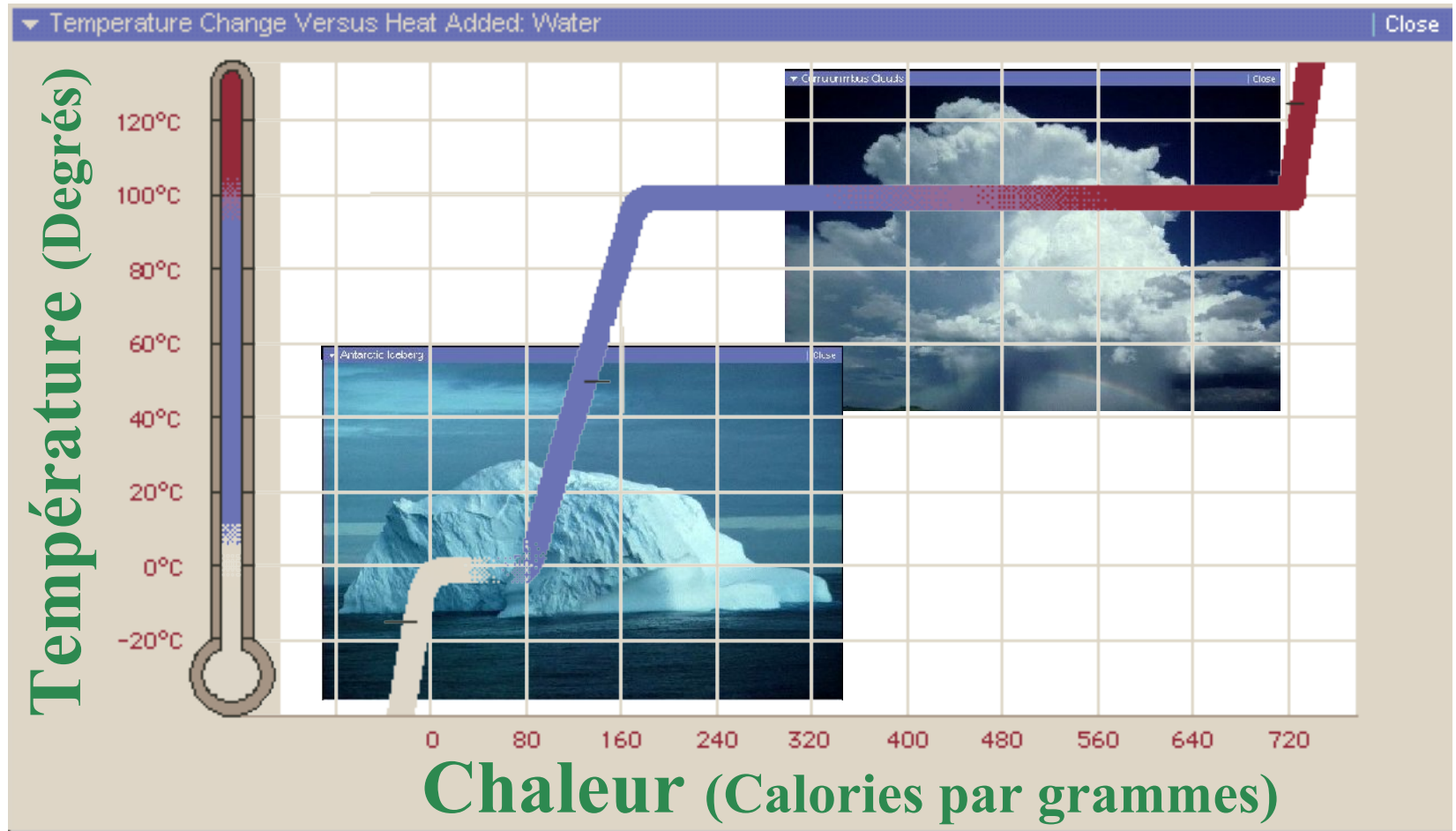
- ◆ Melting of Na clusters
- ◆ Superfluidity in atomic nuclei
- ◆ Fragmentation of nuclei
- ◆ Fragmentation of H-clusters

Introduction

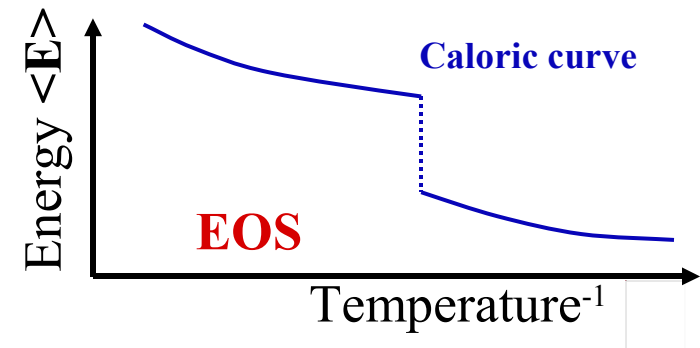
Phase transition

Anomaly in thermo

Transition de Phase



Phase transition in infinite systems



● Ex: first order:
discontinuous EOS:

Phase transition in infinite systems

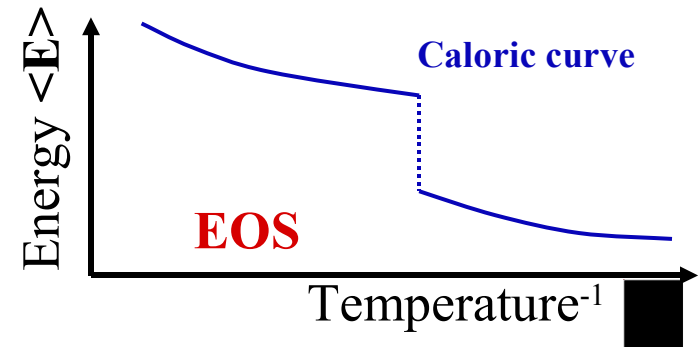
- Thermodynamical potentials: $e.g. F = -T \log Z$
non analytical at

L.E. Reichl, Texas Press (1980)

$$Z = \sum_{(n)} e^{-bE^{(n)}}$$

- Order n :
discontinuity in

Ehrenfest's definition



- Ex: first order:
discontinuous EOS:

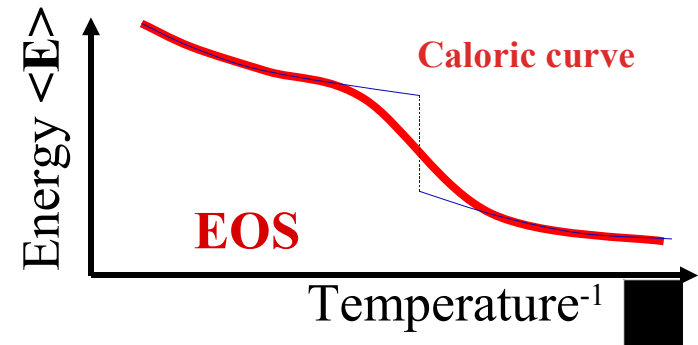
R. Balian, Springer (1982)

Phase transition in finite systems

- Thermodynamical potentials: $e.g. F = -T \log Z$
 Z analytical $\varepsilon > 0$, $\mathcal{T}$ $\mathcal{Z} = \sum_{(n)} e^{-bE^{(n)}}$

- No phase transition
 continuous

Ehrenfest's definition



- Ex:
 Continuous EOS:

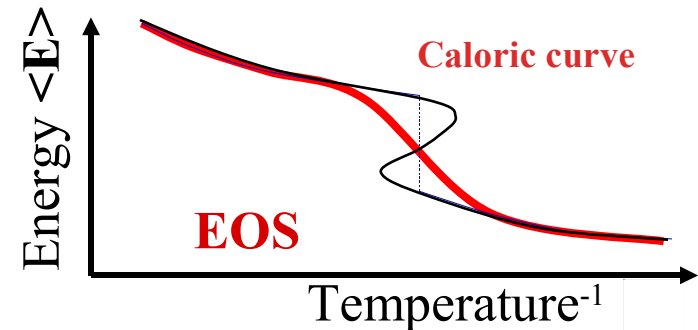


Phase transition in finite systems

- Thermodynamical potentials: $e.g. F = -T \log Z$
 Z analytical $\varepsilon > 0$, $\lim_{N \rightarrow \infty} \frac{1}{N} \log Z = \lim_{N \rightarrow \infty} \frac{1}{N} \log \sum_n e^{-bE^{(n)}}$

- No phase transition
continuous

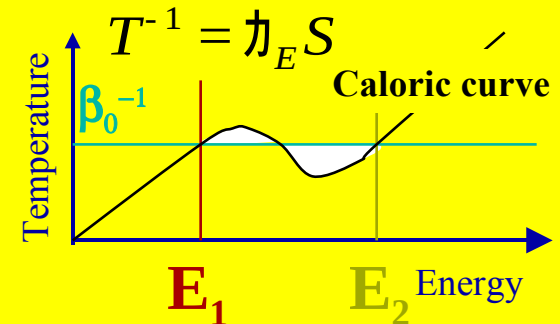
Ehrenfest's definition



- But anomaly in the entropy
Ex: negative heat capacity

$$C^{-1} = \kappa_E T$$

K. Binder, D.P. Landau Phys Rev B30 (1984) 1477;
 Lynden-Bell, D. & Wood, R. 1968, Mon. Not. R. Astr. Soc. 138, 495.;
 W. Thirring, Z. Phys. 235, 339 (1970),
 D.H.E. Gross, Rep. Prog. Phys. 53, 605 (1990),



1st order in finite systems

PC & Gulminelli Phys A (2003)

Free order param. (canonical)

- ◆ Zeroes of Z reach real axis

Yang & Lee Phys Rev 87(1952)404

- ◆ Bimodal distribution ($P_\beta(E)$)

K.C. Lee Phys Rev E 53 (1996) 6558

Fixed order para. (microcanonical)

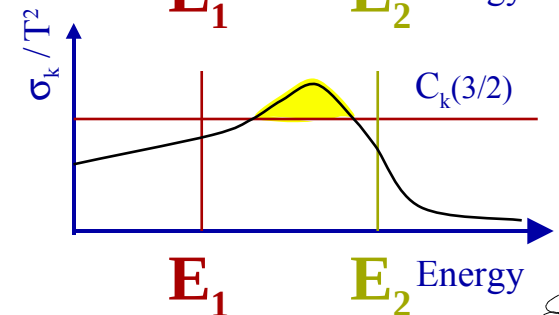
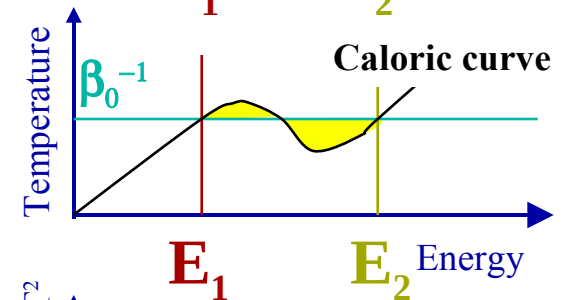
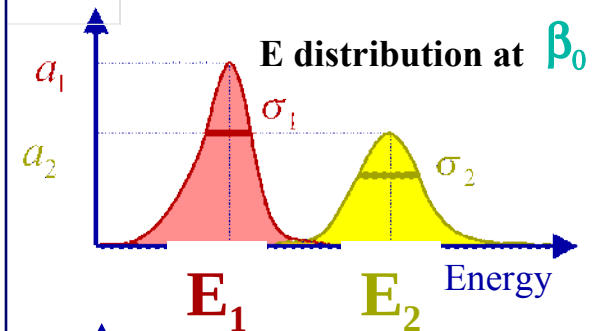
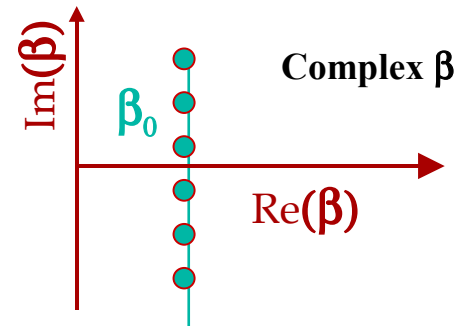
- ◆ Back Bending in EOS ($T(E)$)

K. Binder, D.P. Landau Phys Rev B30 (1984) 1477

Lynden-Bell, D. & Wood, R. 1968, Mon. Not. R. Astr. Soc. 138, 495.

- ◆ Abnormal fluctuation ($\sigma_k(E)$)

J.L. Lebowitz (1967), PC & Gulminelli, NPA 647(1999)153



- II -

Origin of singularities

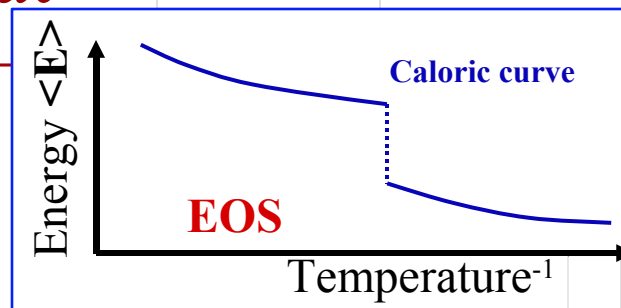
Zeroes of $Z(\beta)$

Bimodal $P(E)$

Thermodynamical potentials: *e.g.* $F = -T \log Z$
 non analytical at

L.E. Reichl, Texas Press (1980)

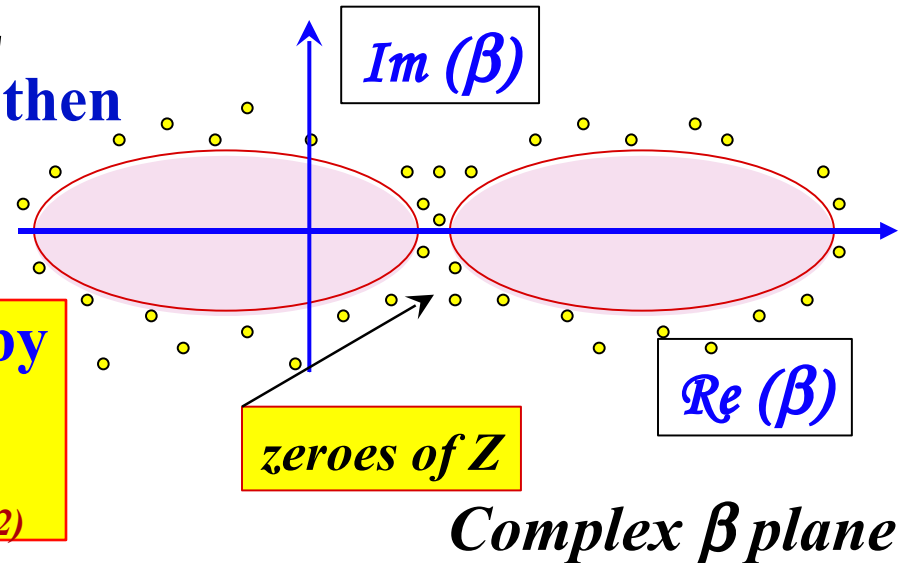
$$Z = \sum_{(n)} e^{-bE^{(n)}}$$



Discontinuities from zeroes of Z

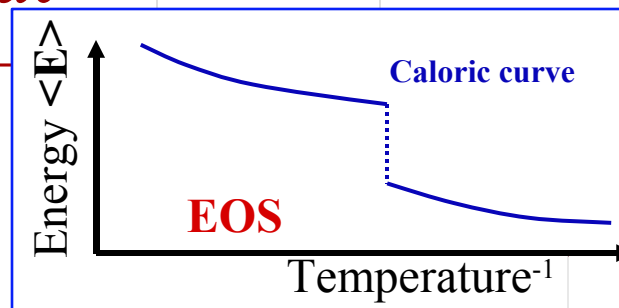
- If, when , no zeroes of Z converge on the *Real* β axis, then $\log Z$ remains analytical
 $\Rightarrow$ No phase transition

- Phase transitions are defined by the asymptotic distribution of zeroes of the partition sum Z
(C.N. Yang T.D. Lee 1952)



Thermodynamical potentials: $e.g. F = -T \log Z$
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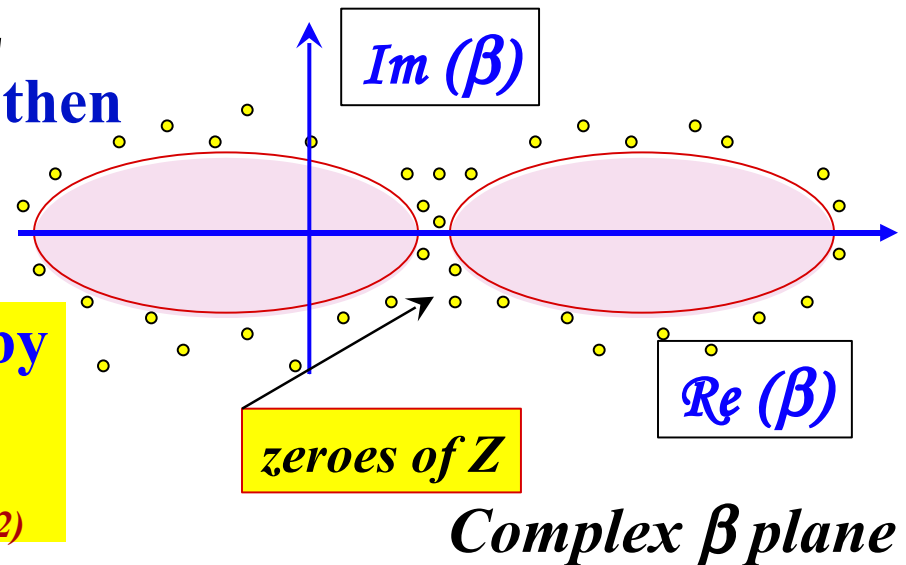


$$Z = \sum_{(n)} e^{-bE^{(n)}}$$

Discontinuities from zeroes of Z

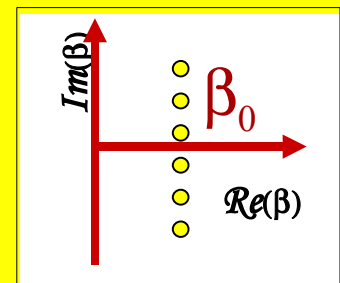
- If, when $\beta \rightarrow \infty$, no zeroes of Z converge on the *Real* β axis, then $\log Z$ remains analytical
 $\Rightarrow$ **No phase transition**

- Phase transitions are defined by the asymptotic distribution of zeroes of the partition sum Z
(C.N. Yang T.D. Lee 1952)



- **First order phase transitions**
A uniform density of zero's on a line crossing the real β axis perpendicularly at β_0

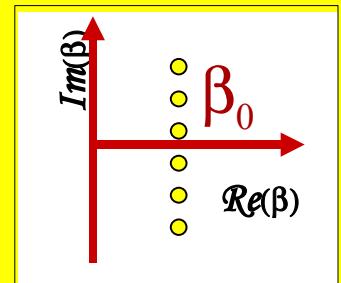
Yang - Lee theorem



■ First order phase transitions

A uniform density of zero's on a line crossing the real β axis perpendicularly at β_0

Yang - Lee theorem



■ A theoretical definition

◆ Only valid asymptotically: $V = \infty$

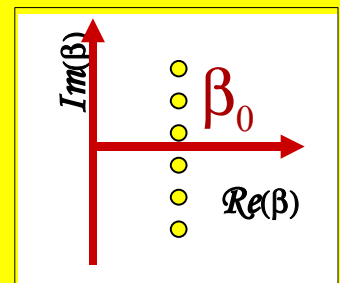
◆ How to extend it, $V \neq \infty$?

■ How to use it experimentally ?

■ First order phase transitions

A uniform density of zero's on a line crossing the real β axis perpendicularly at β_0

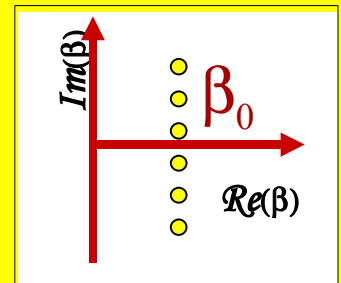
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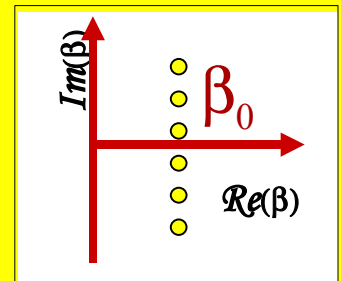
Link with energy distribution $P(E)$

Z Laplace transform of P :

Ph. Ch., F. Gulminelli, Physica A 2003

First order phase transitions

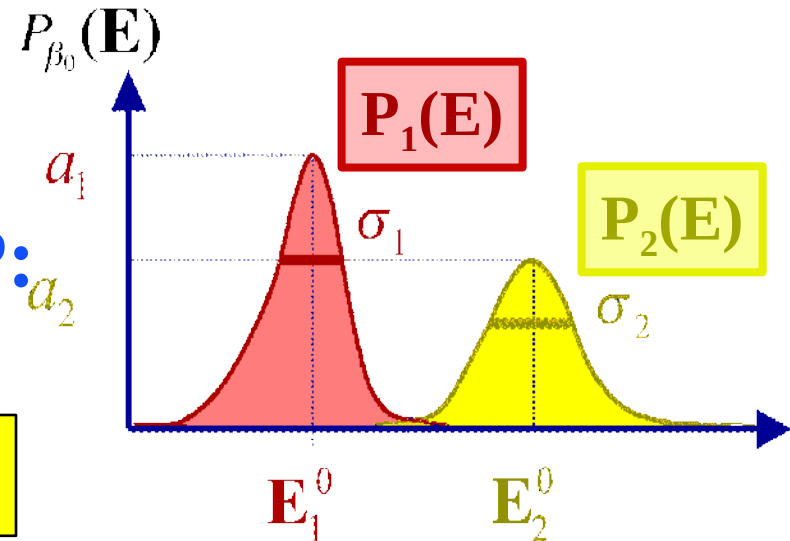
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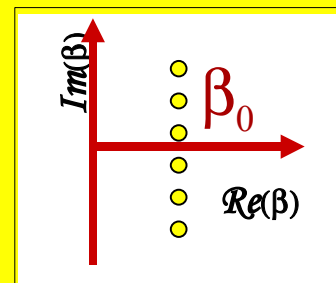
Bimodal P with $\Delta E \propto N$



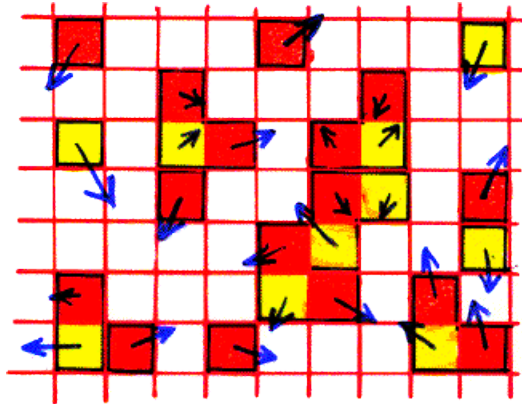
Ph. Ch., F. Gulminelli, Physica A 2003

First order phase transitions

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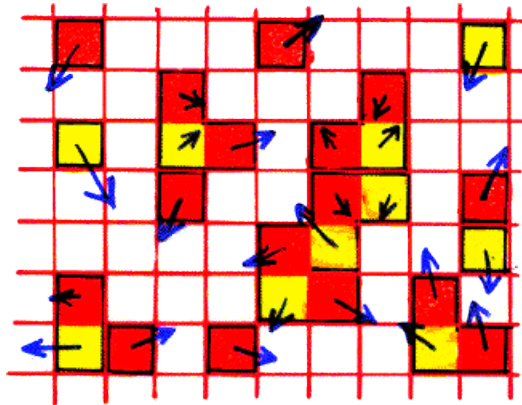


Lattice gas model



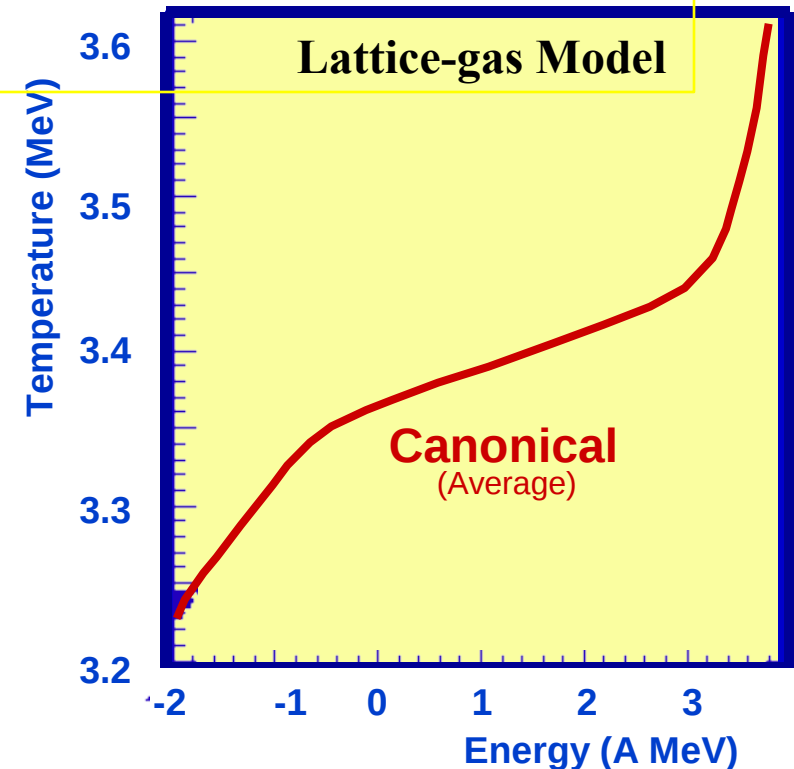
- Closest neighbors interaction
- Metropolis Boltzmann weight $e^{-\beta E}$
- Average volume $\langle V \rangle$ (pressure)

Lattice gas model

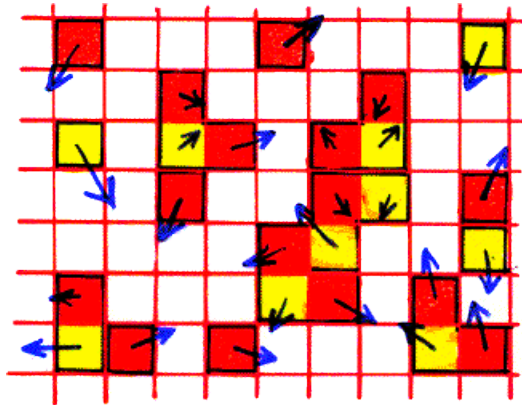


■ $\langle E \rangle$ smooth function of β
(canonical caloric curve)

- Closest neighbors interaction
- Metropolis Boltzmann weight $e^{-\beta E}$
- Average volume $\langle V \rangle$ (pressure)

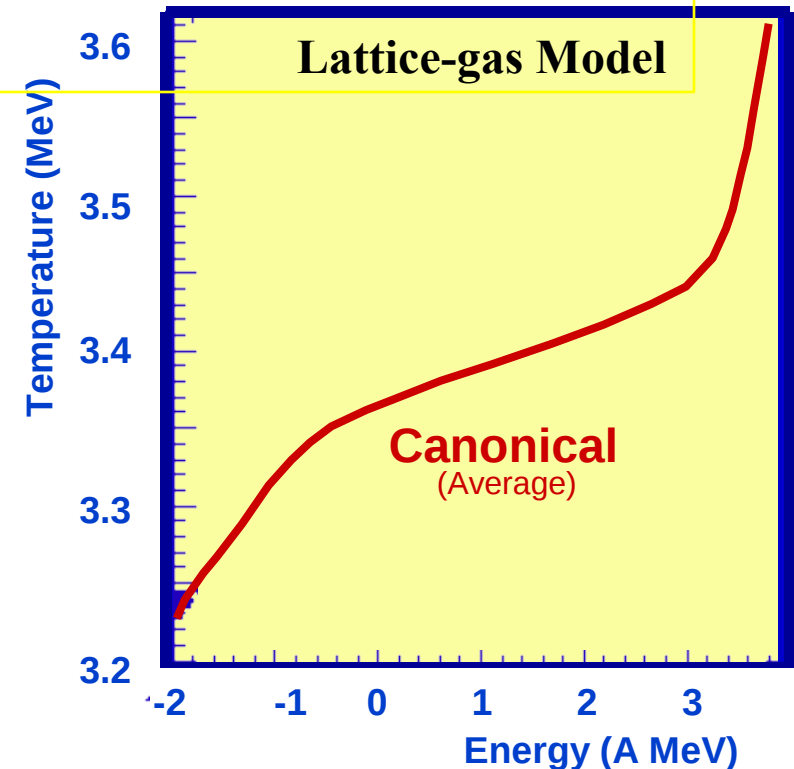


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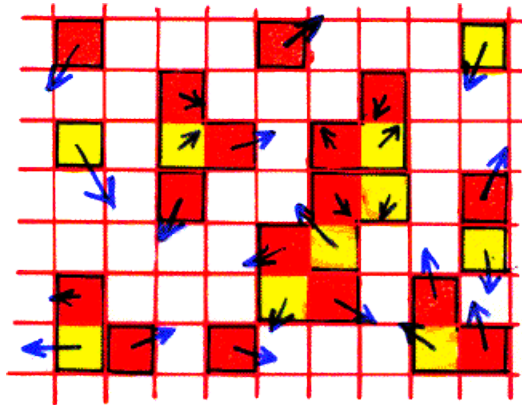


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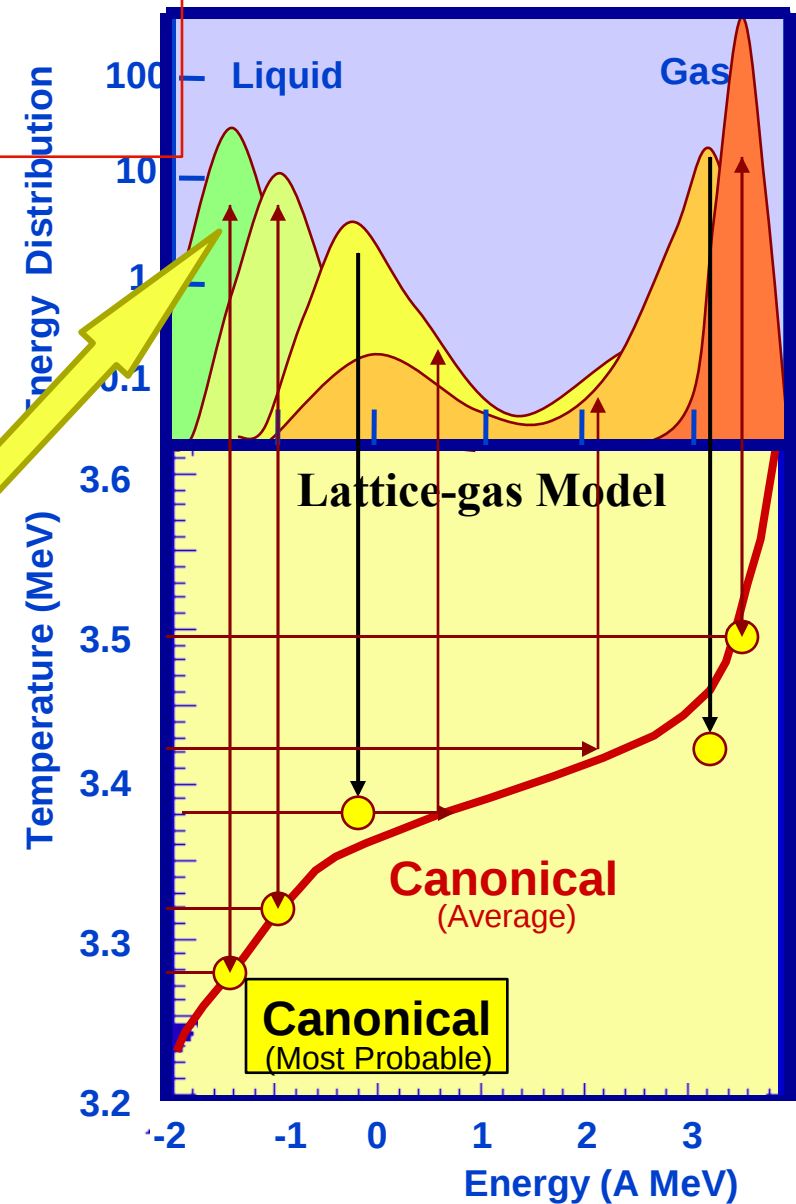
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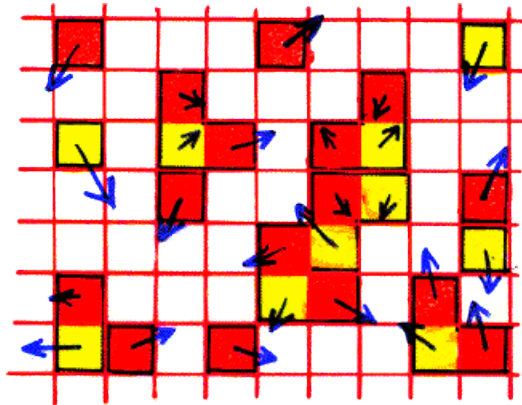
Lattice gas model



■ Energy distribution at a fix temperature

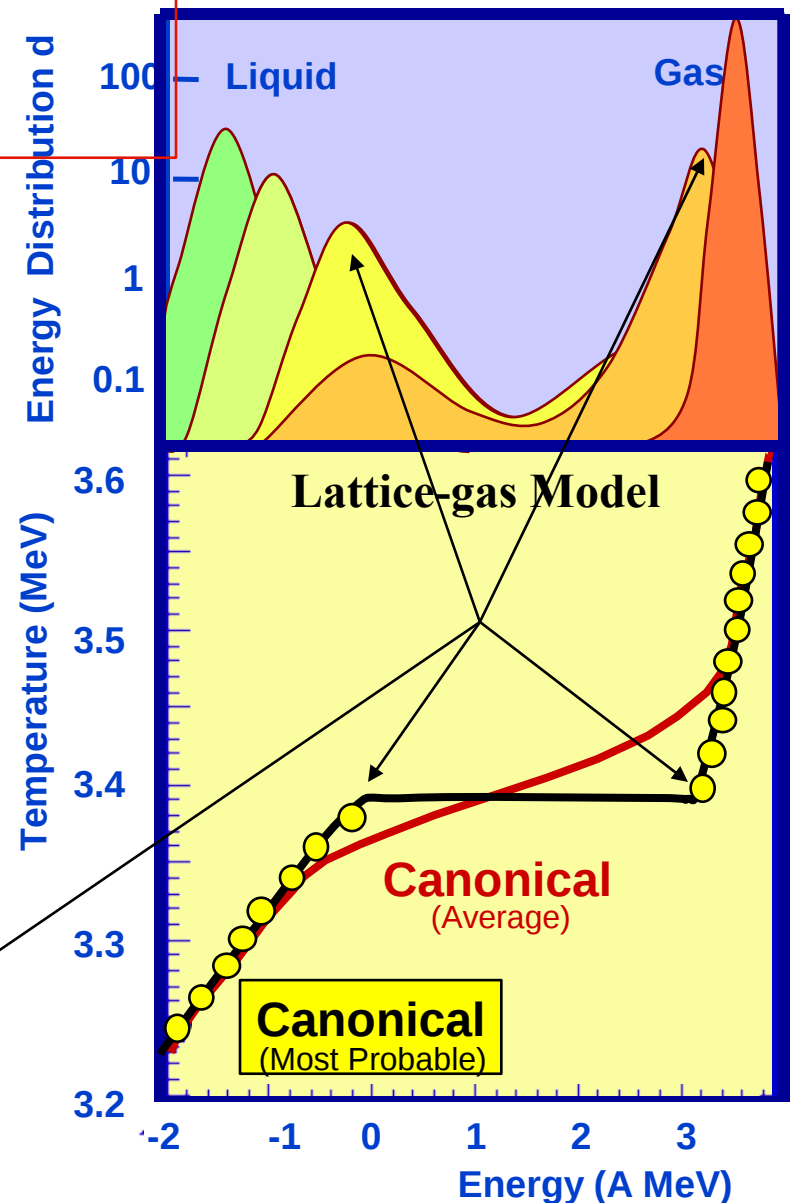


Lattice-gas model

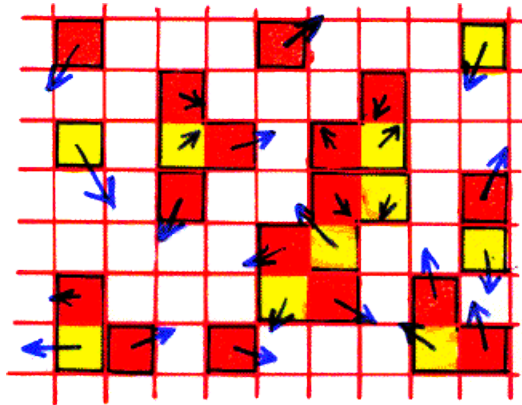


■ Energy distribution at a fixed temperature

■ Discontinuity of the most probable



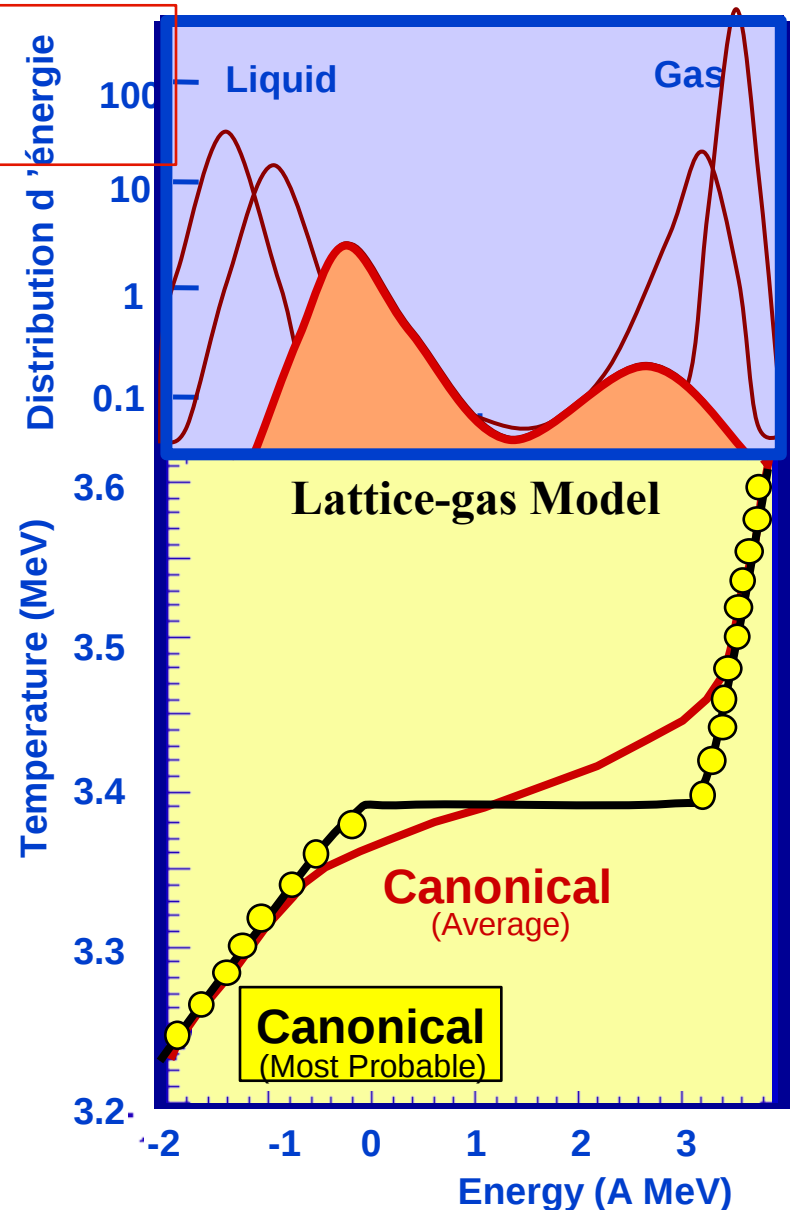
Lattice-gas model



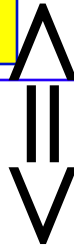
■ Energy distribution at a fixed temperature

■ Bimodal distribution

■ Discontinuity of the most probable



A first order phase transition in finite systems



A bimodal (event) distribution of
energy

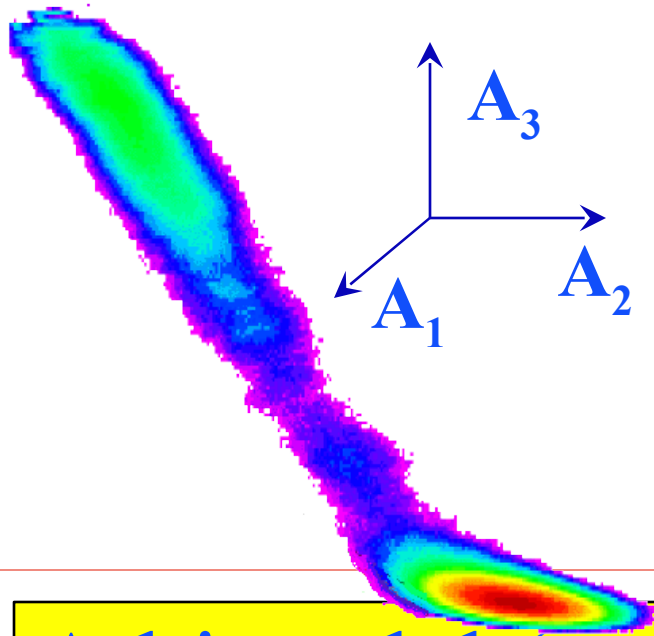
If the energy is free to fluctuate
(canonical - energy reservoir)

A first order phase transition in finite systems



A bimodal (event) distribution of an
observable (order parameter)

If this observable is free to
fluctuate (intensive ensemble)



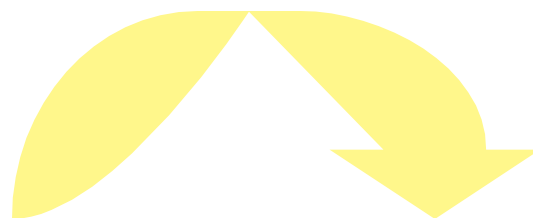
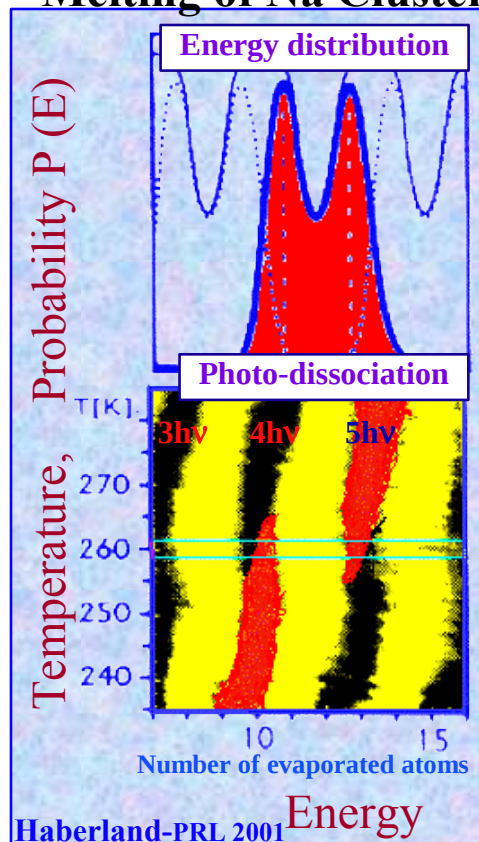
- **Access to order parameters**
 - ◆ **Nature of order parameter:**
 - ✦ Is it collective?
 - ◆ **Scaling of the jump with N:**
 - ✦ What happen at the thermo limit ($N = \infty$) ?
 - ✦ Is it a macro. phase transition?

A bimodal (event) distribution of an observable (order parameter)

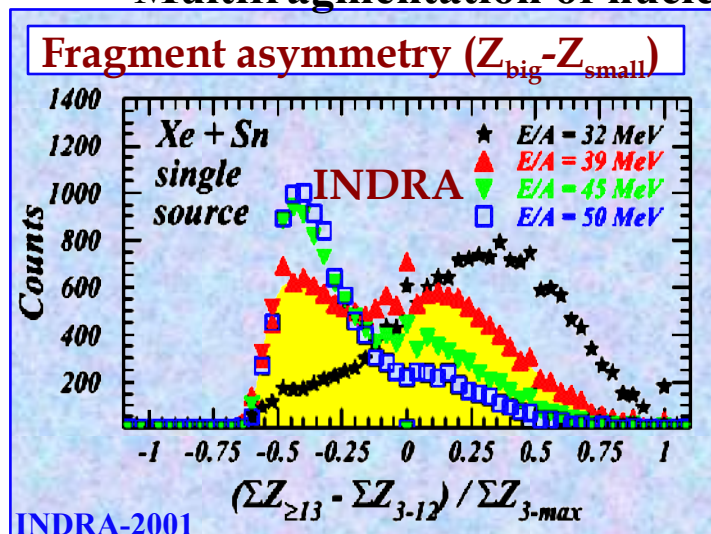
If this observable is free to fluctuate (intensive ensemble)

Bimodal distributions

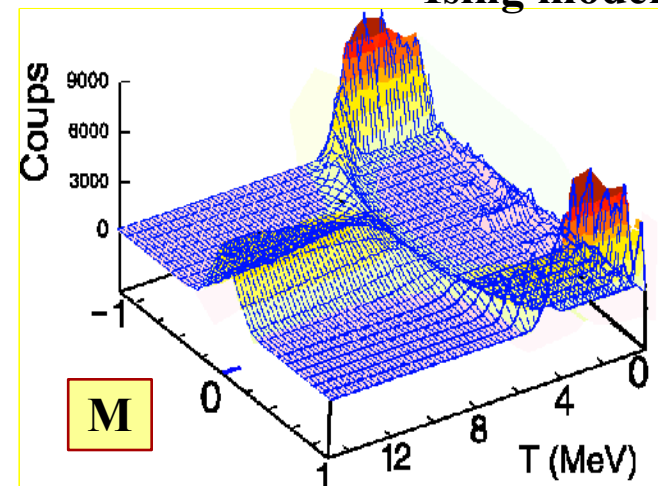
Melting of Na Cluster



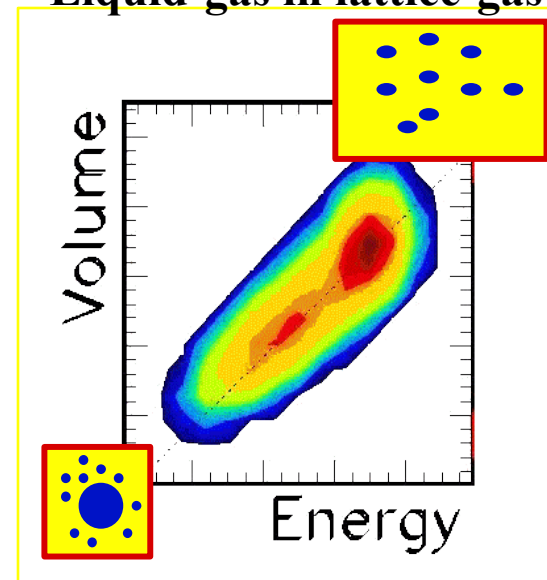
Multifragmentation of nuclei



Ising model

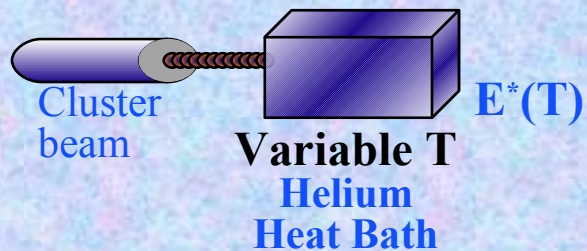
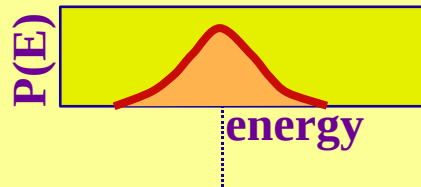


Liquid-gas in lattice-gas



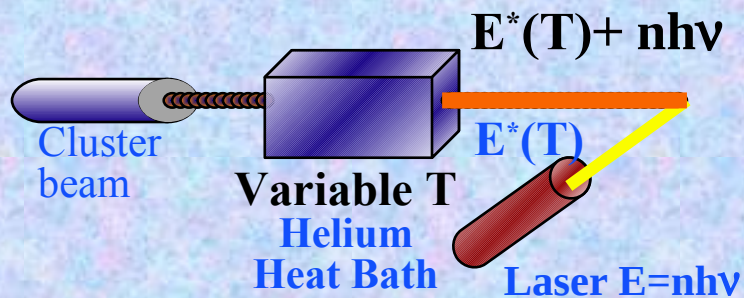
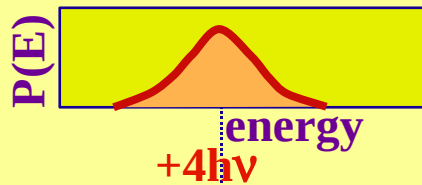
Melting of Na_{147}^{+}

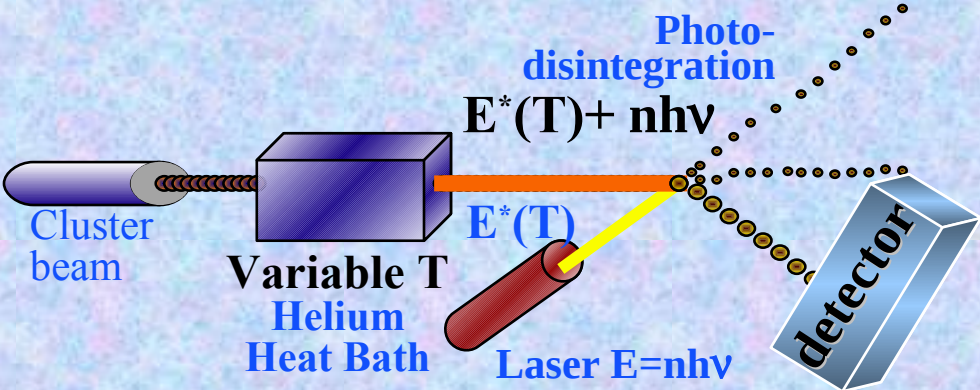
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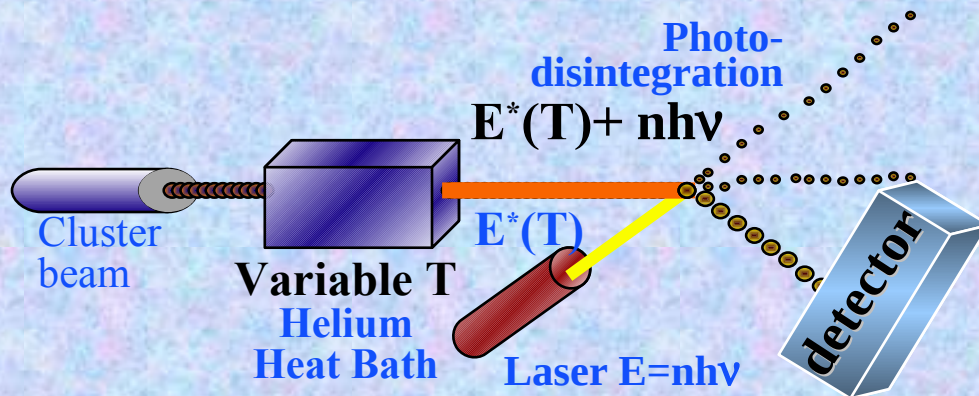
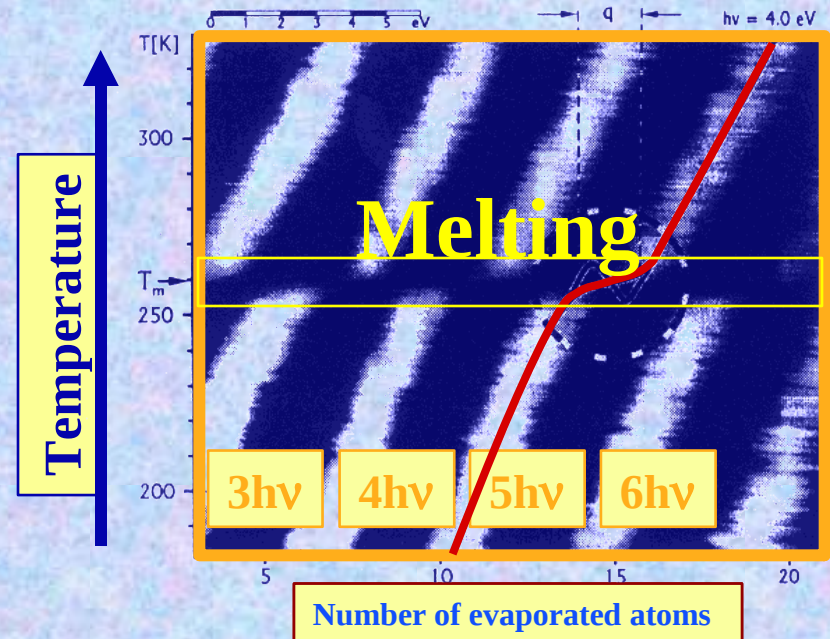
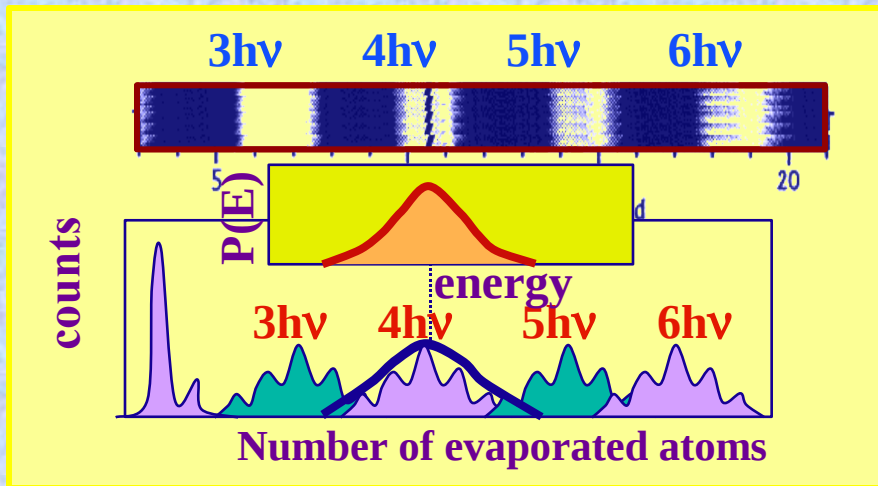
Melting of Na_{147}^{+}

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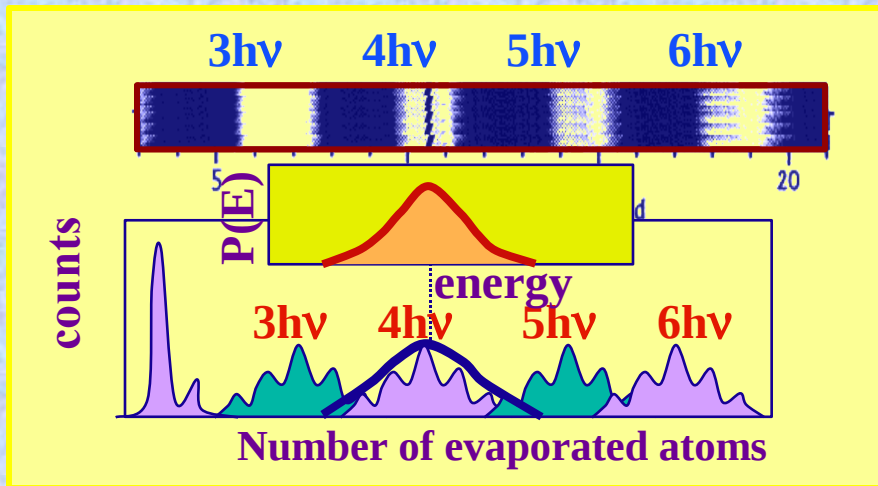




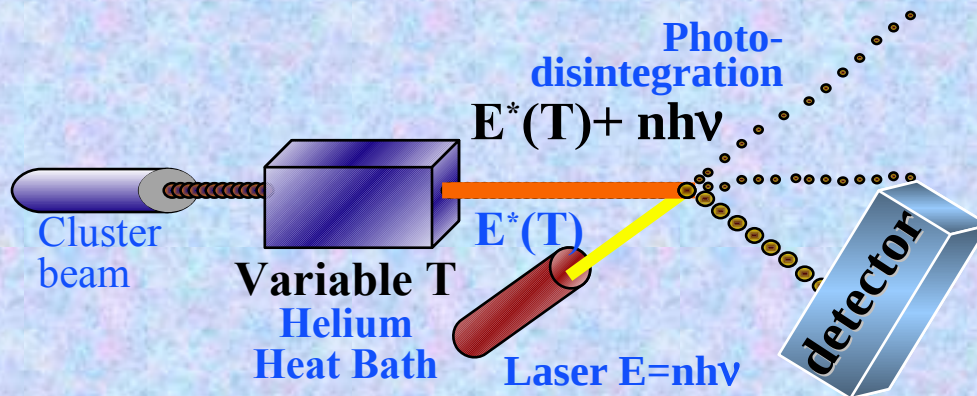
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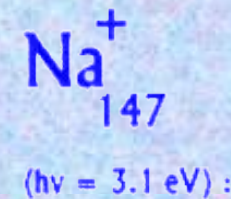
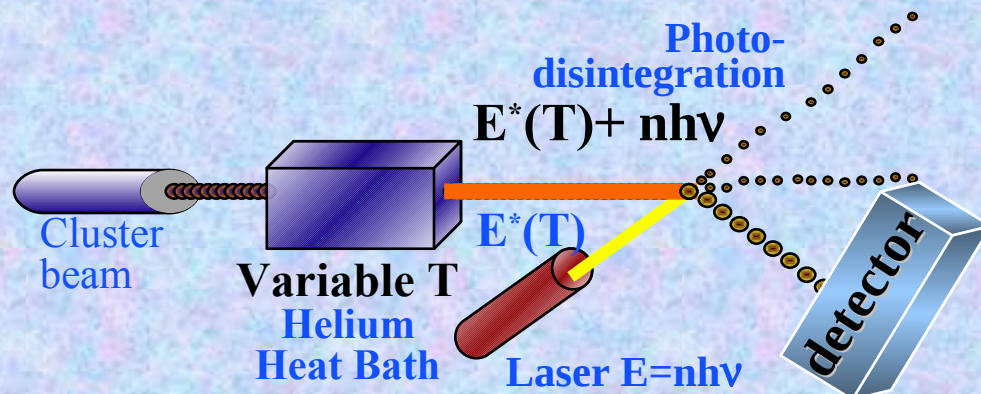
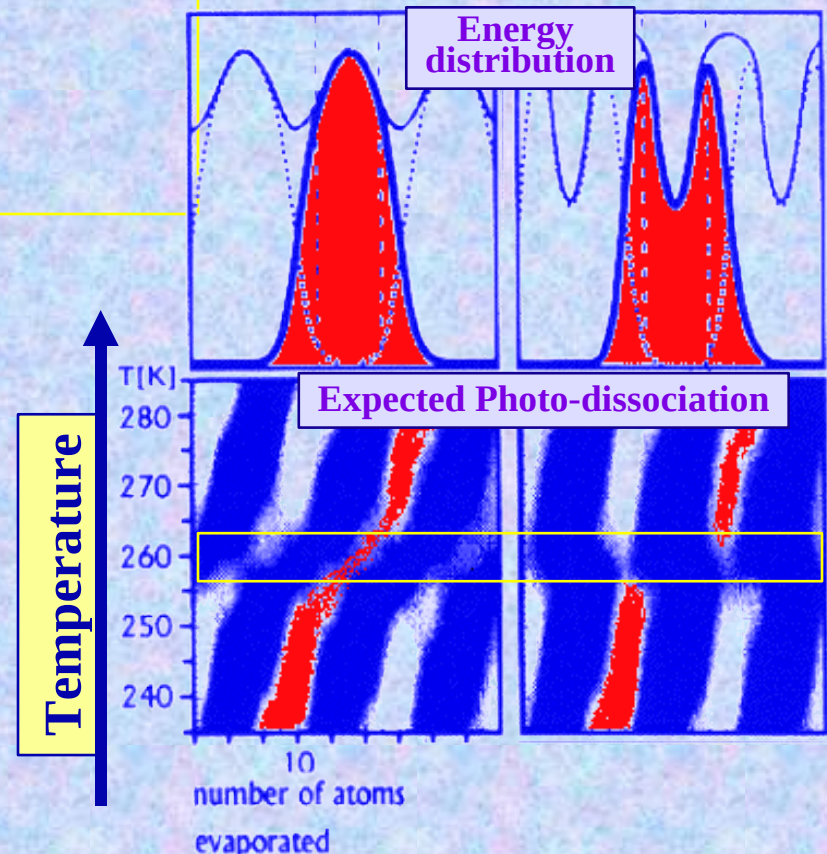
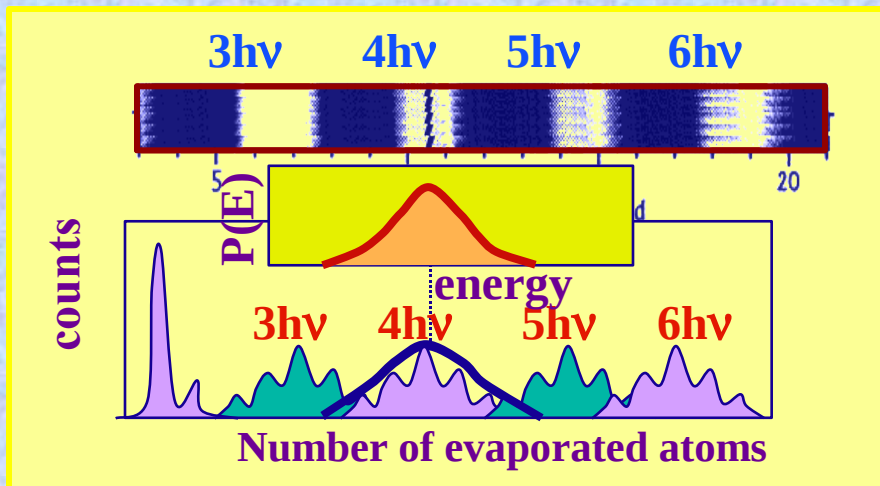


Temperature ↑



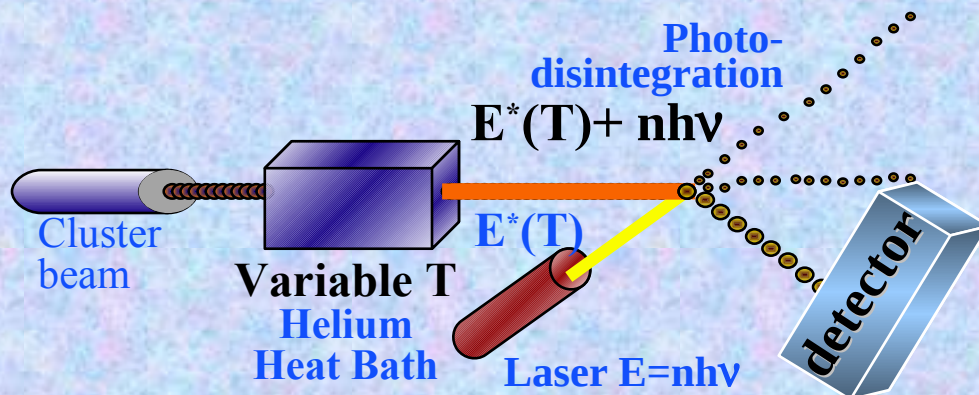
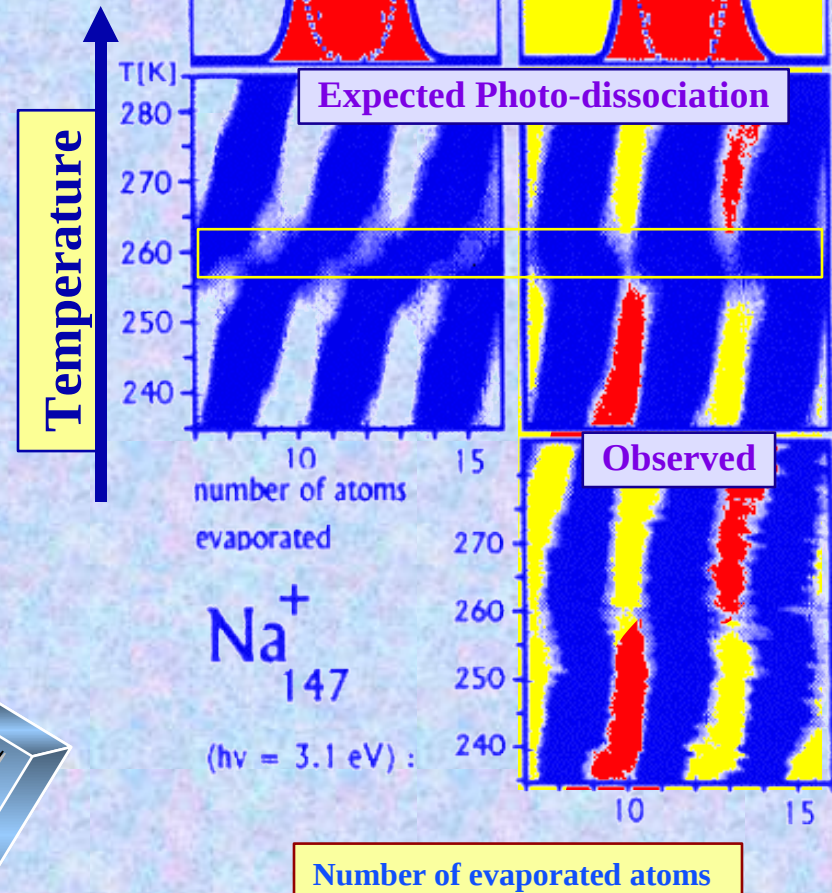
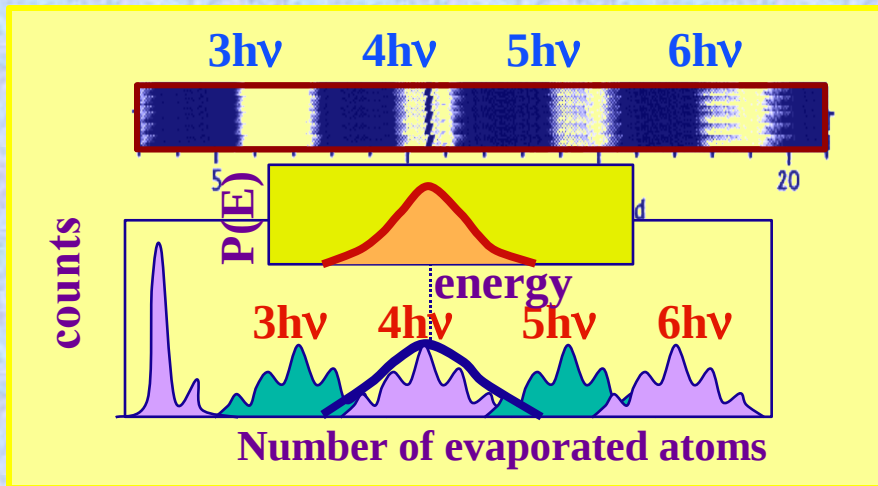
Number of evaporated atoms

Melting of Na_{147}^{+}



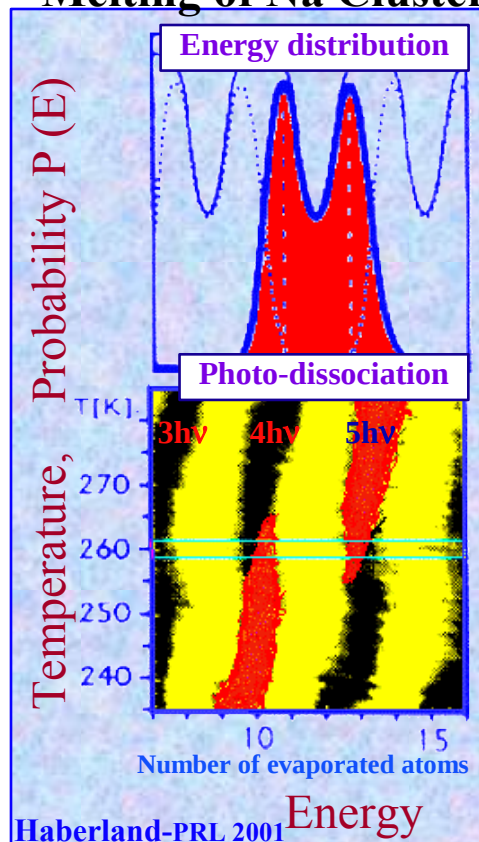
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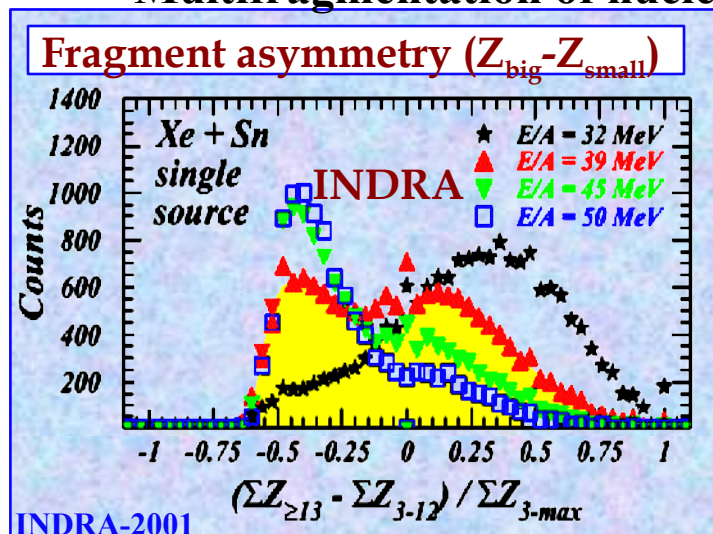


Bimodal distributions

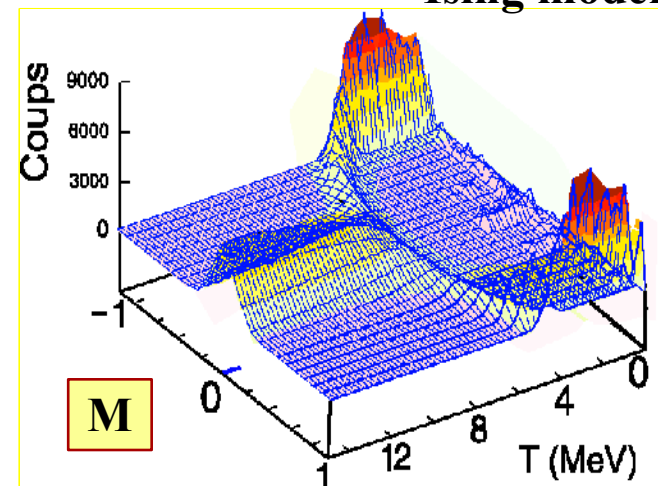
Melting of Na Cluster



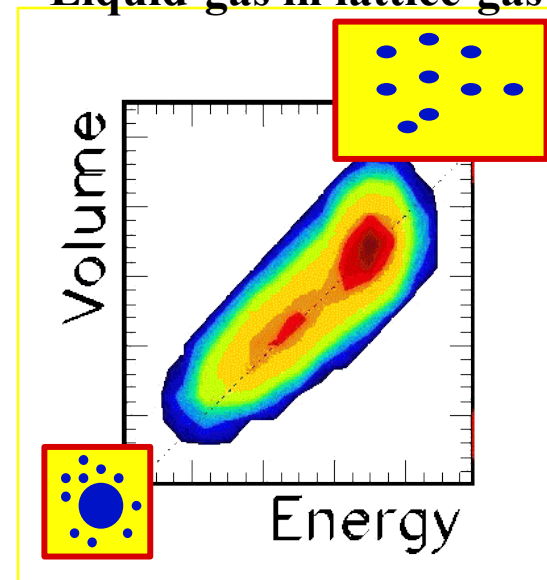
Multifragmentation of nuclei



Ising model



Liquid-gas in lattice-gas



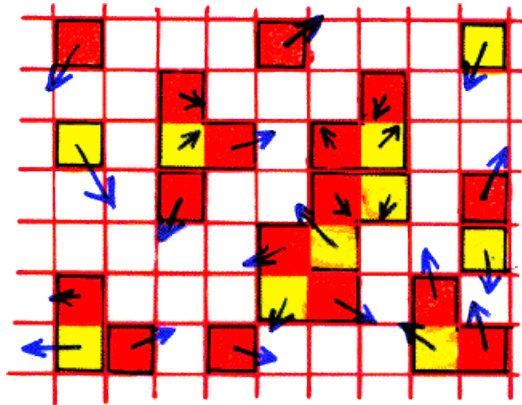
- III -

Convex S

Back bending T

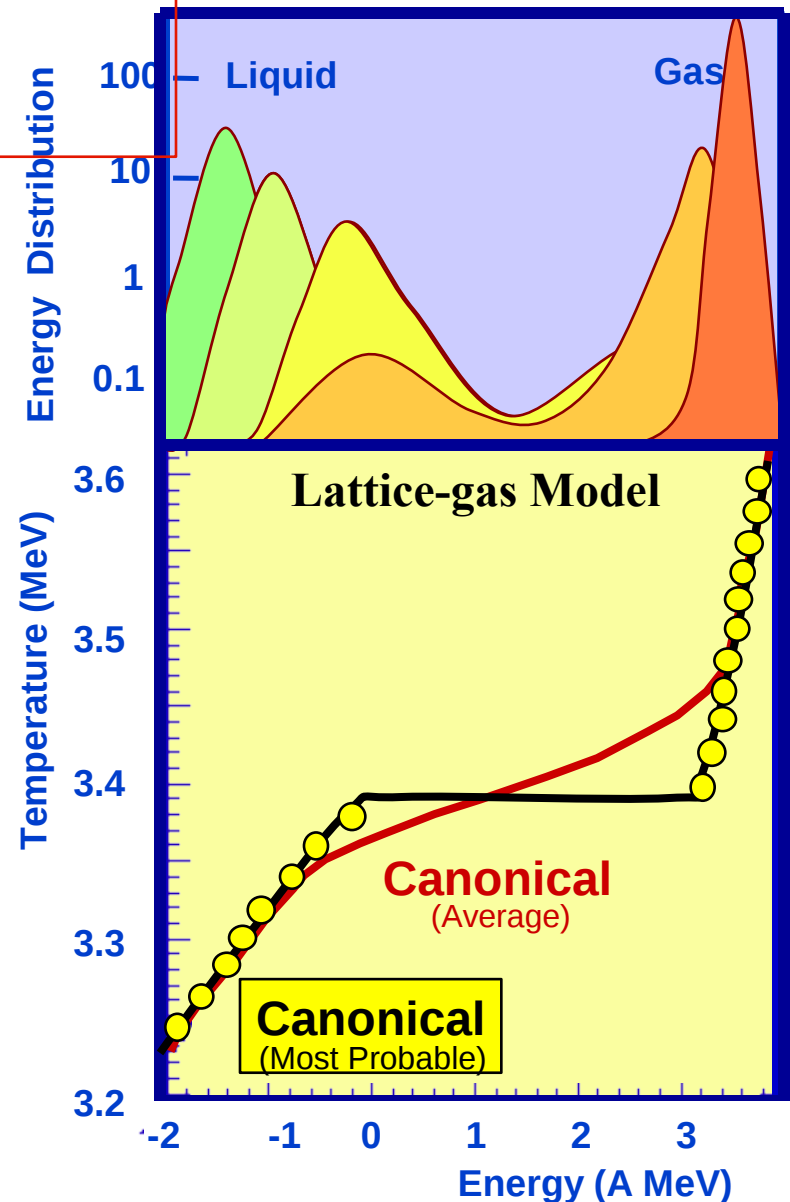
Negative heat capacity

Lattice-gas model



■ Energy distribution at a fixed temperature

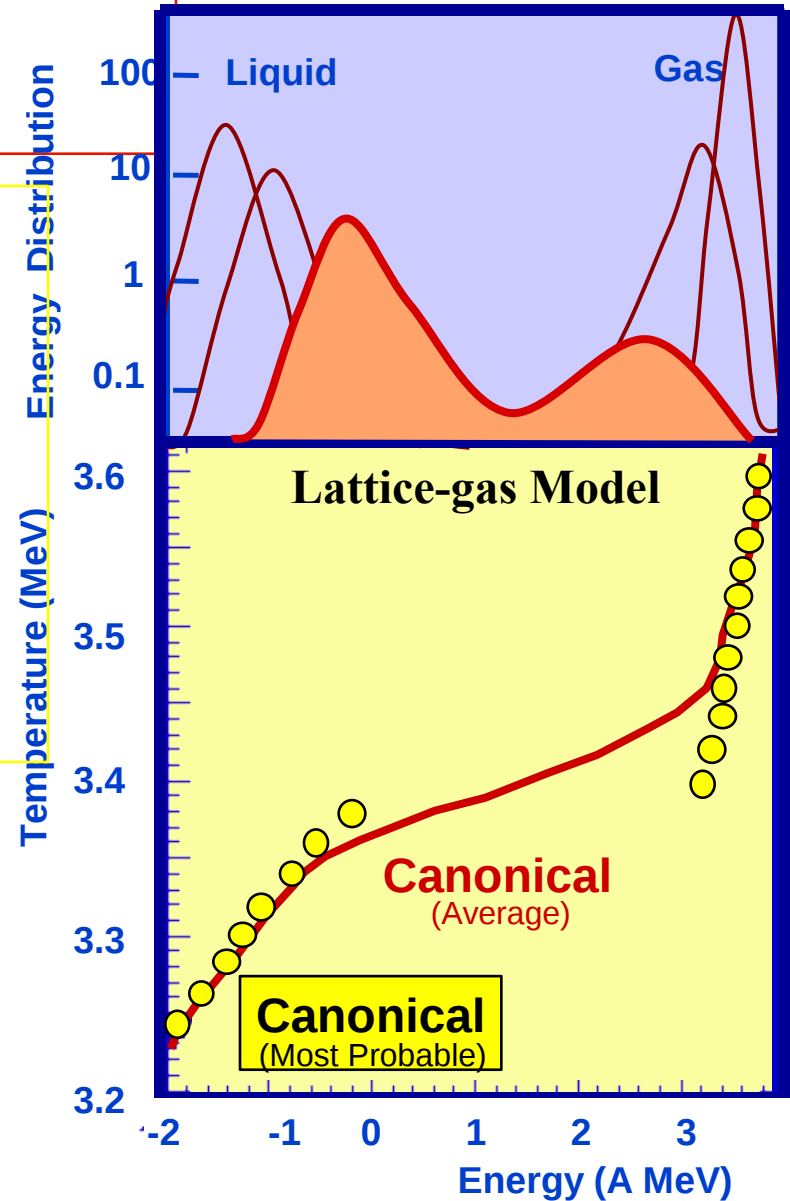
■ Discontinuity of the most probable



Lattice-gas model

■ E distribution

$$\diamond P_{\beta}(E) = W(E)e^{-\beta E} / Z_{\beta}$$



Lattice-gas model

■ Microcanonic: $S = \log W$

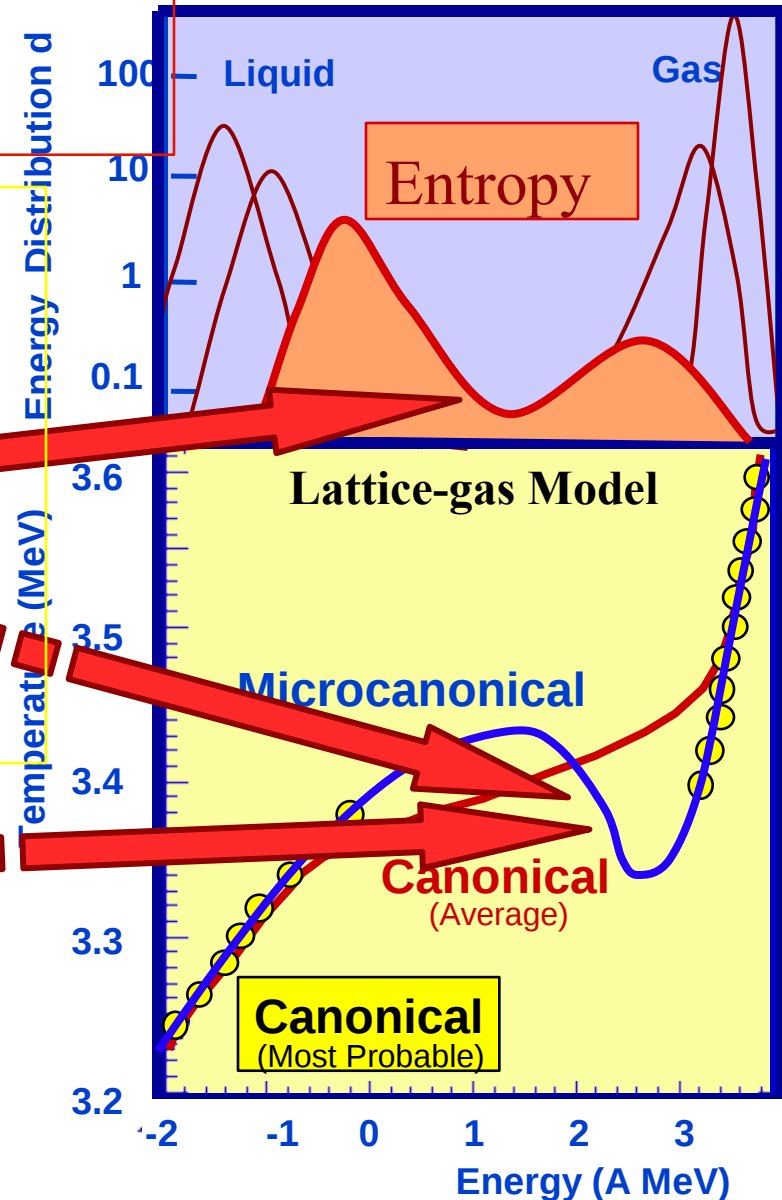
♦ $\log Z_\beta P_\beta(E) = S(E) - \beta E$

■ Convex Entropy

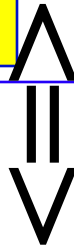
■ Decreasing T: $T^{-1} = d_E S$

♦ $T^{-1} = \beta - d_E (\log P_\beta)$

■ Negative Heat Capacity



A first order phase transition in finite systems

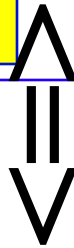


A negative heat capacity

If the energy is fixed
(microcanonical)



A first order phase transition in finite systems

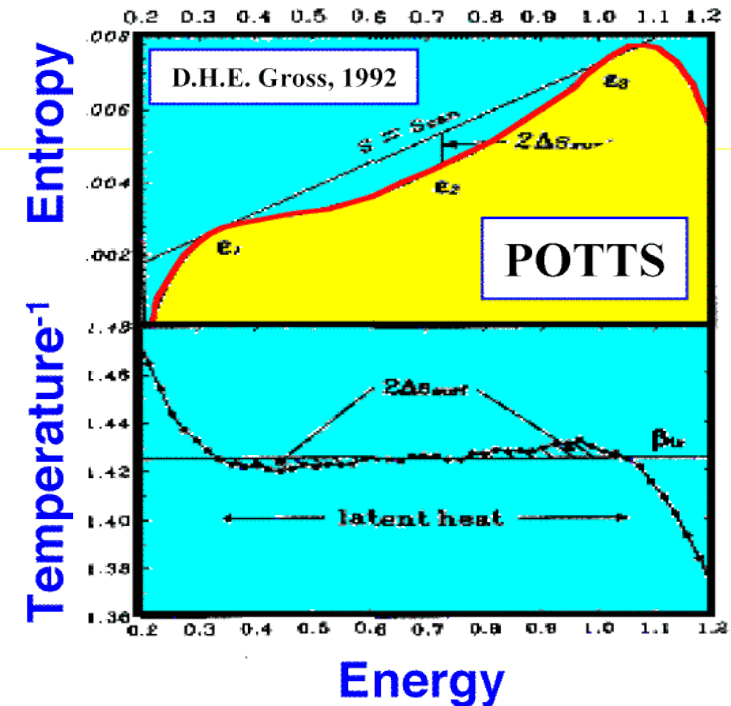
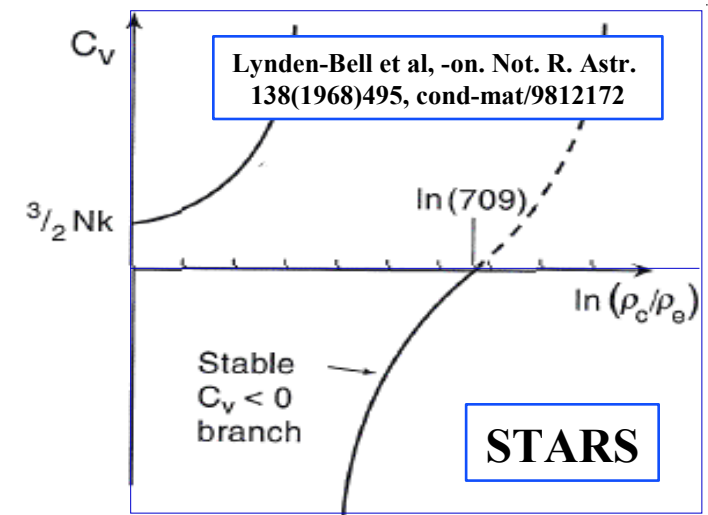
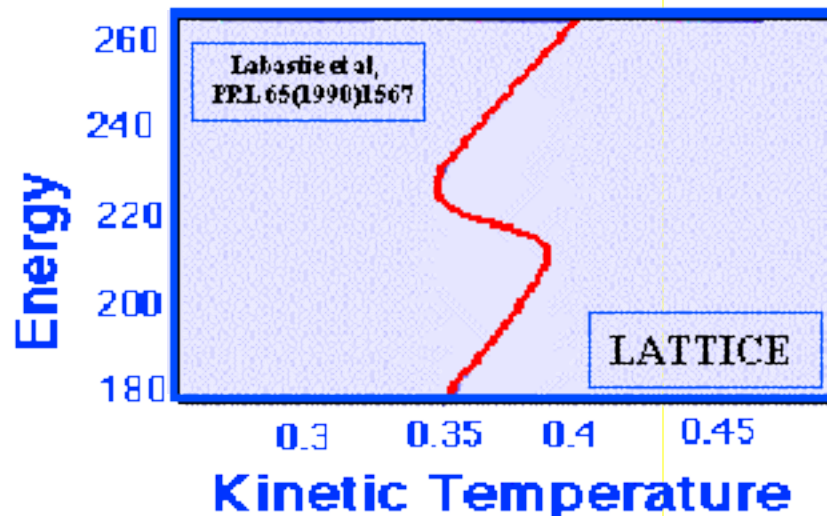


A inverted curvature of the
thermodynamical potential

If the observables (order param.)
are fixed (intensive ensemble)

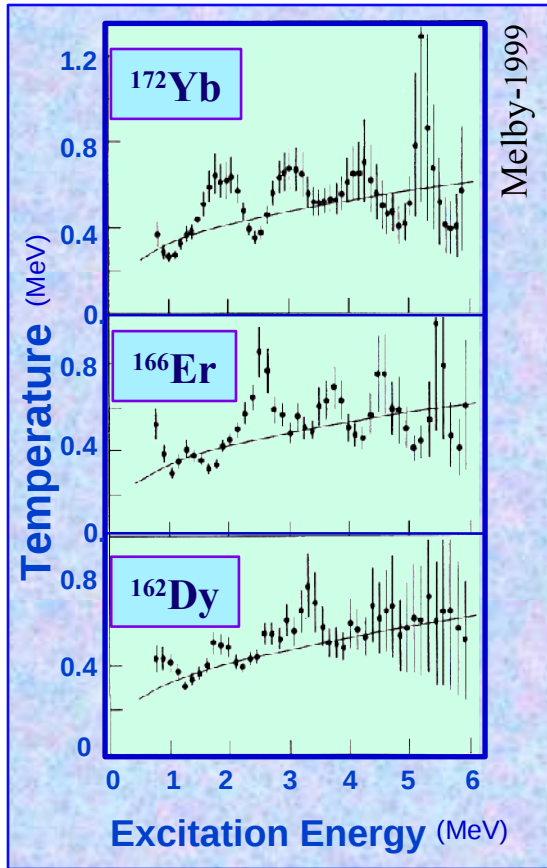
Negative Heat Capacity

First Order Phase Transition in Finite Systems



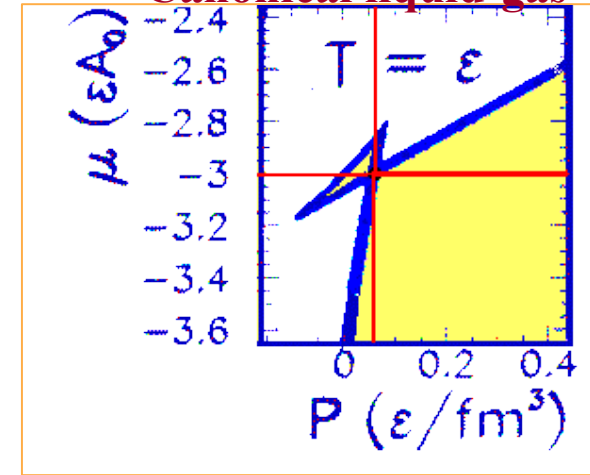
Abnormal curvatures

Nuclear superfluidity



Convex entropy

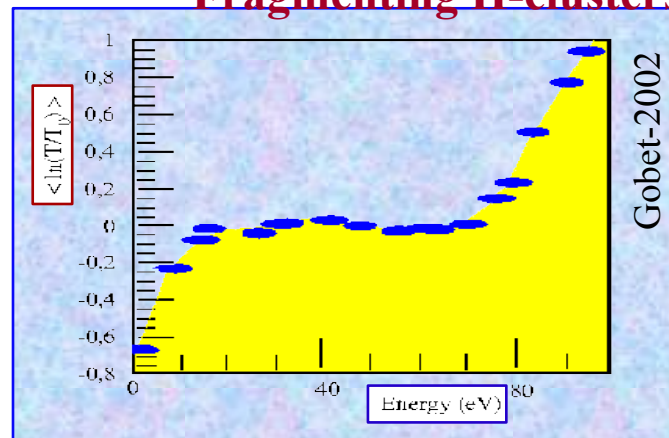
Canonical liquid-gas



F. Gulminelli, Ph. Ch. PRL(1998)

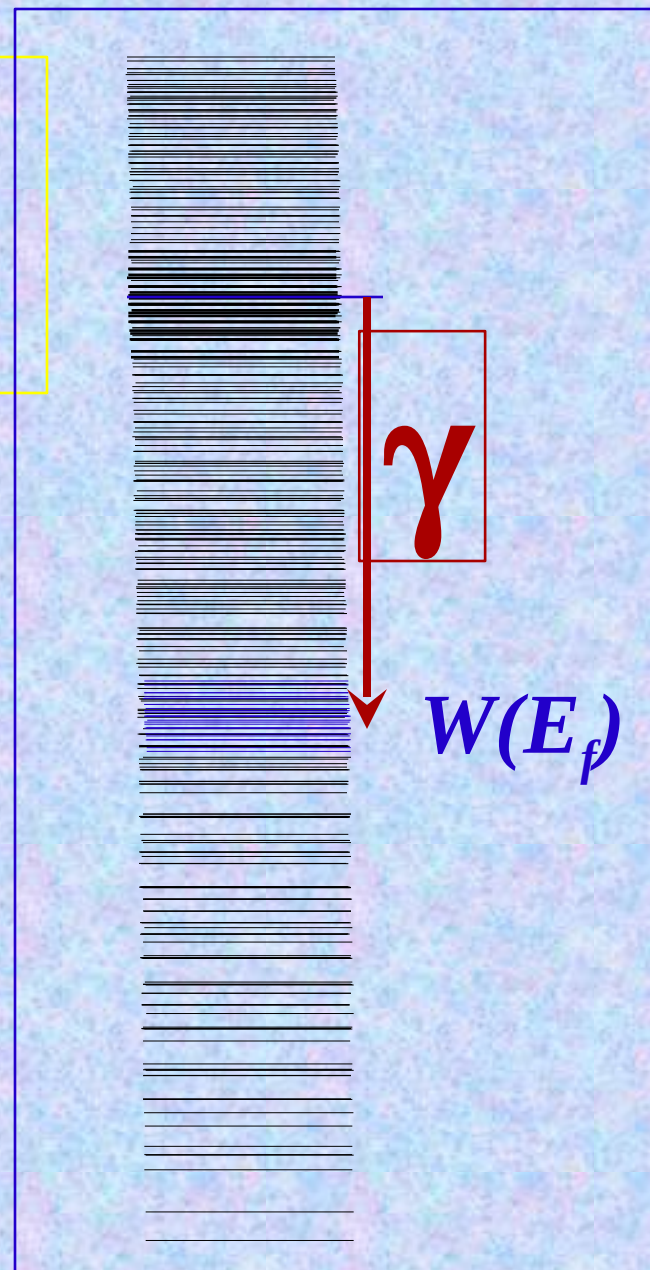
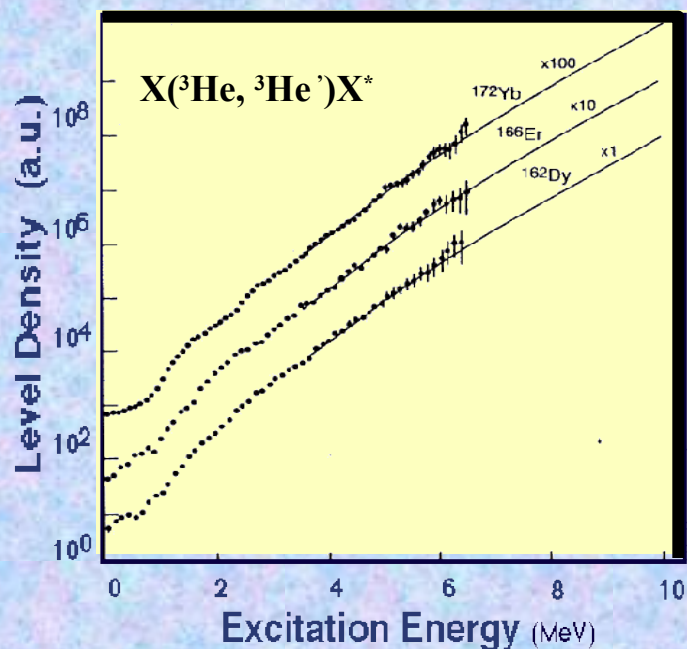
Compressibility < 0
Susceptibility < 0

Fragmenting H-clusters



T decreasing with E

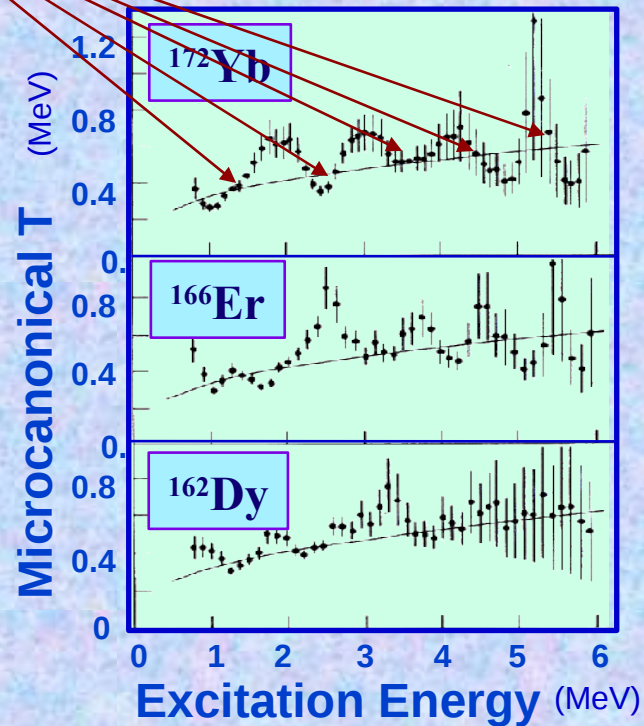
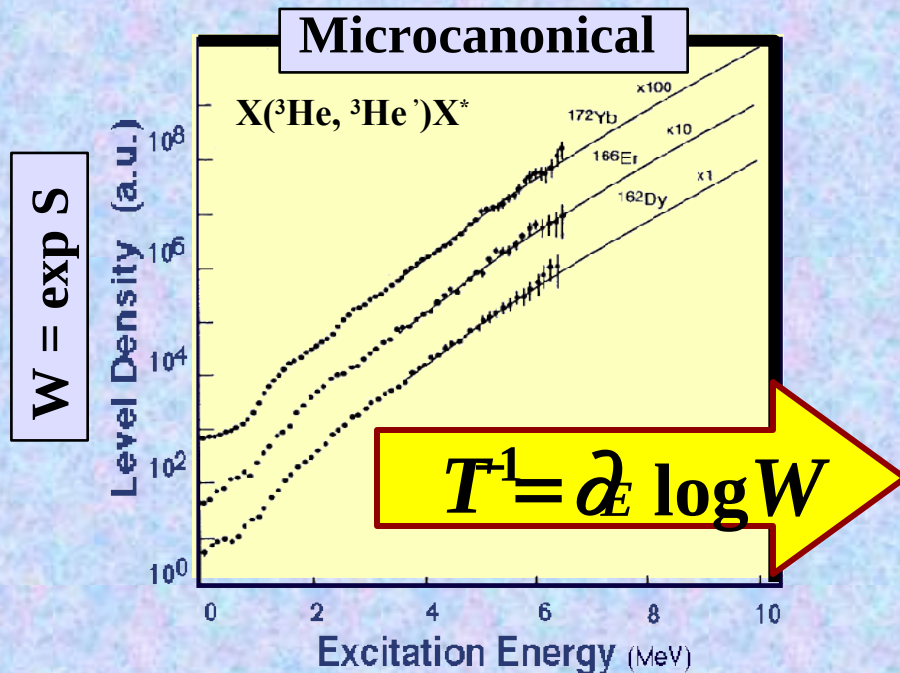
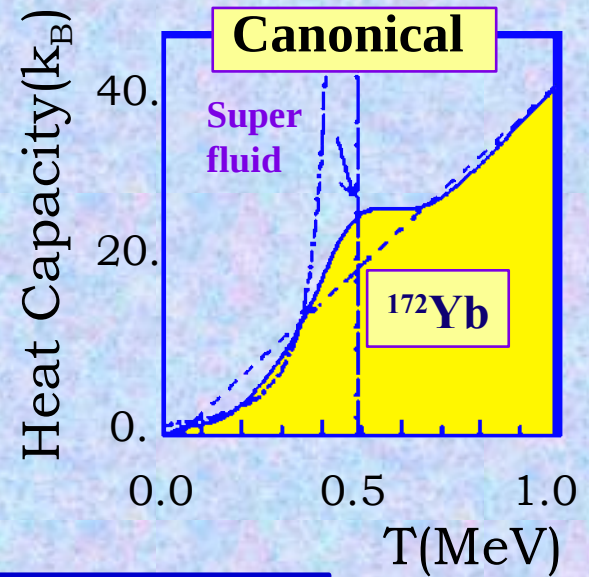
At low energy γ Decay: level density



Superfluid Nuclei

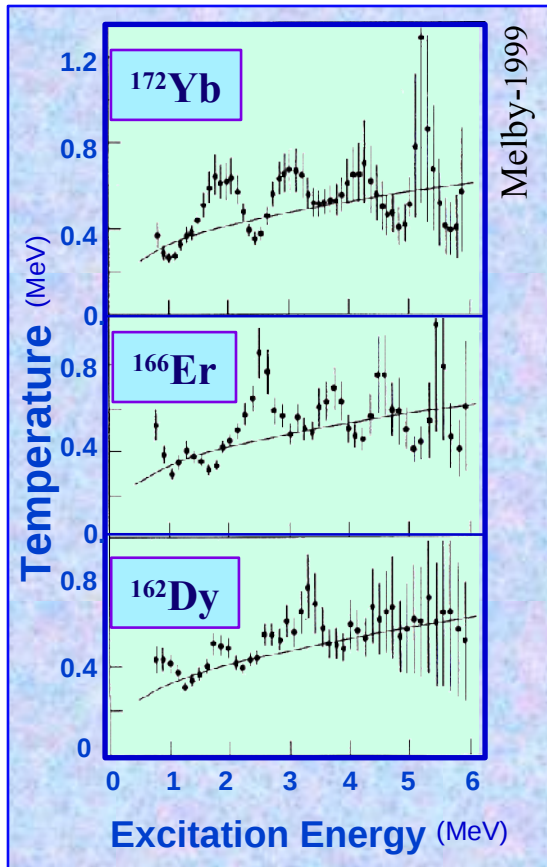
γ Decay: level density

■ Breaking of pairs



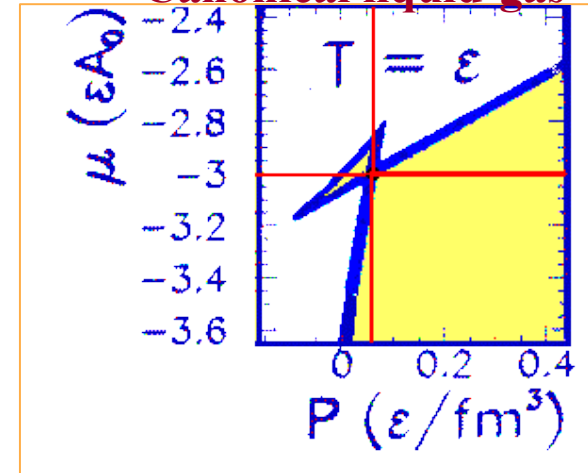
Abnormal curvatures

Nuclear superfluidity



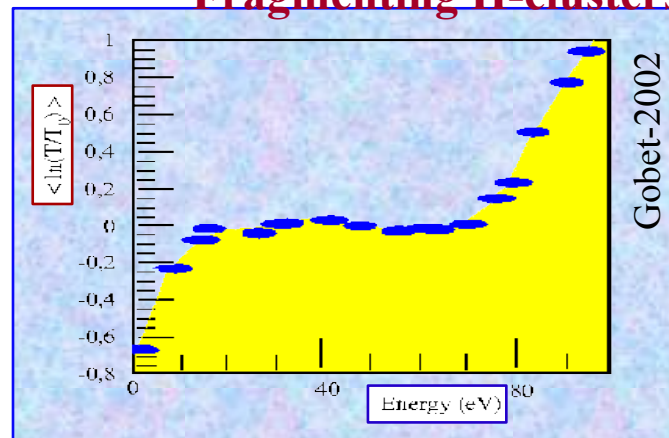
Convex entropy

Canonical liquid-gas



Compressibility < 0
Susceptibility < 0

Fragmenting H-clusters



T decreasing with E

- IV -

Negative heat capacities
Abnormal energy fluctuations
Multifragmentation



Heat capacity from energy fluctuation

■ Canonical: fluct. E_{tot}

$$Z_{\text{tot}} = Z_k Z_p$$

$$\sigma_E^2 = \sigma_k^2 + \sigma_p^2$$

$$\partial_{\beta}^2 \log Z_i = C_i / \beta^2 = \sigma_i^2$$

■ Microcanonical:
fluct. partial energy

$$W_{\text{tot}} = W_k \otimes W_p$$

$$\sigma_E^2 = 0, \quad \sigma_k^2 = \sigma_p^2$$

$$\partial_{\beta}^2 \log W_{\text{tot}} = -1/CT^2 = f(\sigma_k^2)$$

PC, Gulminelli NPA(1999)

Heat capacity from energy fluctuation

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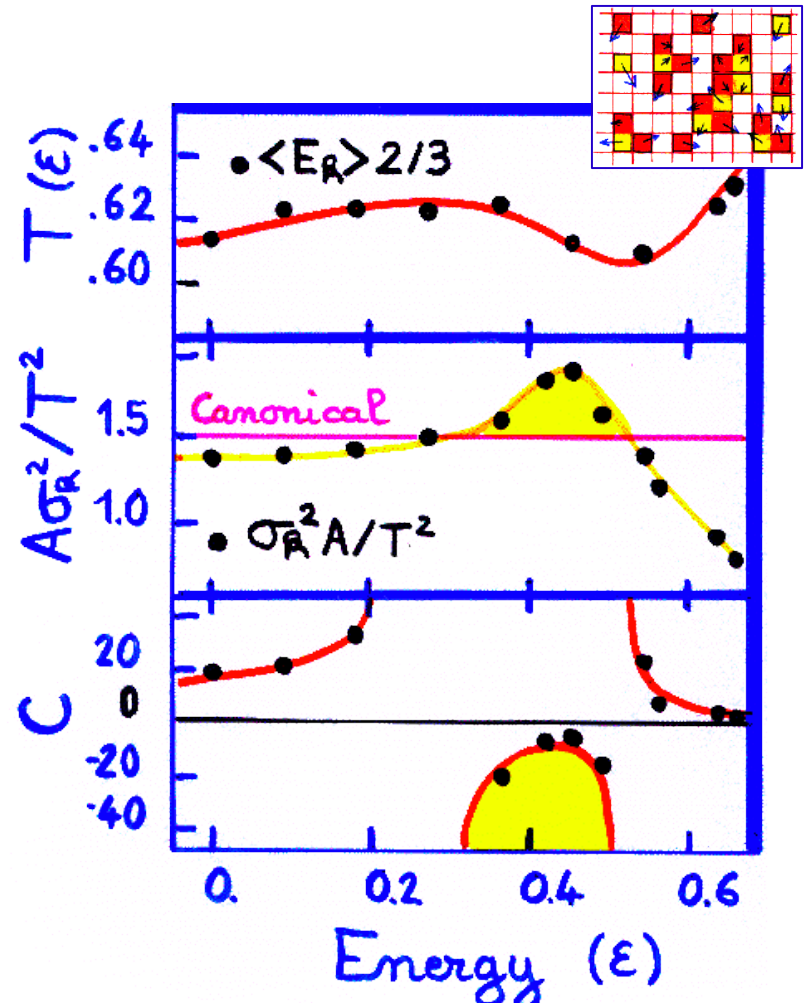
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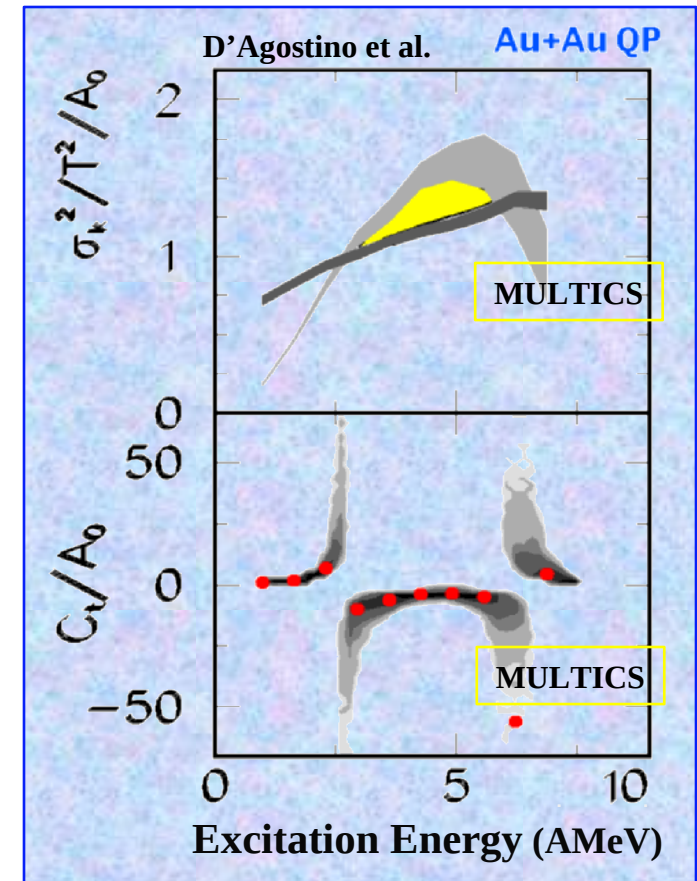
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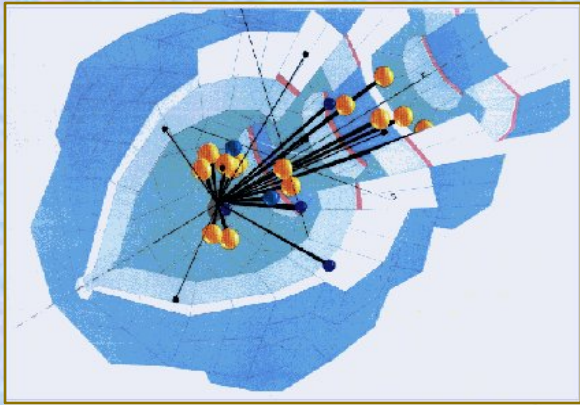
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PC, Gulminelli NPA(1999)

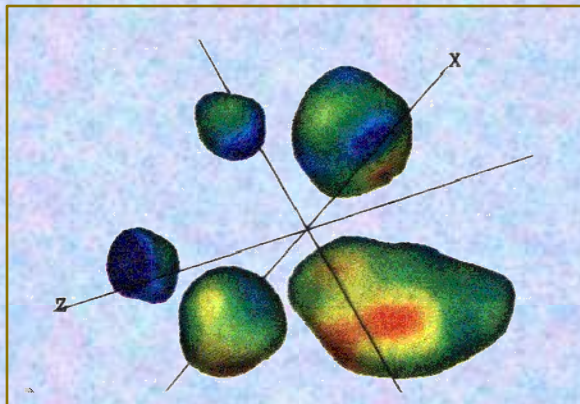
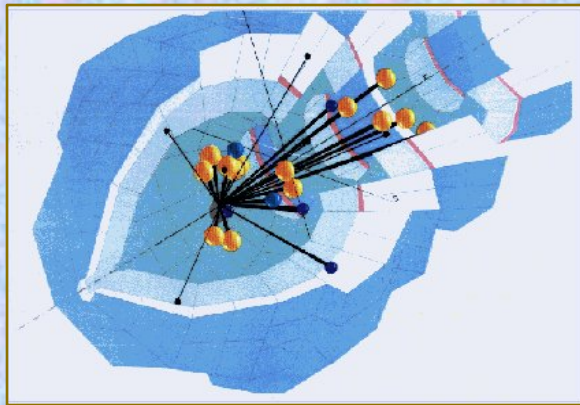


Multifragmentation experiment



■ Sort events in E^* (Calorimetry)

Multifragmentation experiment



- Sort events in E^* (Calorimetry)
- Reconstruct a freeze-out partition

1- Primary fragments:

$$m_i$$

2- Freeze-out volume:

$$E_{\text{coul}}$$

$$\bullet E_2 = \sum_i B_i + E_{\text{coul}}$$

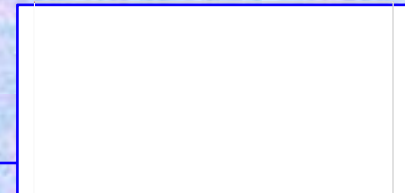
$$\bullet E_1 = E^* - E_2 \Rightarrow \langle E_1 \rangle, \sigma_1$$

3- Kinetic EOS:

$$T, C_1, \sigma_{\text{can}}^2 = C_1 T^2$$

$$\bullet \langle E_1 \rangle = \langle \sum_i a_i \rangle T^2 + 3/2 \langle M-1 \rangle T$$

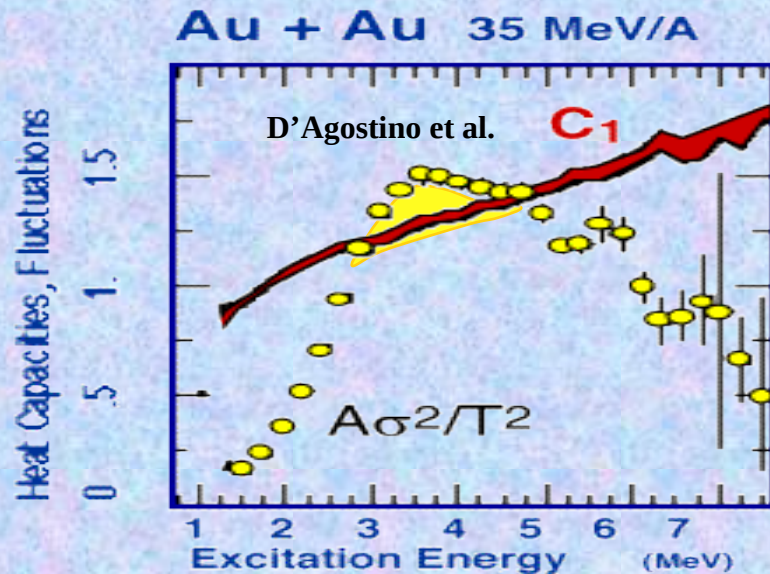
■ Deduce C:



Heat capacity from energy fluctuation

■ Large fluctuations

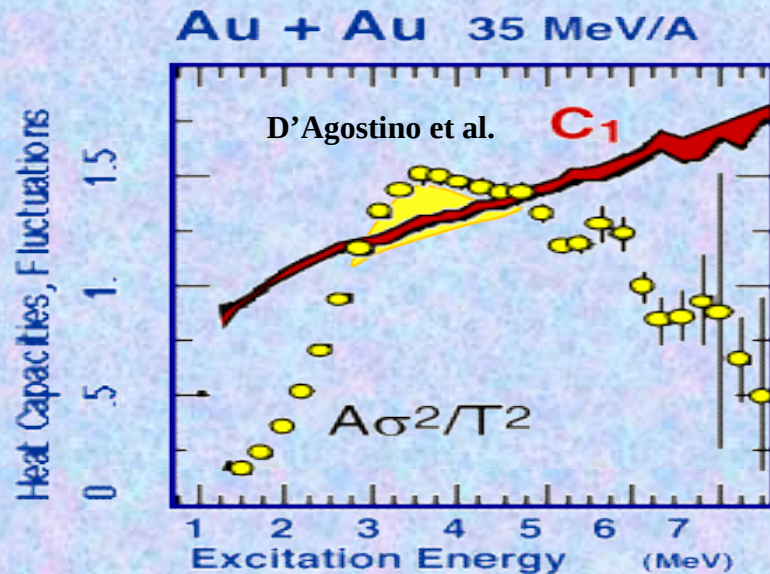
$$C = \frac{C_1^2}{C_1 - \sigma^2 / T^2}$$



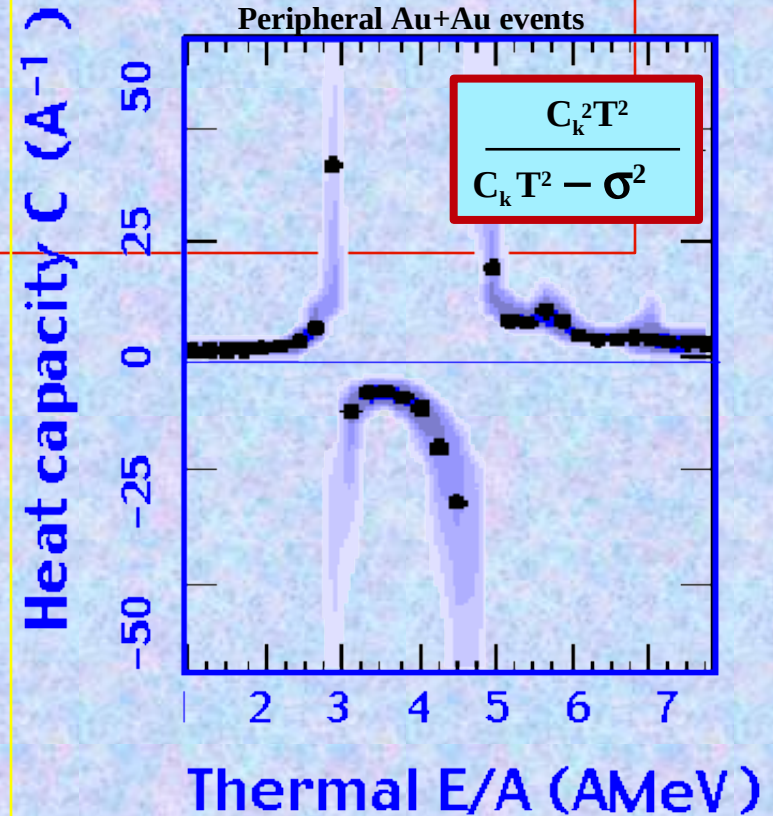
Heat capacity from energy fluctuation

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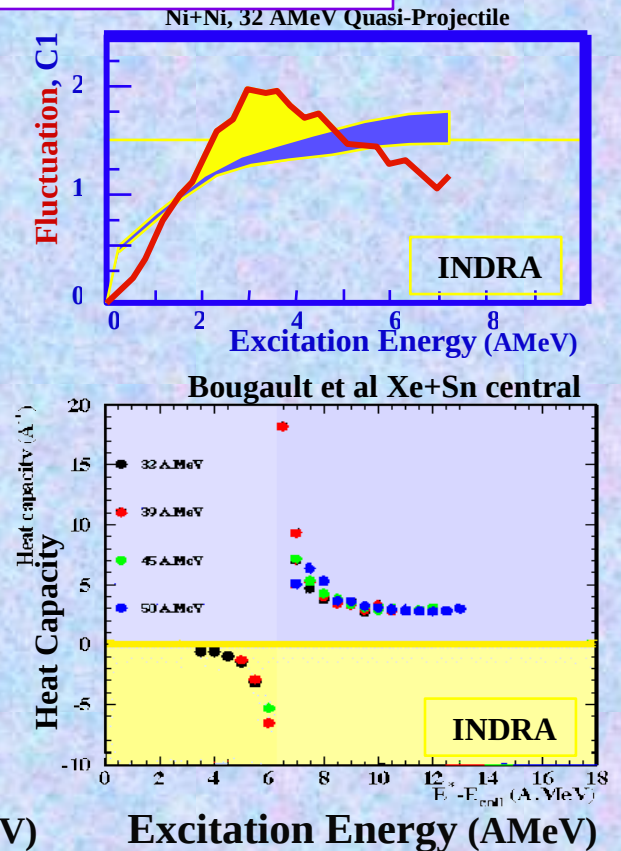
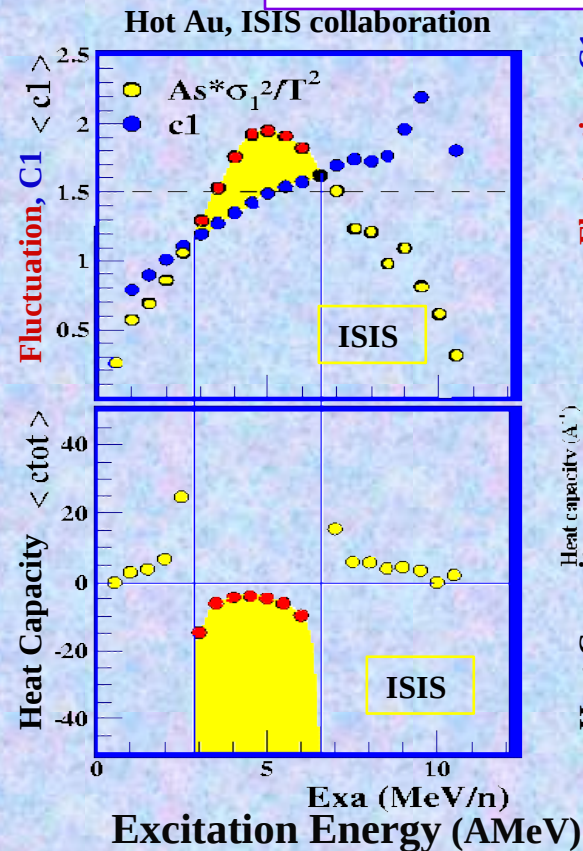
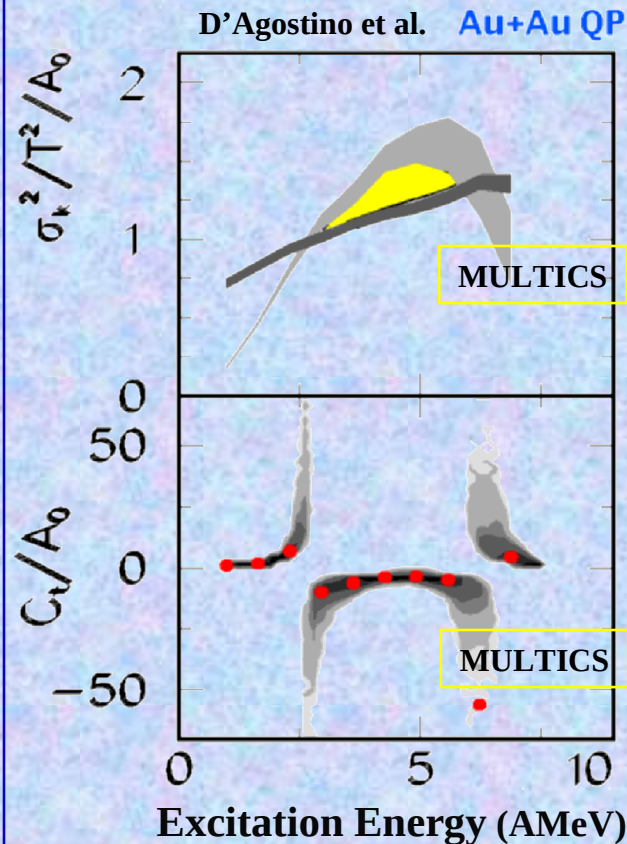


Negative C

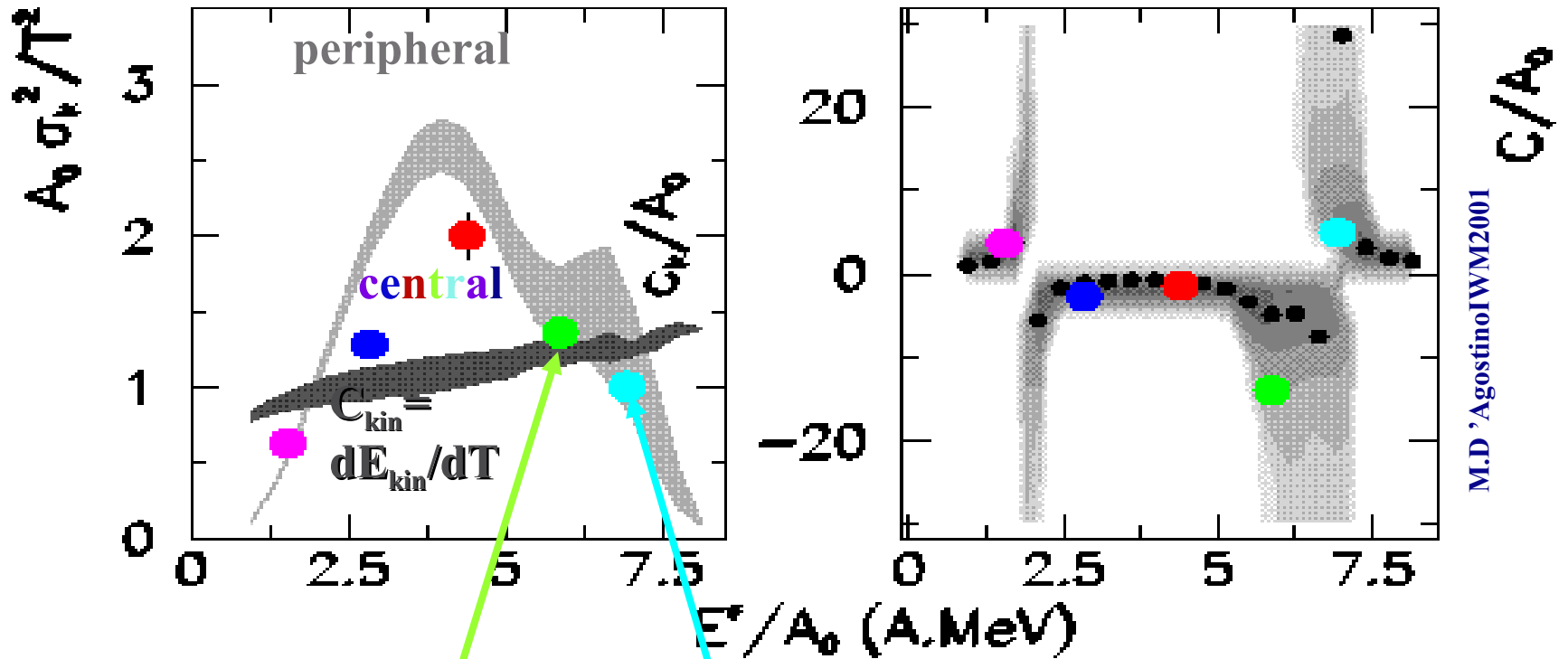


Heat capacity from energy fluctuation

Multics $E_1=2\pm0.3$ $E_2=6.5\pm0.7$
 Isis $E_1=2.5$ $E_2=7.$
 Indra $E_2=6.\pm0.5$



Comparison central / peripheral



Au+Au 35 $E_F=1$

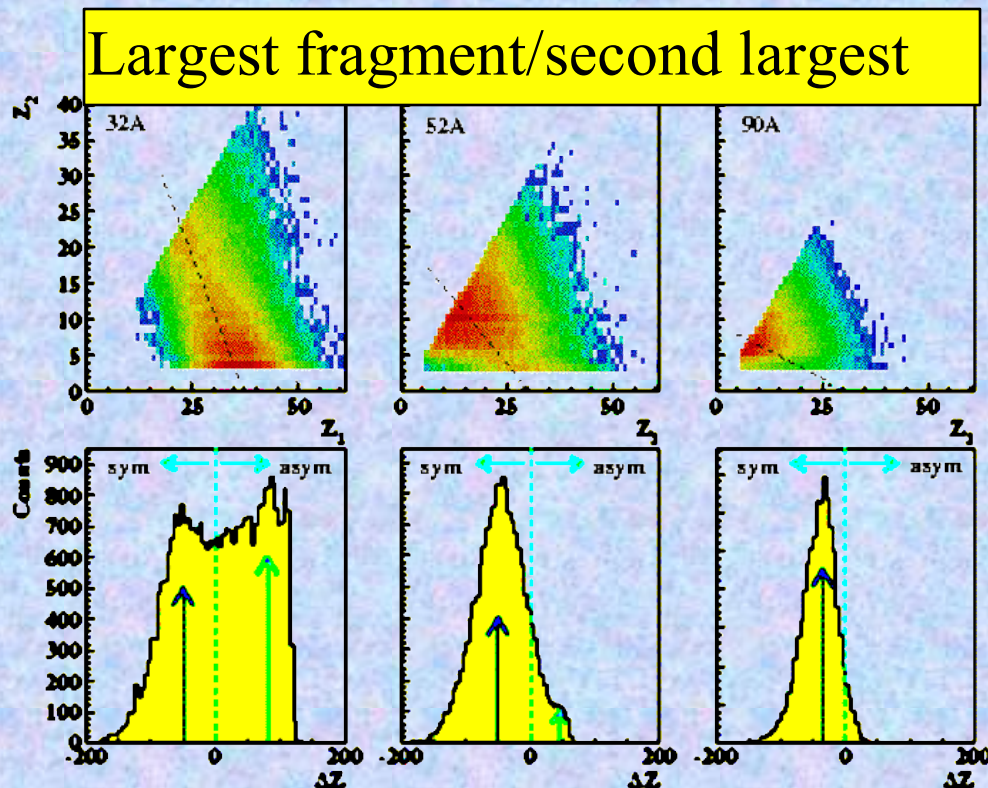
Au+Au 35 $E_F=0$

Up to ≈ 35 A.MeV the flow ambiguity
is a small effect

Multifragmentation of nuclei

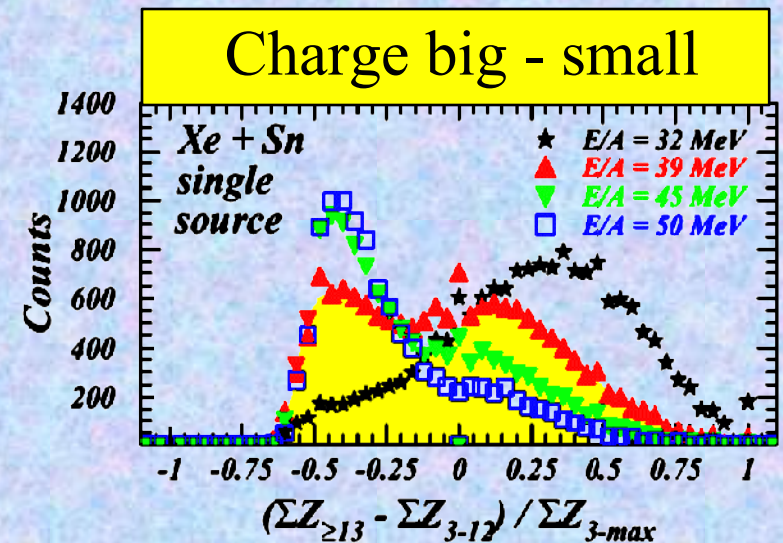
Bimodality in event distribution

Distribution of events.



Ni+Au INDRA data

O.Lopez 2001



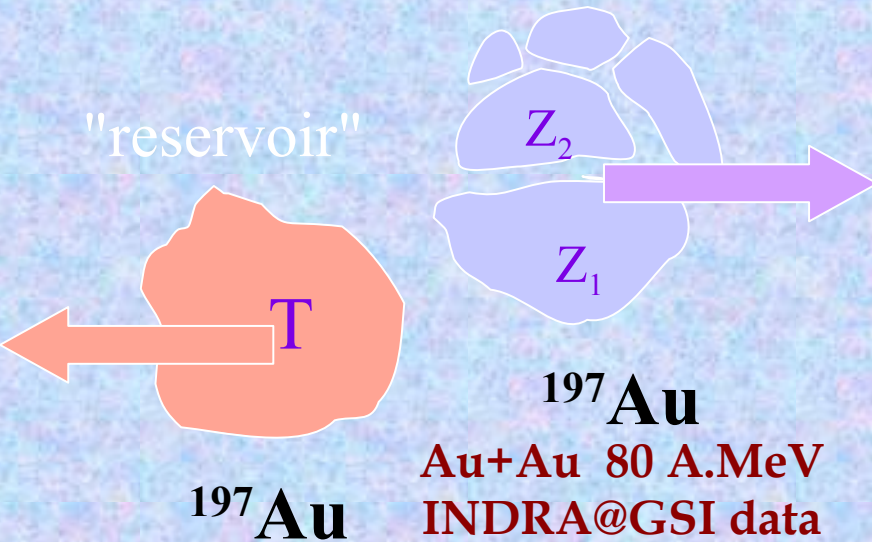
Xe+Sn INDRA data B. Borderie, 2001

Multifragmentation of nuclei

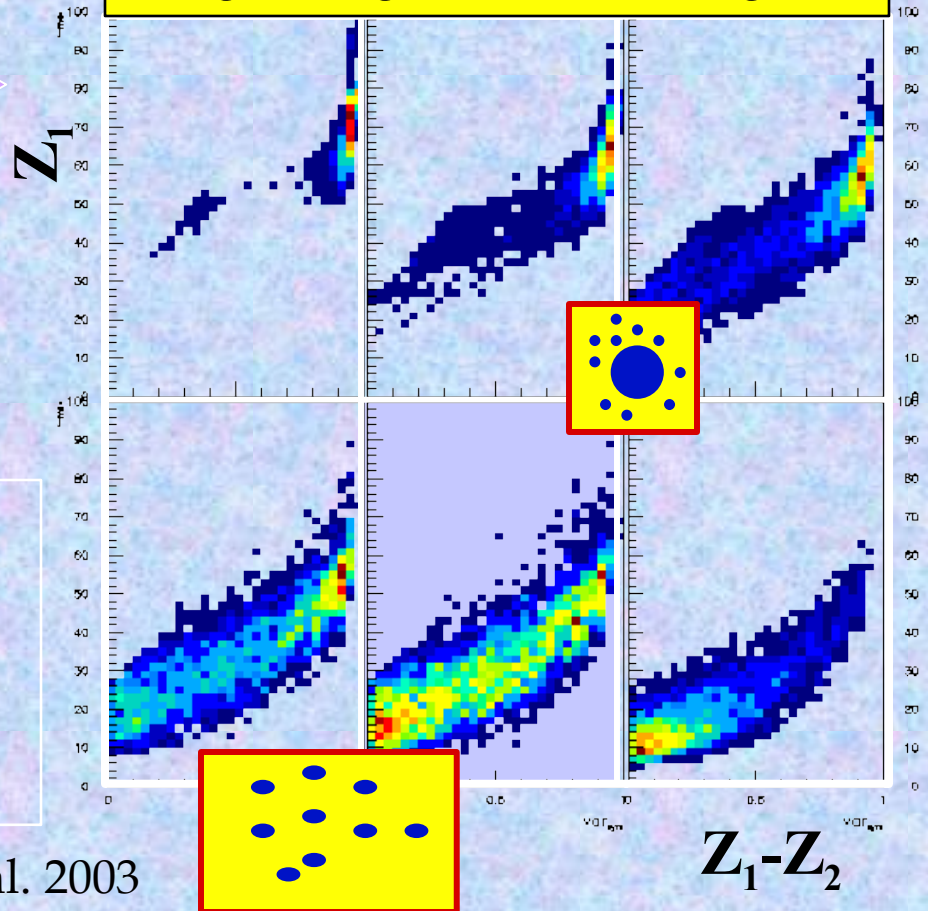
Bimodality in event distribution

Multifragmentation of nuclei

Bimodality in event distribution



Largest fragment/second largest



- Order parameter
- ◆ $Z_1 - Z_2$ (big-small)
- ◆ cf

O.Lopez et al. 2001, B.Tamain et al. 2003



Conclusion

Negative heat capacities

Bimodality, fluctuations

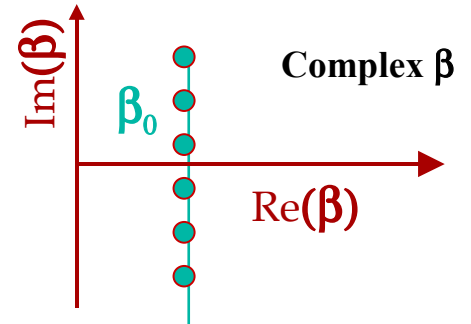
1st order in finite systems

PC & Gulminelli Phys A (2003)

■ Order param. free (canonical)

◆ Zeroes of Z reach real β axis

Yang & Lee Phys Rev 87(1952)404



1st order in finite systems

PC & Gulminelli Phys A (2003)

■ Order par. free (canonical)

- ◆ Zeroes of Z reach real axis

Yang & Lee Phys Rev 87(1952)404

- ◆ Bimodal E distribution ($P_\beta(E)$)

K.C. Lee Phys Rev E 53 (1996) 6558

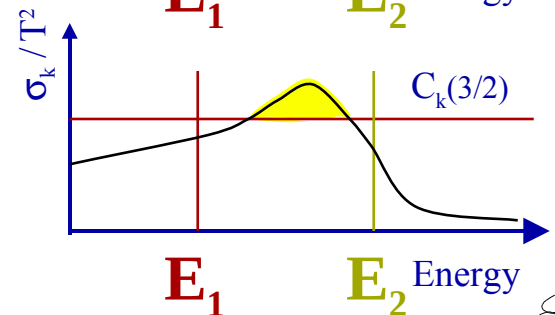
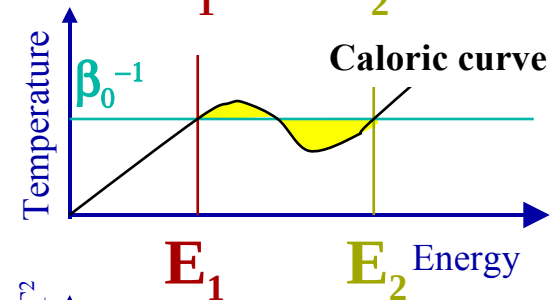
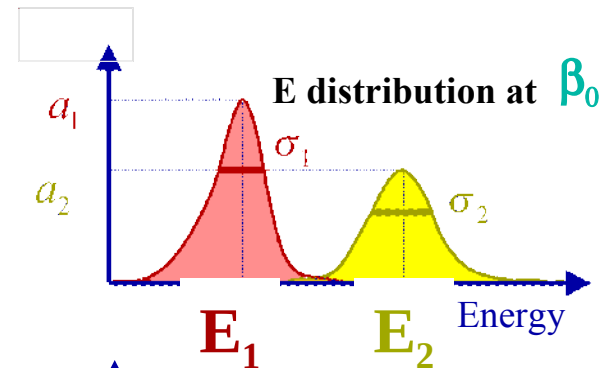
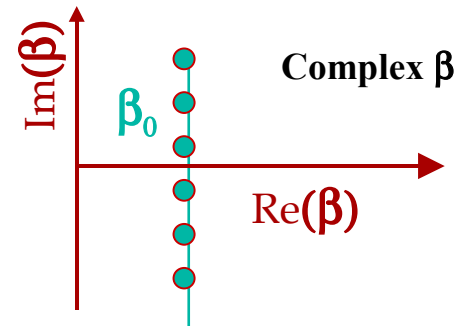
■ Order par. fixed (microcanonical)

- ◆ Back Bending in EOS ($T(E)$)

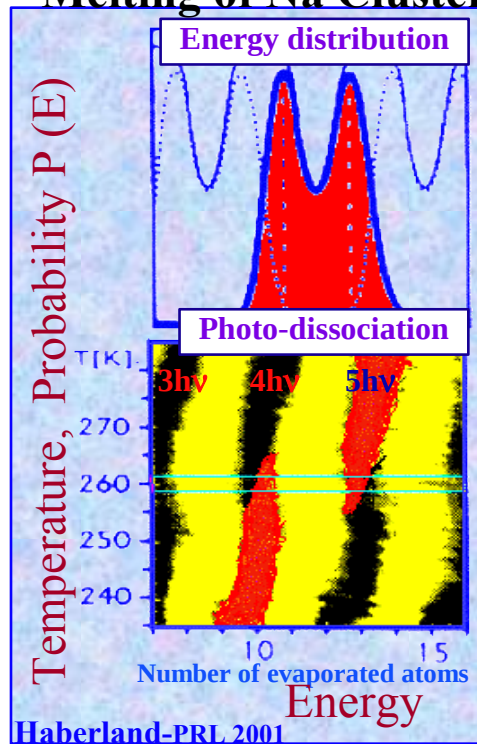
K. Binder, D.P. Landau Phys Rev B30 (1984) 1477

- ◆ Abnormal fluctuation ($\sigma_k(E)$)

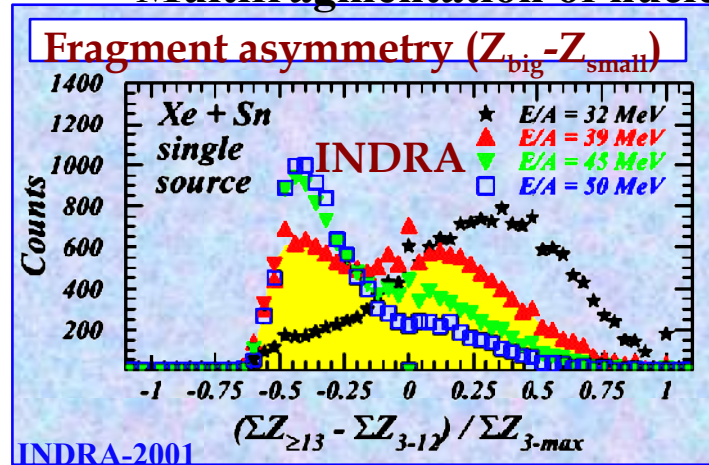
J.L. Lebowitz (1967), PC & Gulminelli, NPA 647(1999)153



Melting of Na Cluster

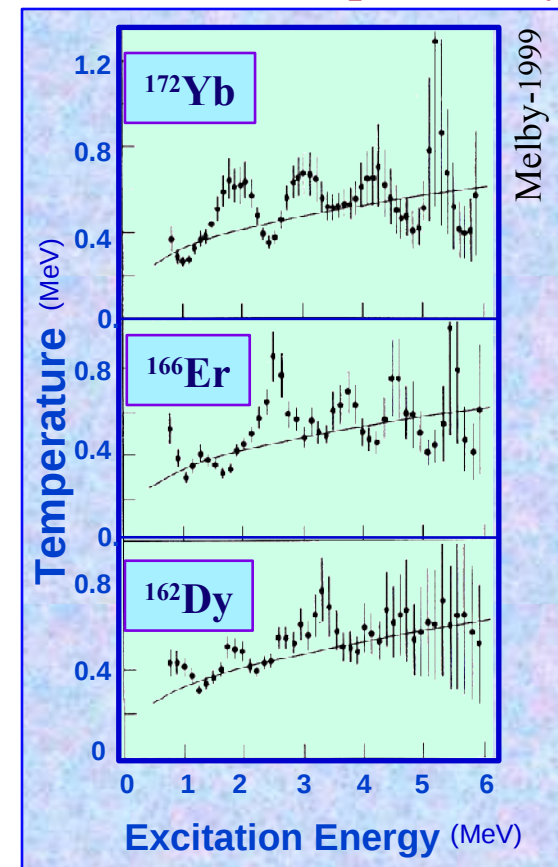


Multifragmentation of nuclei

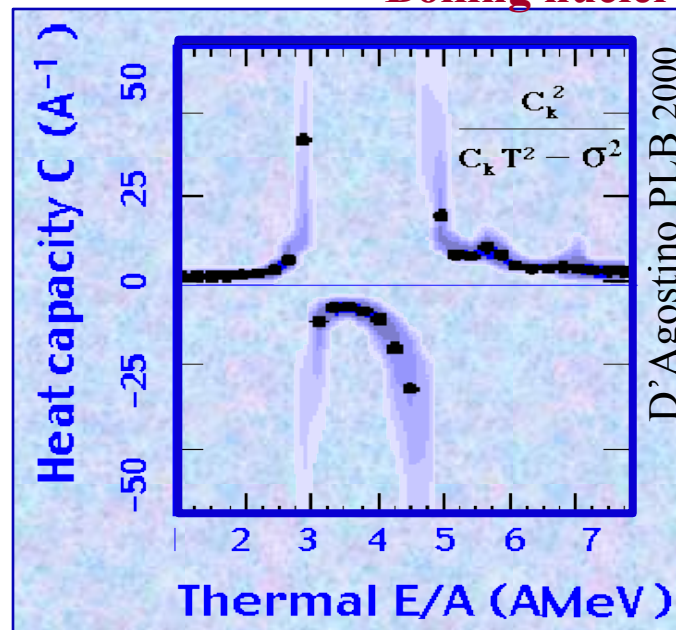


Seen in experiments

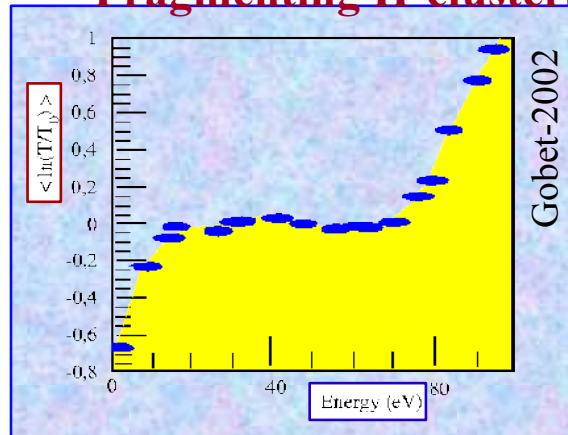
Nuclear superfluidity



Boiling nuclei



Fragmenting H-clusters



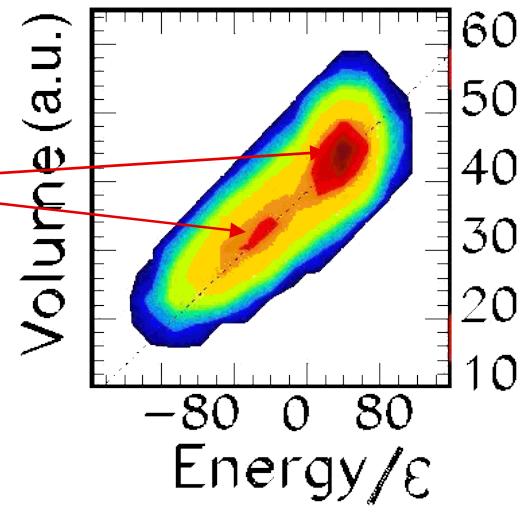
Liquid-gas



Isobare
ensemble

$$P^{(i)} \propto \exp(-\lambda V^{(i)})$$

● Phase transition and bimodality



Liquid-gas

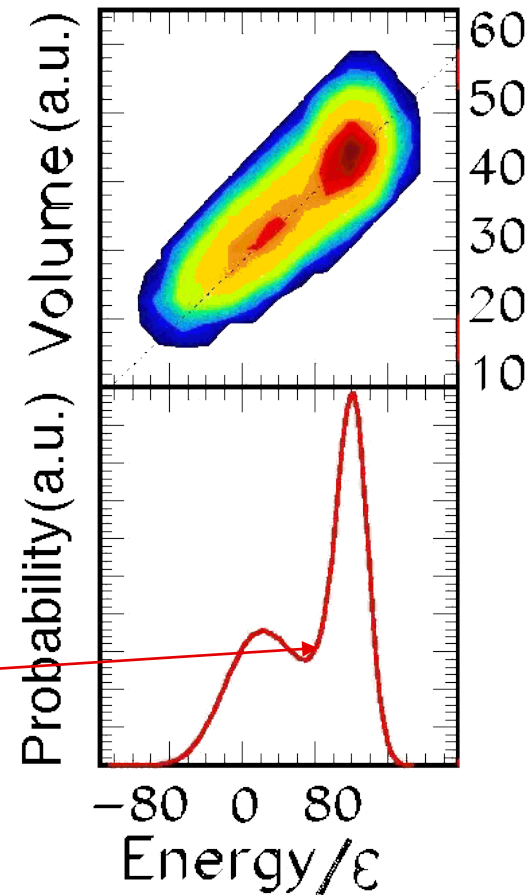


Isobare
ensemble

$$P^{(i)} \propto \exp(-\lambda V^{(i)})$$

- Phase transition and bimodality

- Negative heat capacity



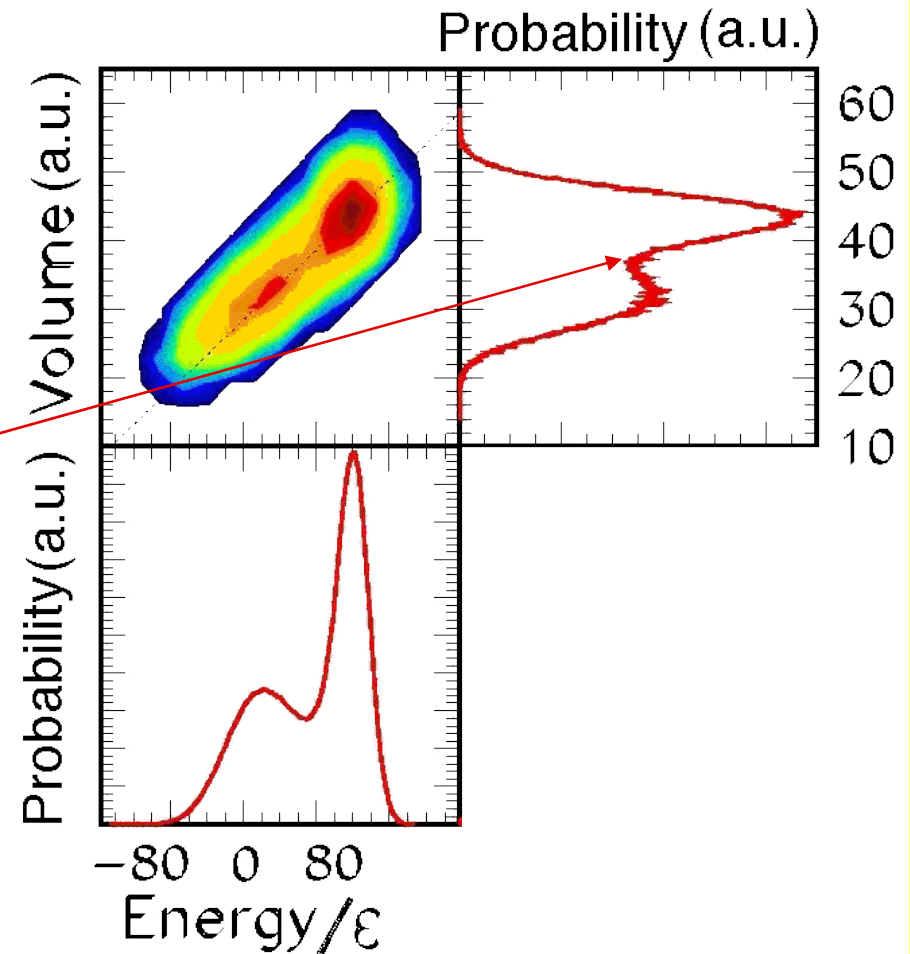
Liquid-gas



Isobare
ensemble

$$P(i) \propto \exp(-\lambda V(i))$$

- **Phase transition
And bimodality**
- **Negative
compressibility**
- **Negative heat
capacity**



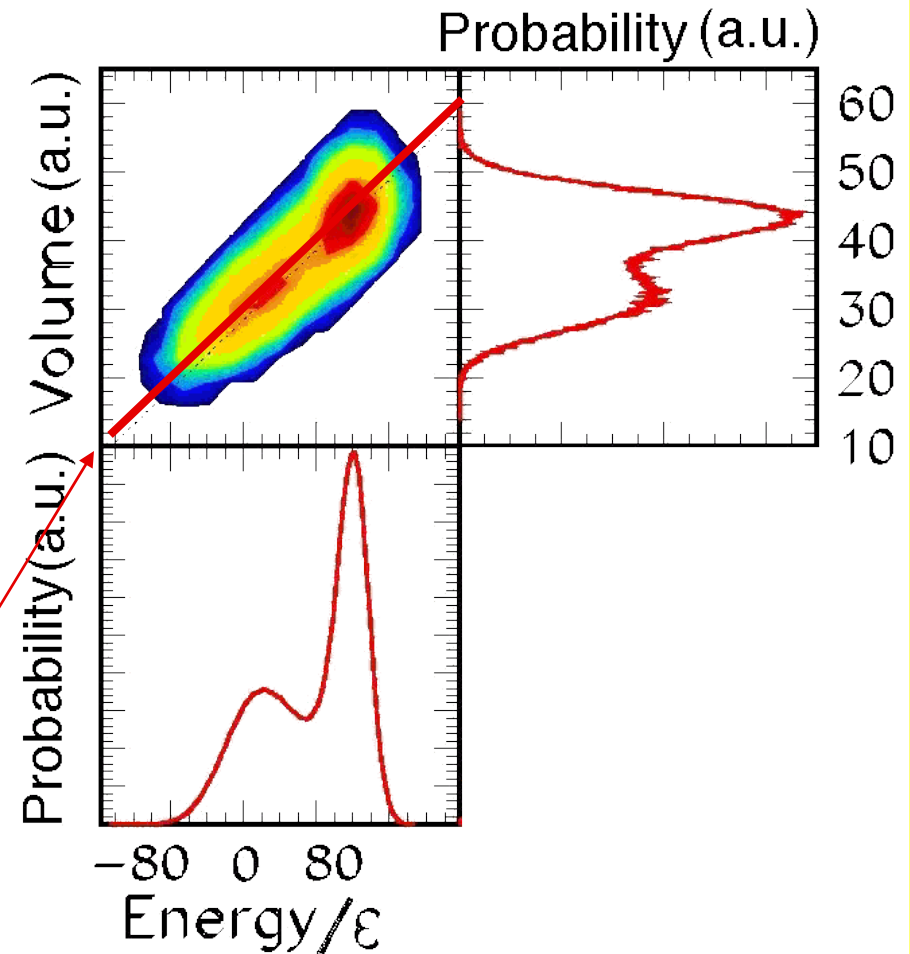
Liquid-gas



Isobare
ensemble

$$P(i) \propto \exp(-\lambda V(i))$$

- Phase transition
And bimodality
- Negative
compressibility
- Negative heat
capacity
- Order parameter



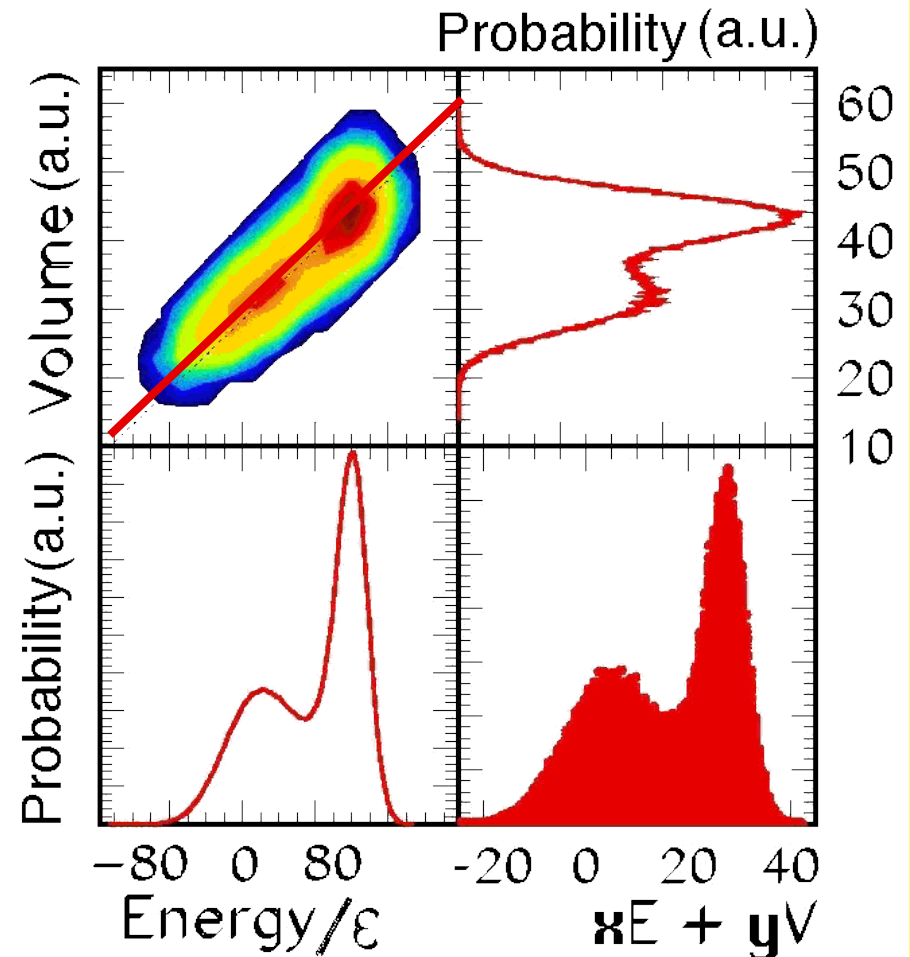
Liquid-gas



Isobare
ensemble

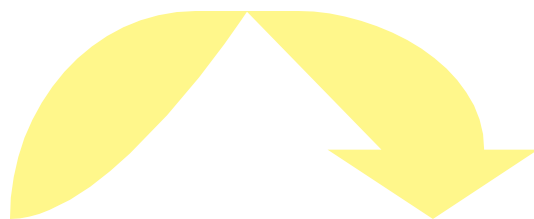
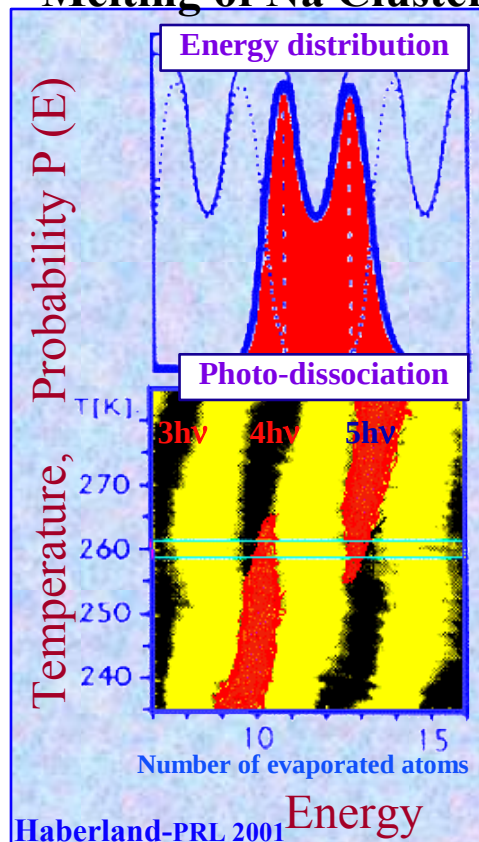
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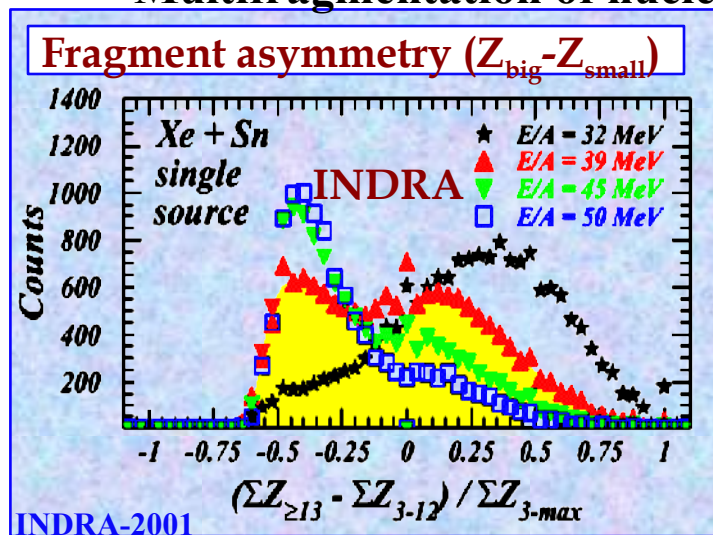


Bimodal distributions

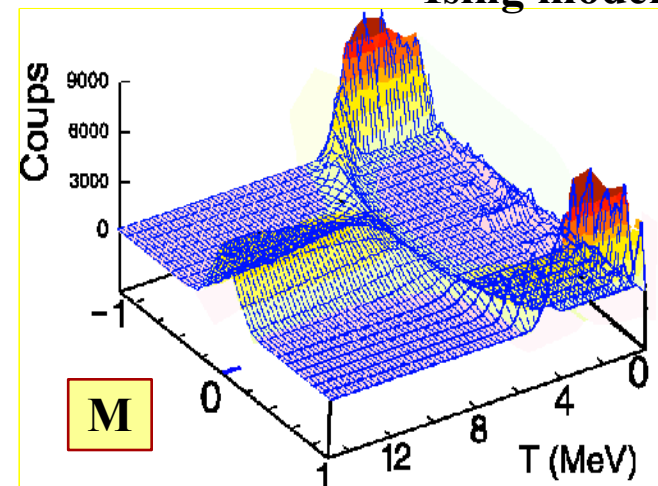
Melting of Na Cluster



Multifragmentation of nuclei



Ising model



Liquid-gas in lattice-gas

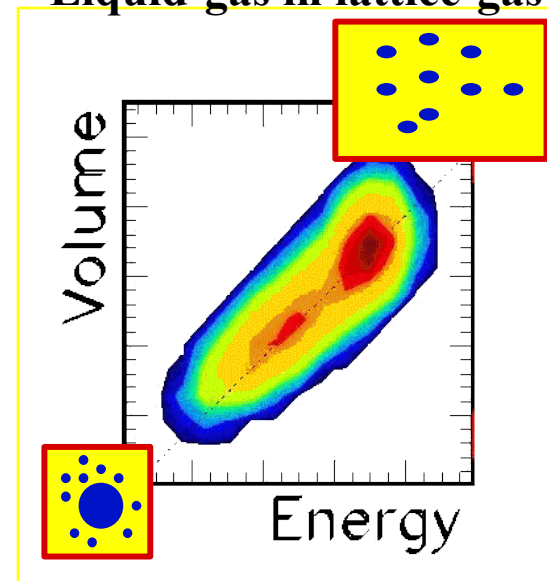
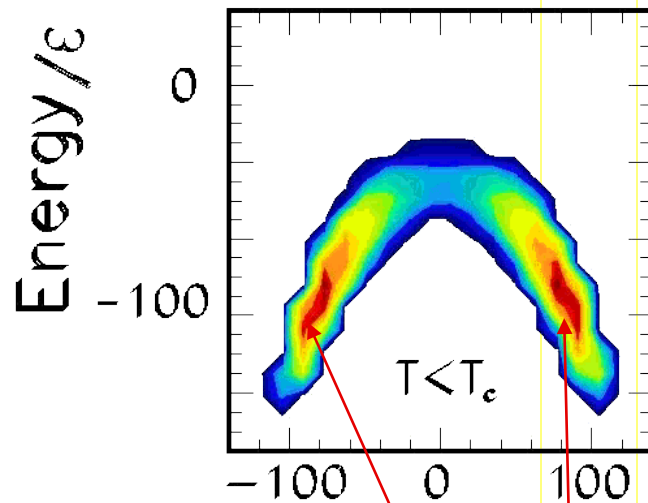


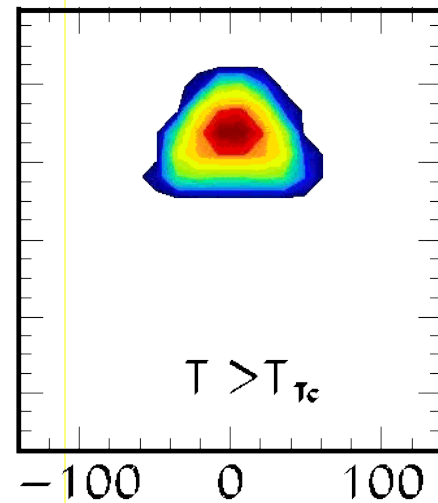
Figure 1 shows a contour plot of the energy landscape $E(\phi)$ for the 2D XY model. The vertical axis is labeled Energy/ϵ and ranges from -100 to 0. The horizontal axis is labeled ϕ and ranges from -100 to 100. A red arrow points to a local minimum at $\phi \approx 0$, which is labeled $T > T_{Tc}$. The energy landscape shows a central minimum surrounded by a ring of higher energy states, with a saddle point at $\phi \approx \pm 90$.



Magnetization in Ising Model



Magnetization

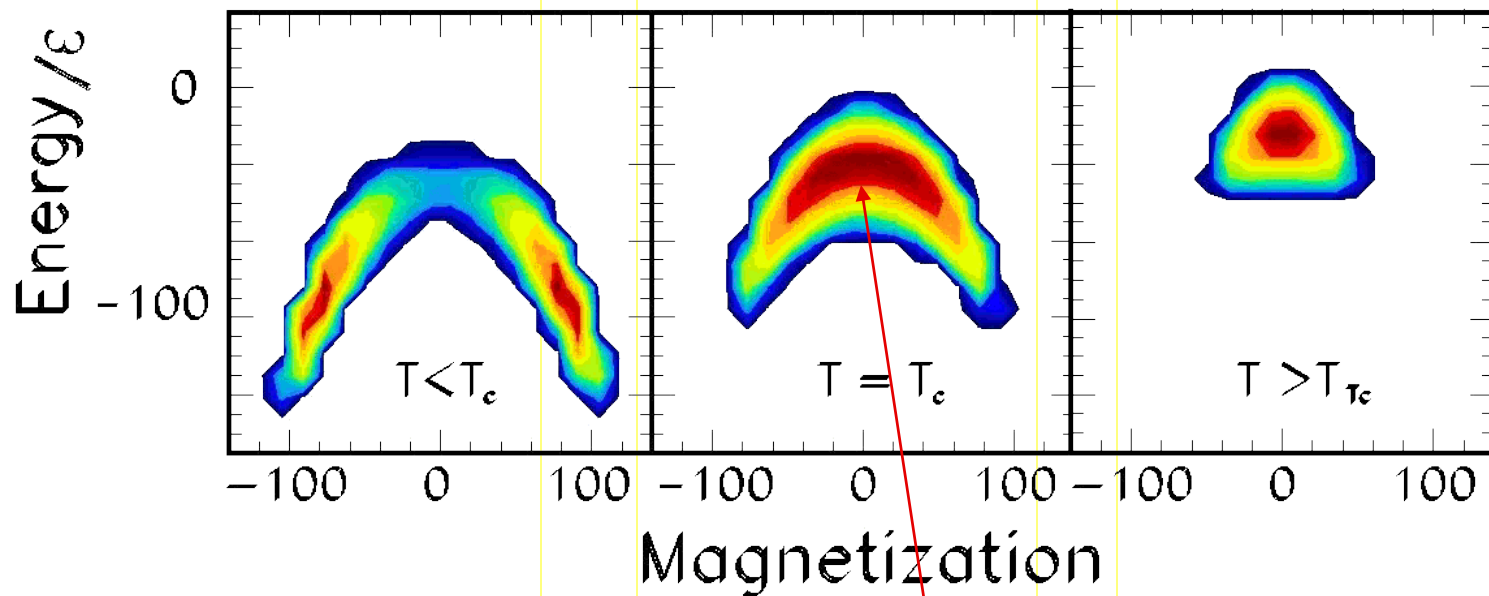


Bimodal Two Phases



One Phase

Magnetization in Ising Model

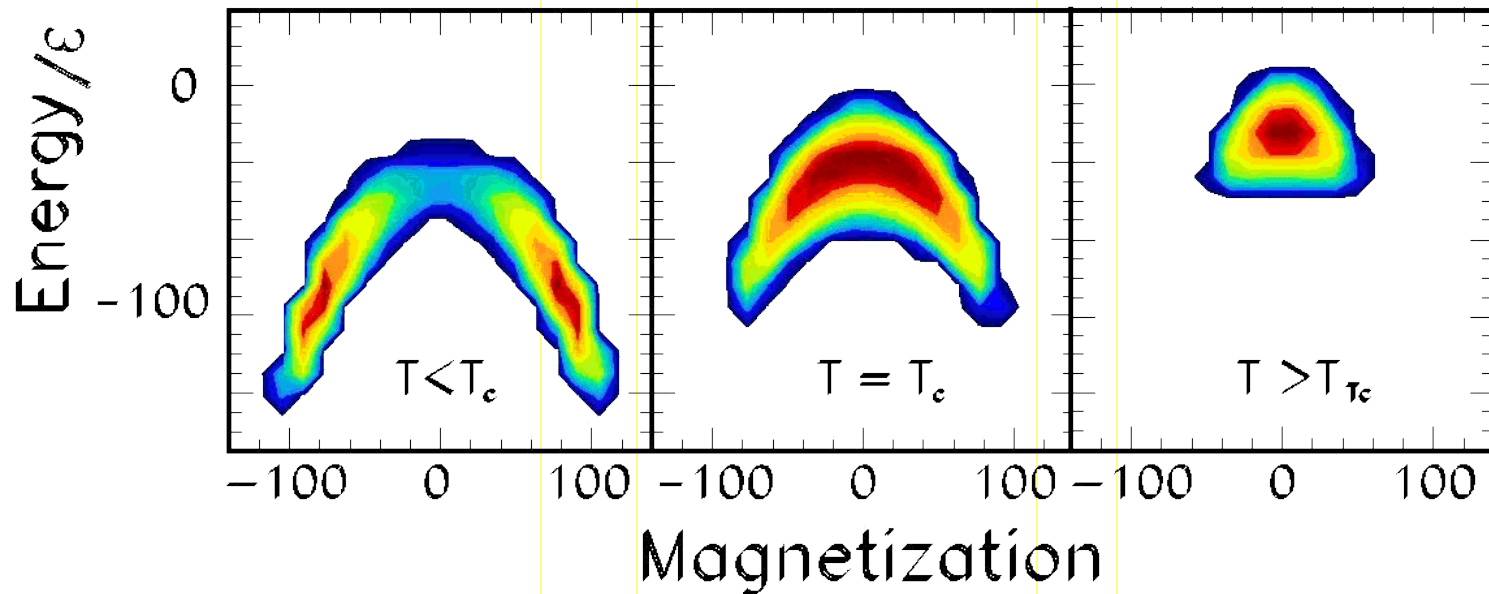


● Bimodal
Two Phases

● Second
Order

● One Phase

Magnetization in Ising Model



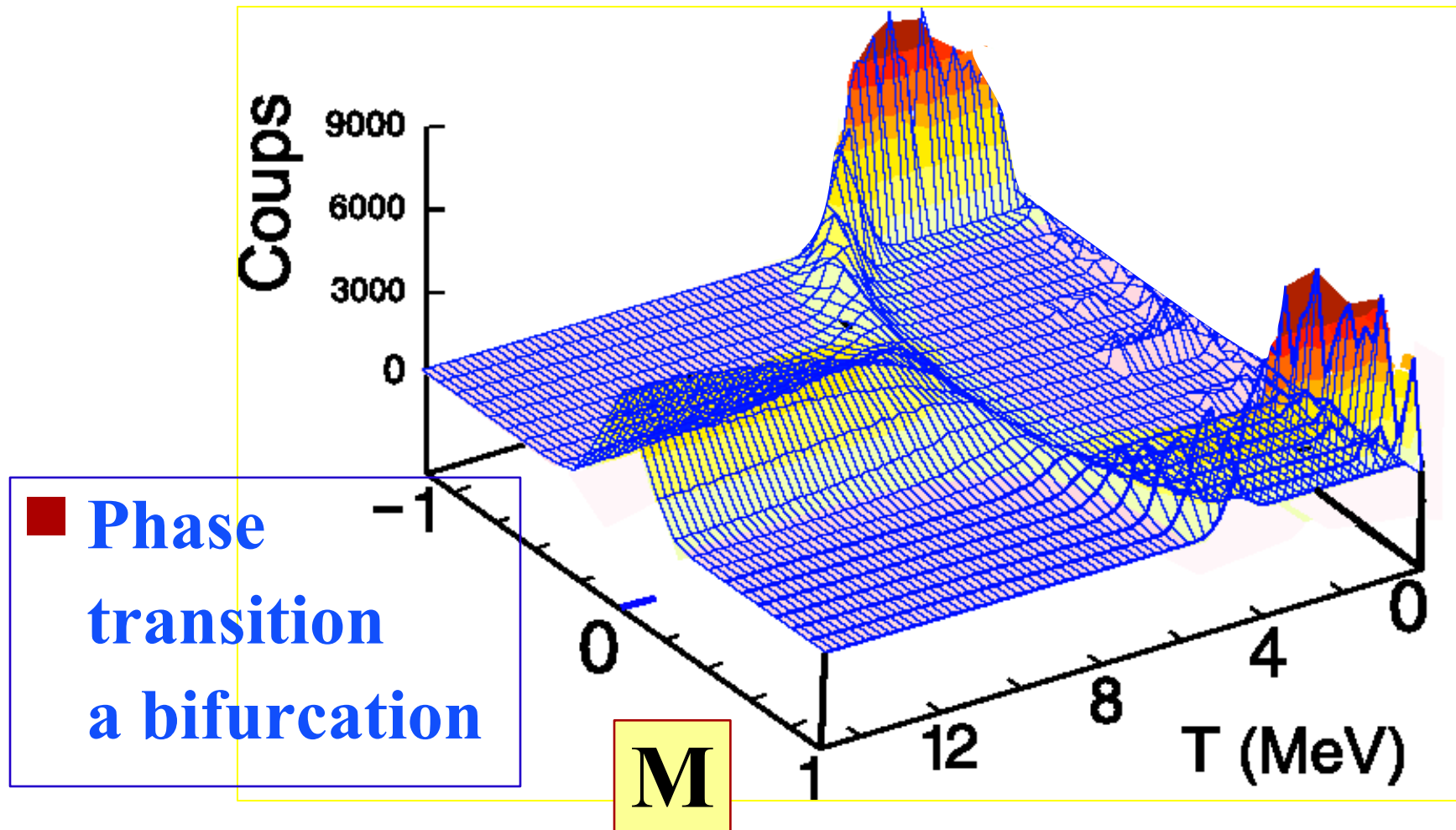
Bimodal Two Phases



Second Order

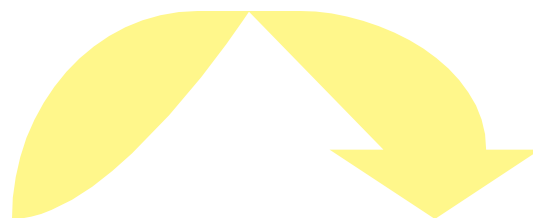
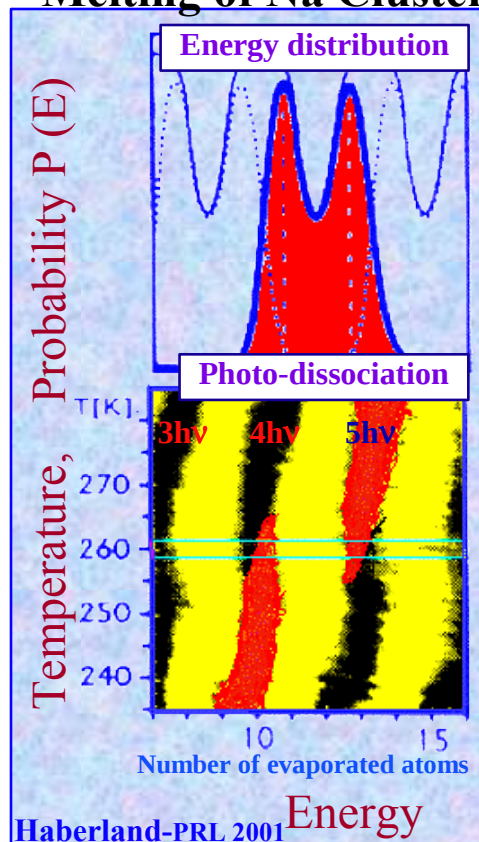
 One Phase

Magnetization in Ising Model

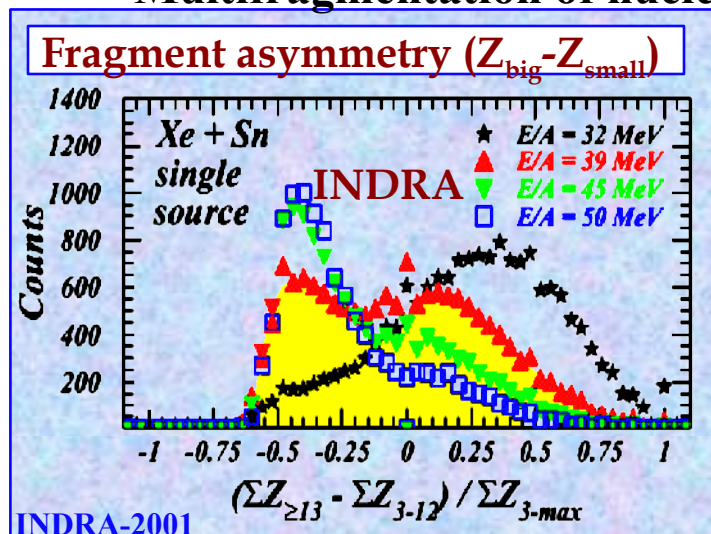


Bimodal distributions

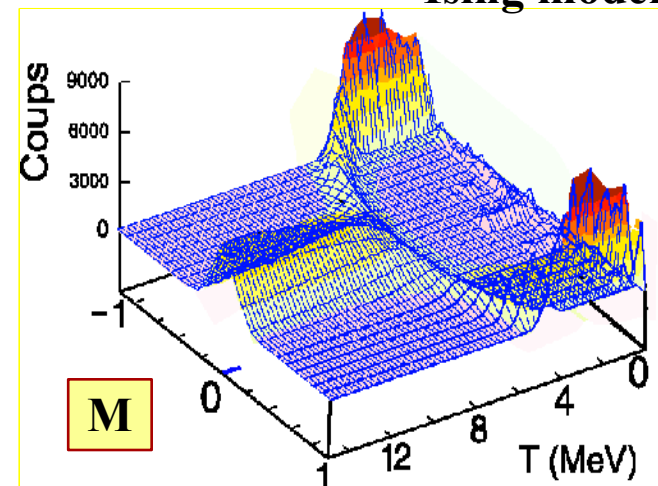
Melting of Na Cluster



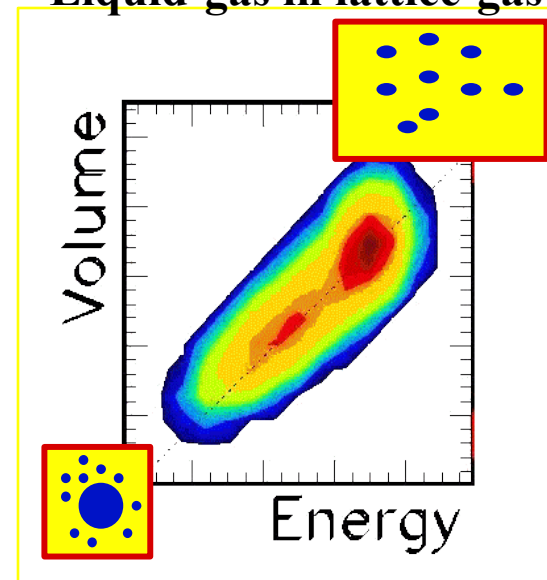
Multiframegmentation of nuclei



Ising model



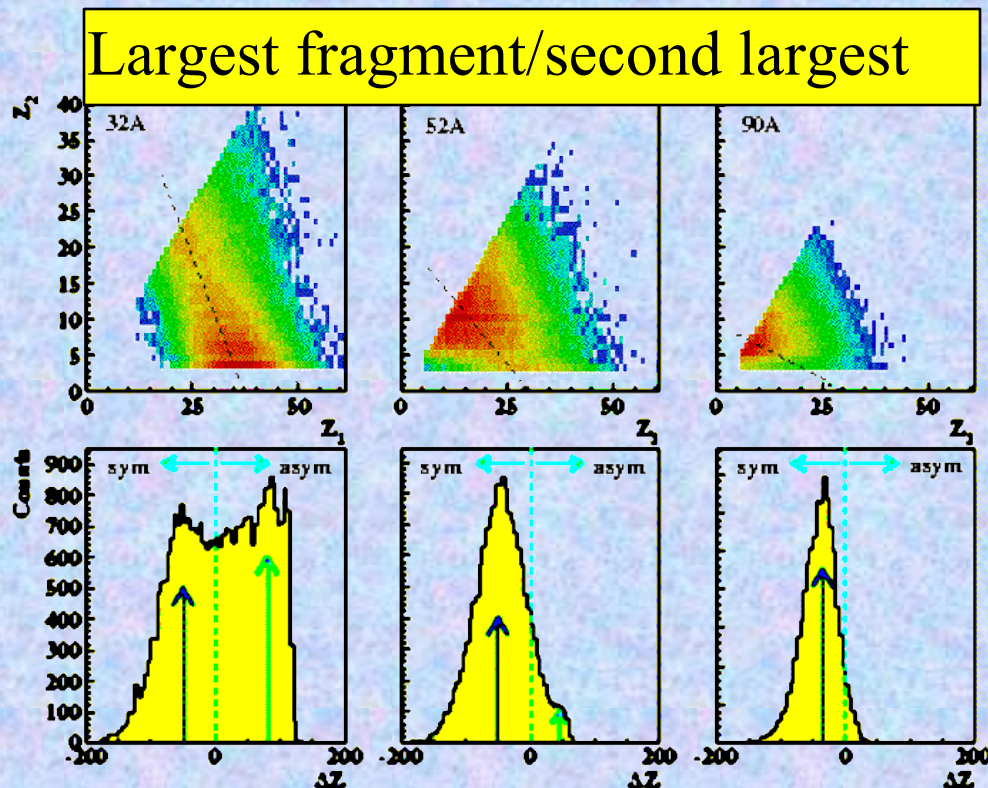
Liquid-gas in lattice-gas



Multifragmentation of nuclei

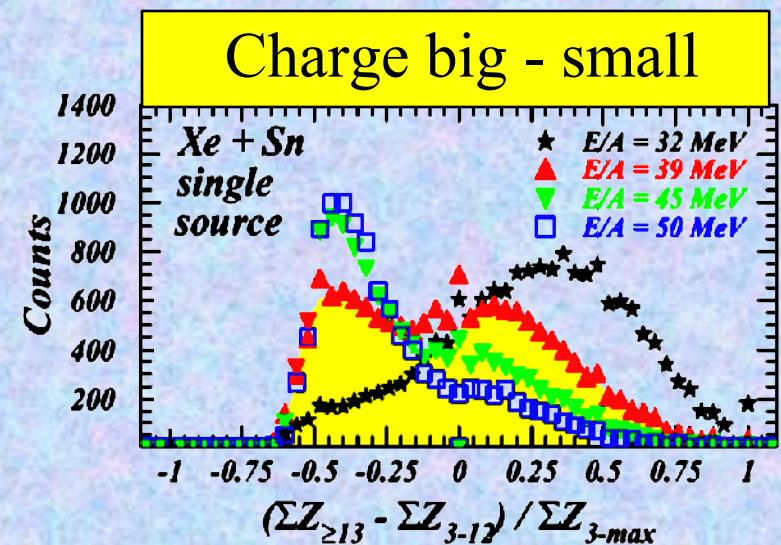
Bimodality in event distribution

Distribution of events.



Ni+Au INDRA data

O.Lopez 2001



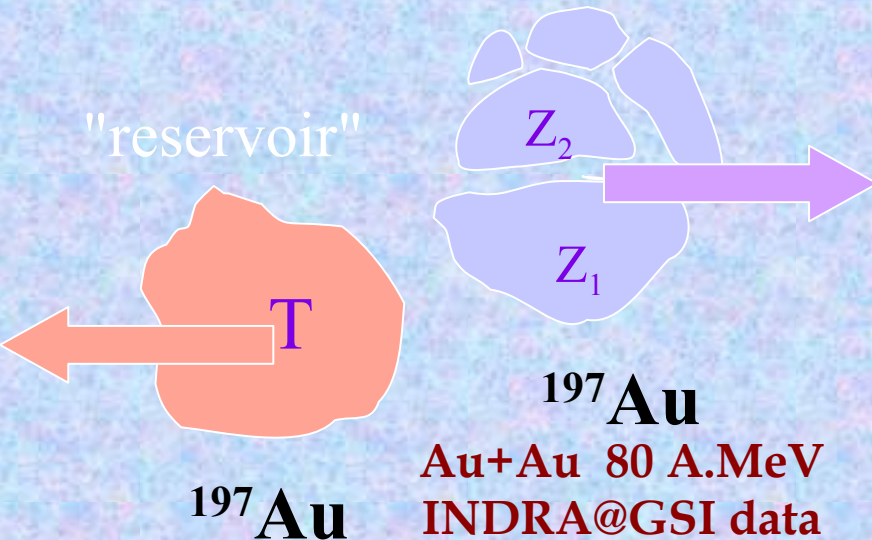
Xe+Sn INDRA data B. Borderie, 2001

Multifragmentation of nuclei

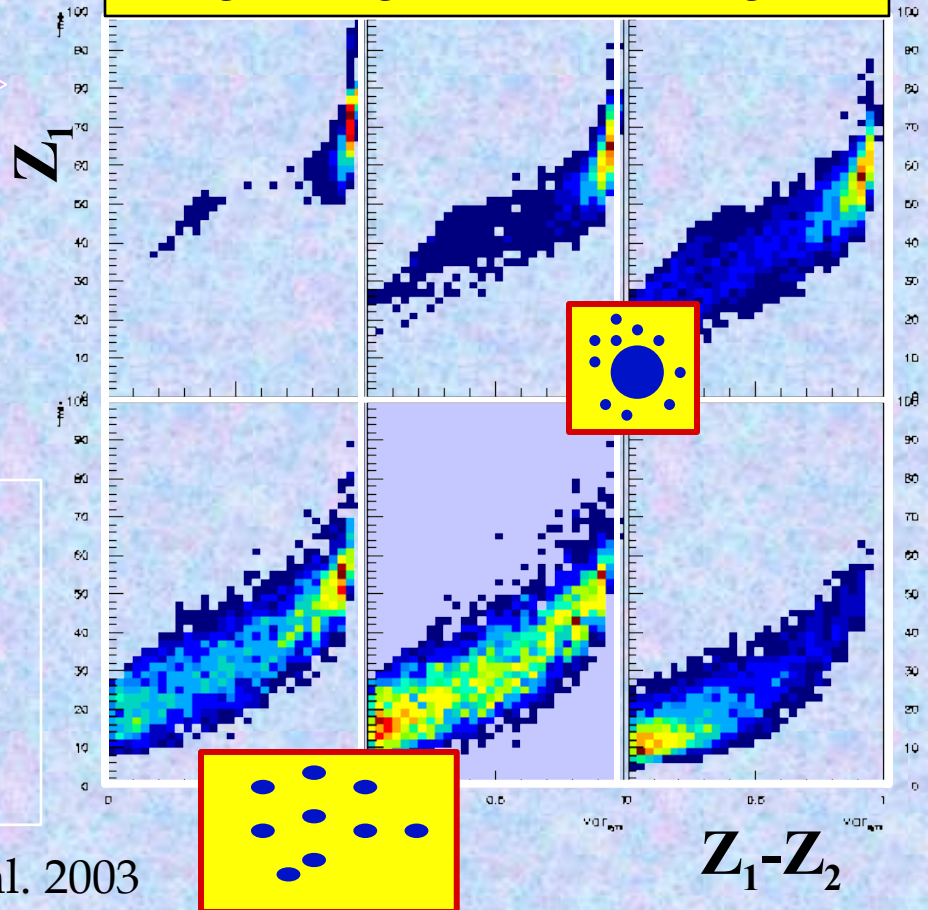
Bimodality in event distribution

Multifragmentation of nuclei

Bimodality in event distribution



Largest fragment/second largest

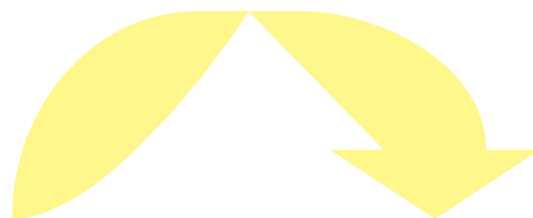
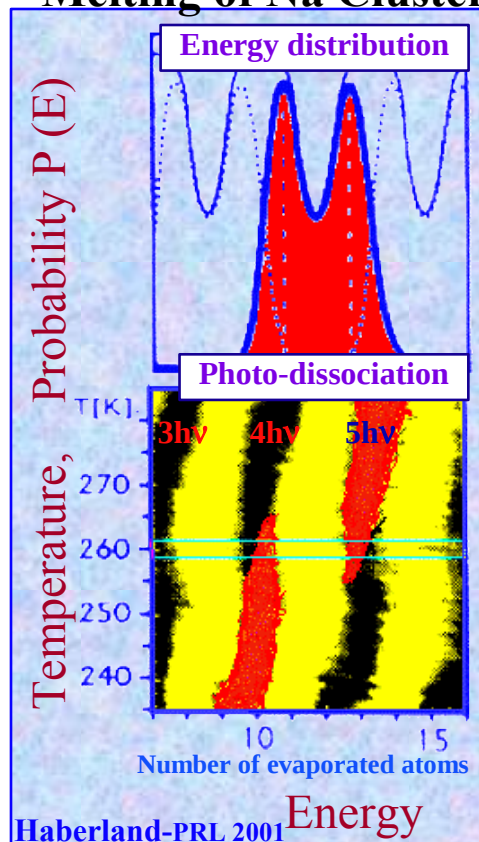


- Order parameter
- ◆ $Z_1 - Z_2$ (big-small)
- ◆ cf

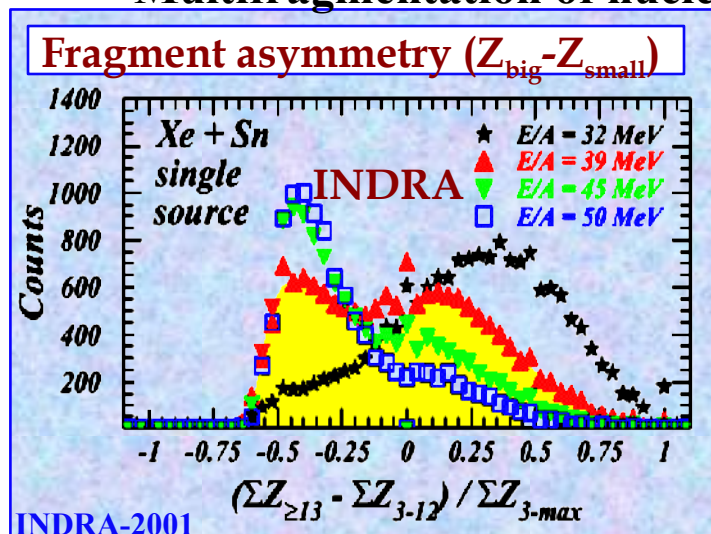
O.Lopez et al. 2001, B.Tamain et al. 2003

Bimodal distributions

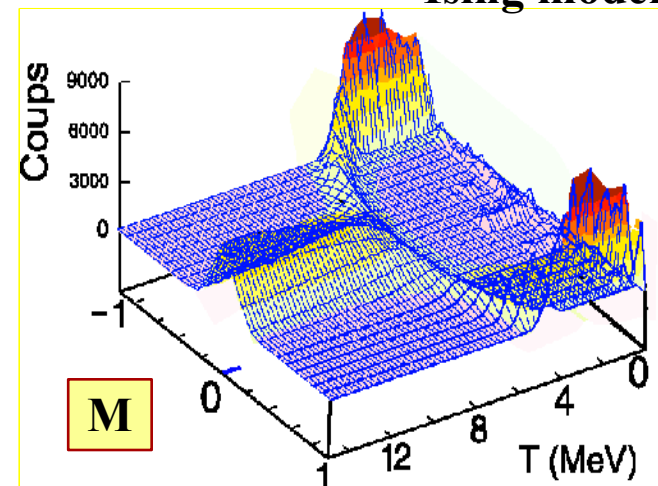
Melting of Na Cluster



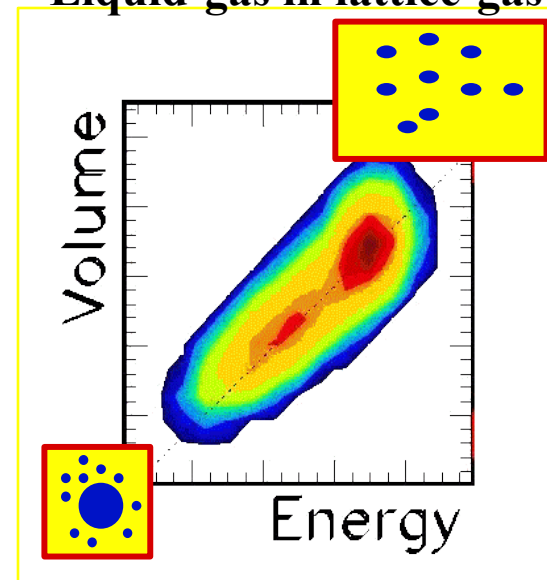
Multiframegmentation of nuclei



Ising model



Liquid-gas in lattice-gas



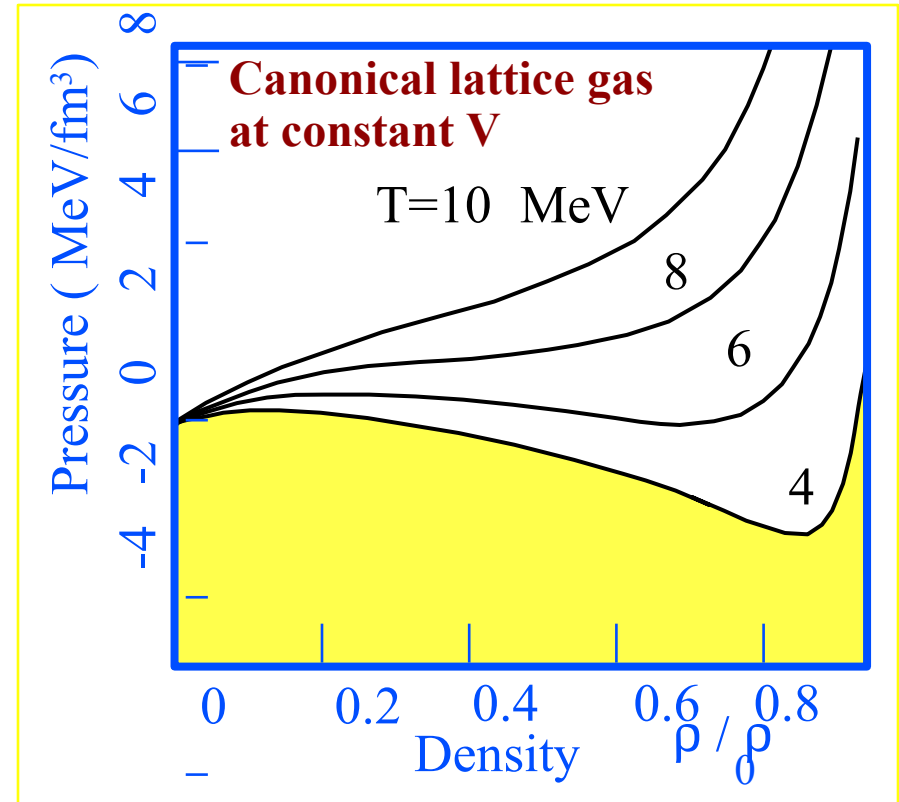
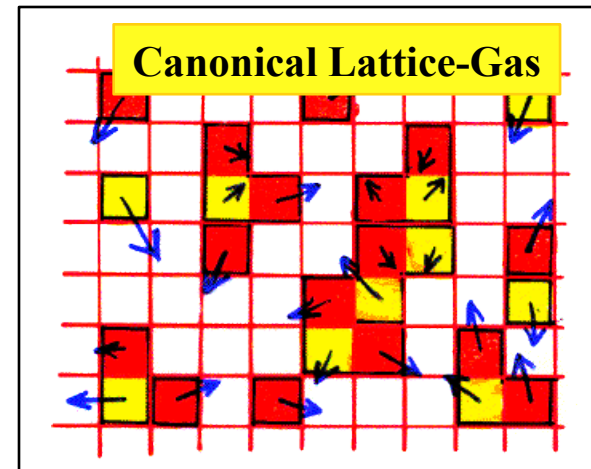
Density: Order parameter:

■ Usually $V = \text{cst}$

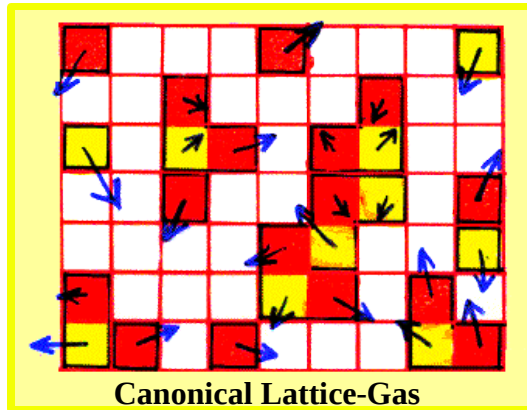
◆ (“box”)

◆ Isochore
ensemble

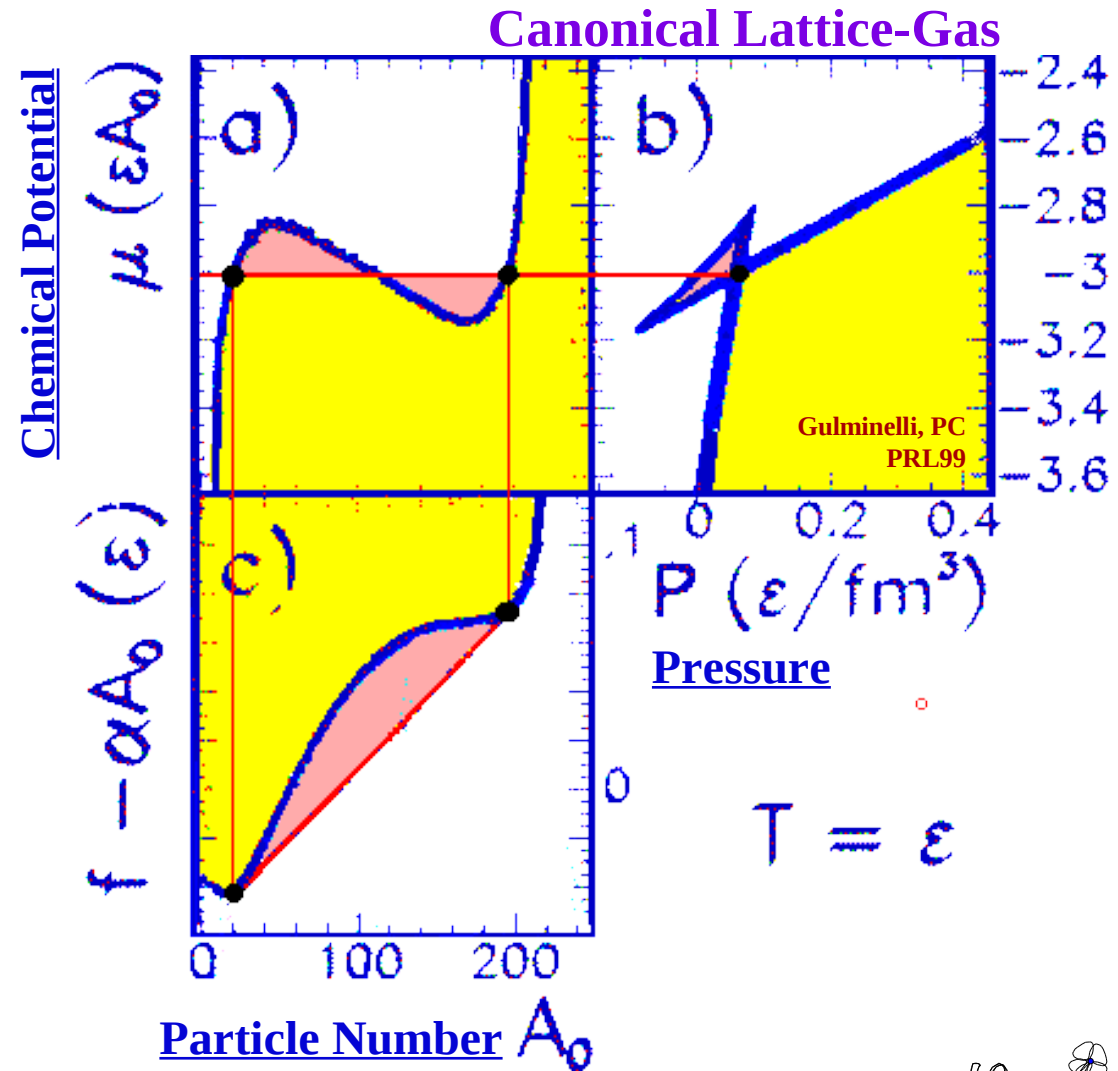
■ Negative
compressibility



Generalization: inverted curvatures

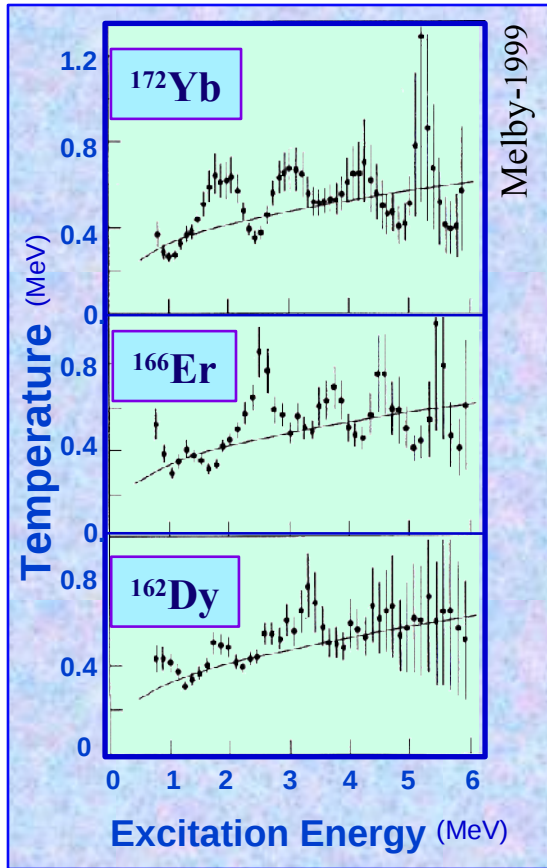


**Negative
Susceptibility
and
Compressibility**



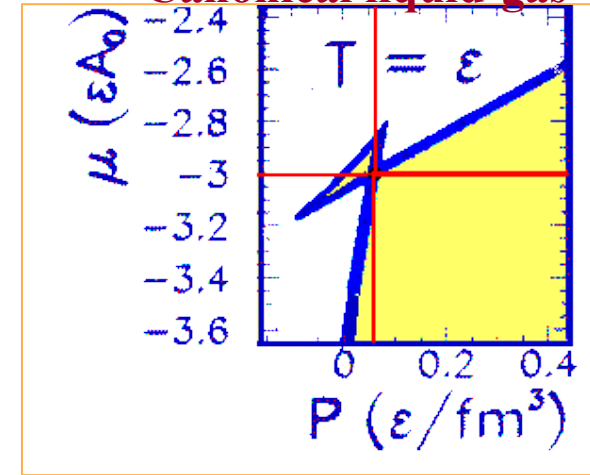
Abnormal curvatures

Nuclear superfluidity



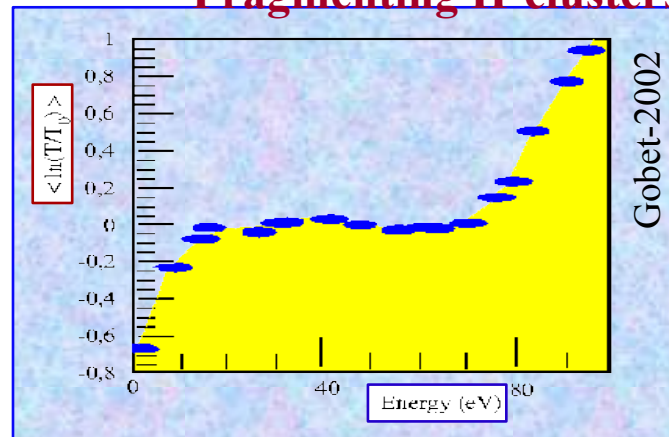
Convex entropy

Canonical liquid-gas



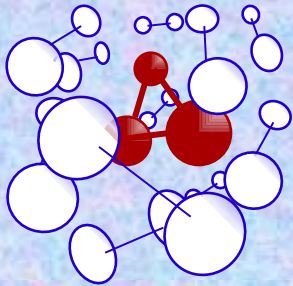
Compressibility < 0
Susceptibility < 0

Fragmenting H-clusters



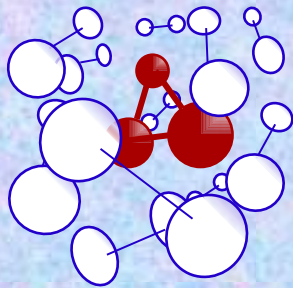
T decreasing with E

Fragmentation of hydrogen clusters



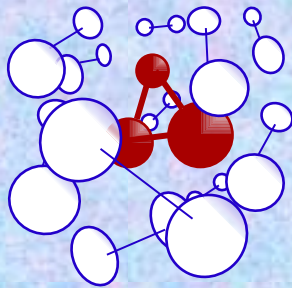
He

Fragmentation of hydrogen clusters



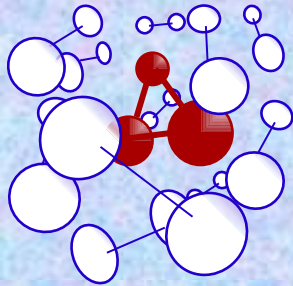
He

Fragmentation of hydrogen clusters



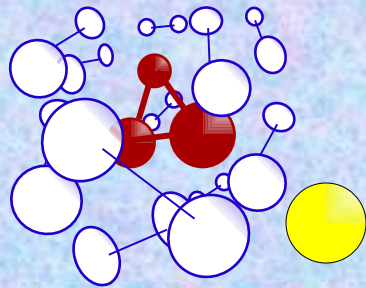
He

Fragmentation of hydrogen clusters



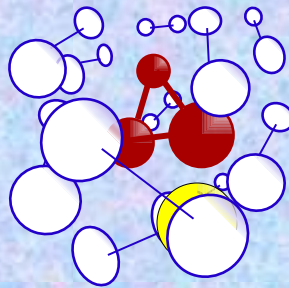
He

Fragmentation of hydrogen clusters



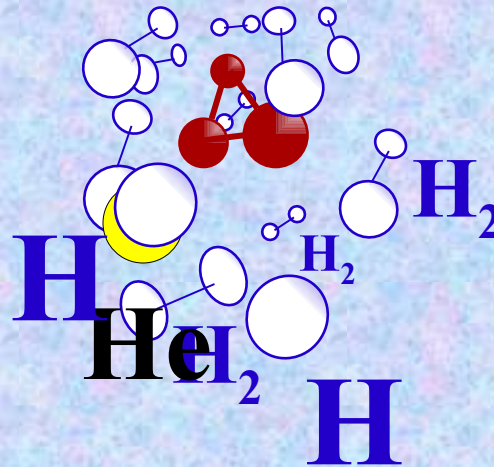
He

Fragmentation of hydrogen clusters

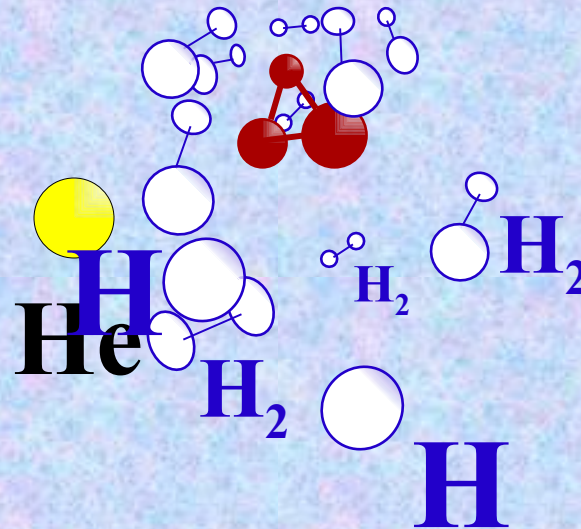


He

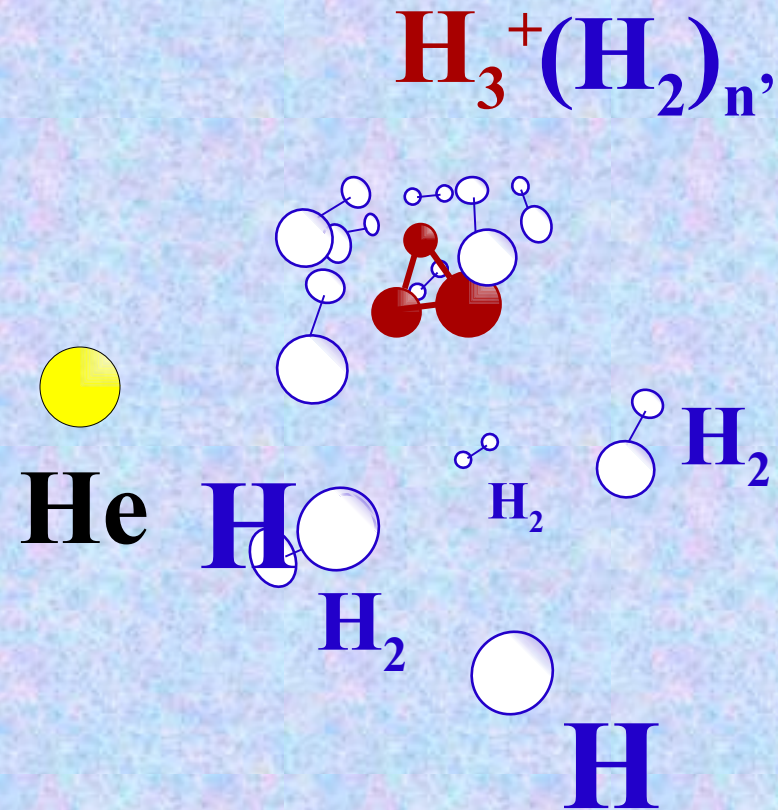
Fragmentation of hydrogen clusters



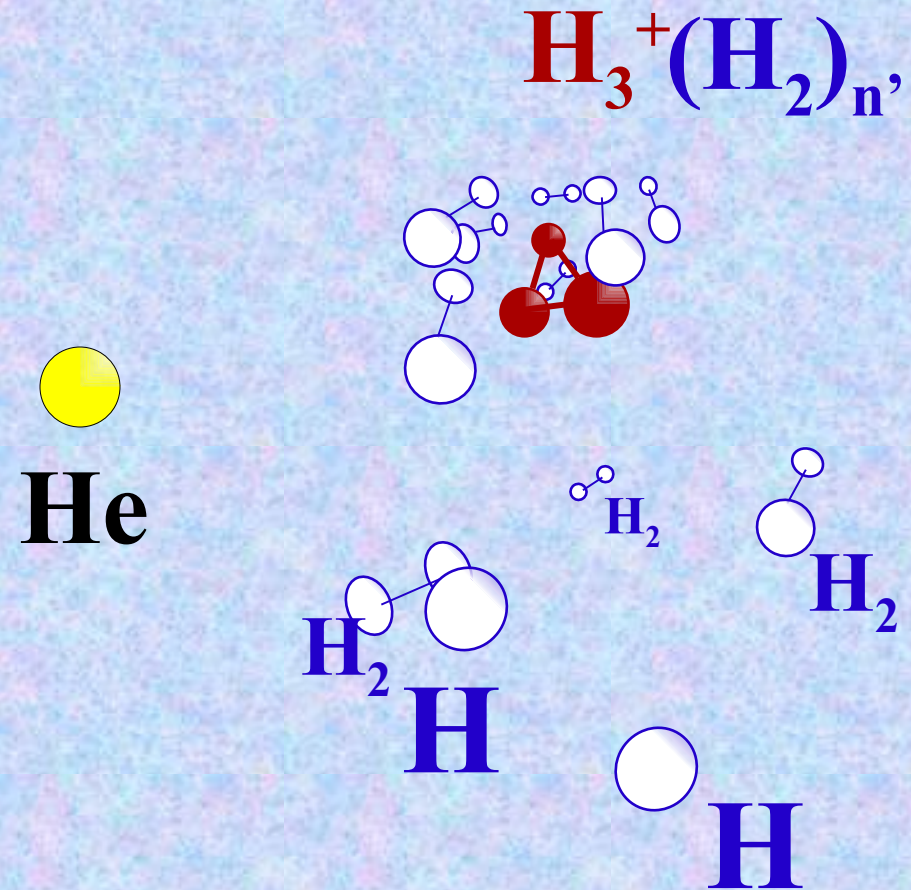
Fragmentation of hydrogen clusters



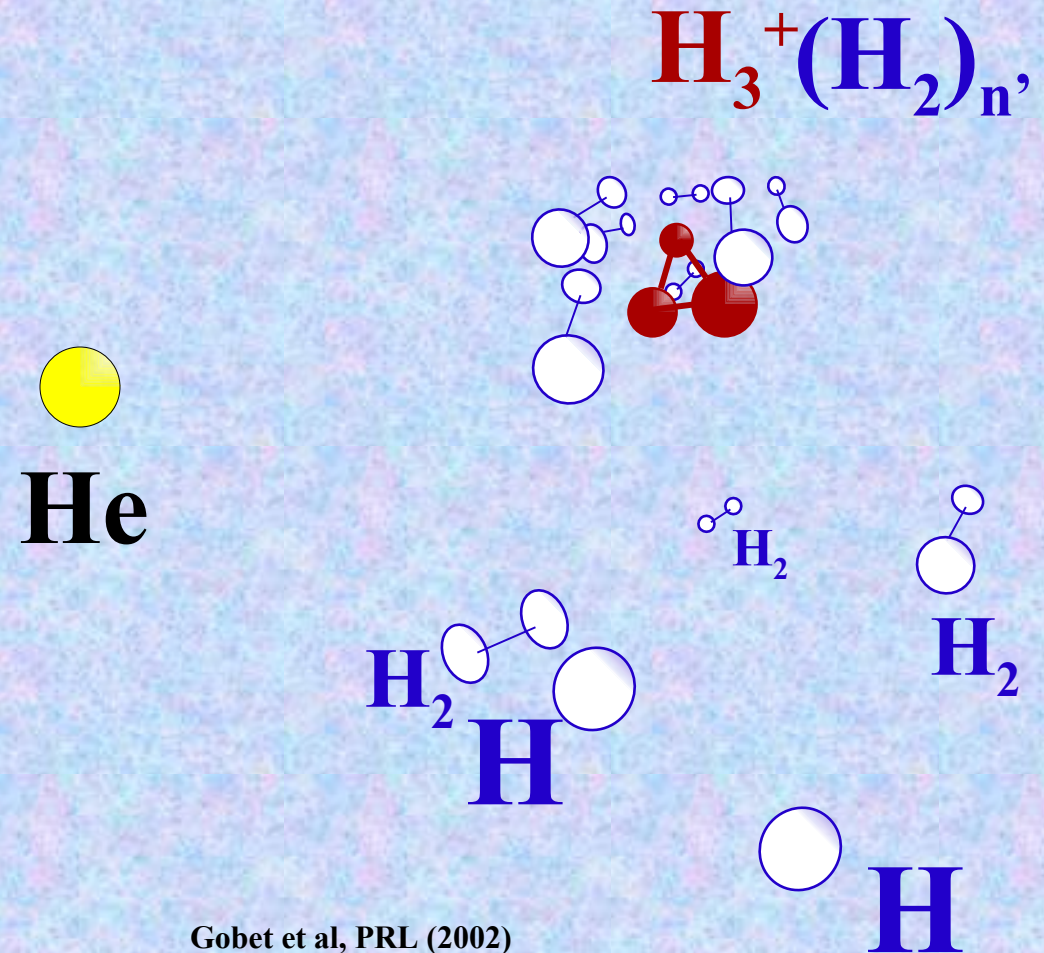
Fragmentation of hydrogen clusters



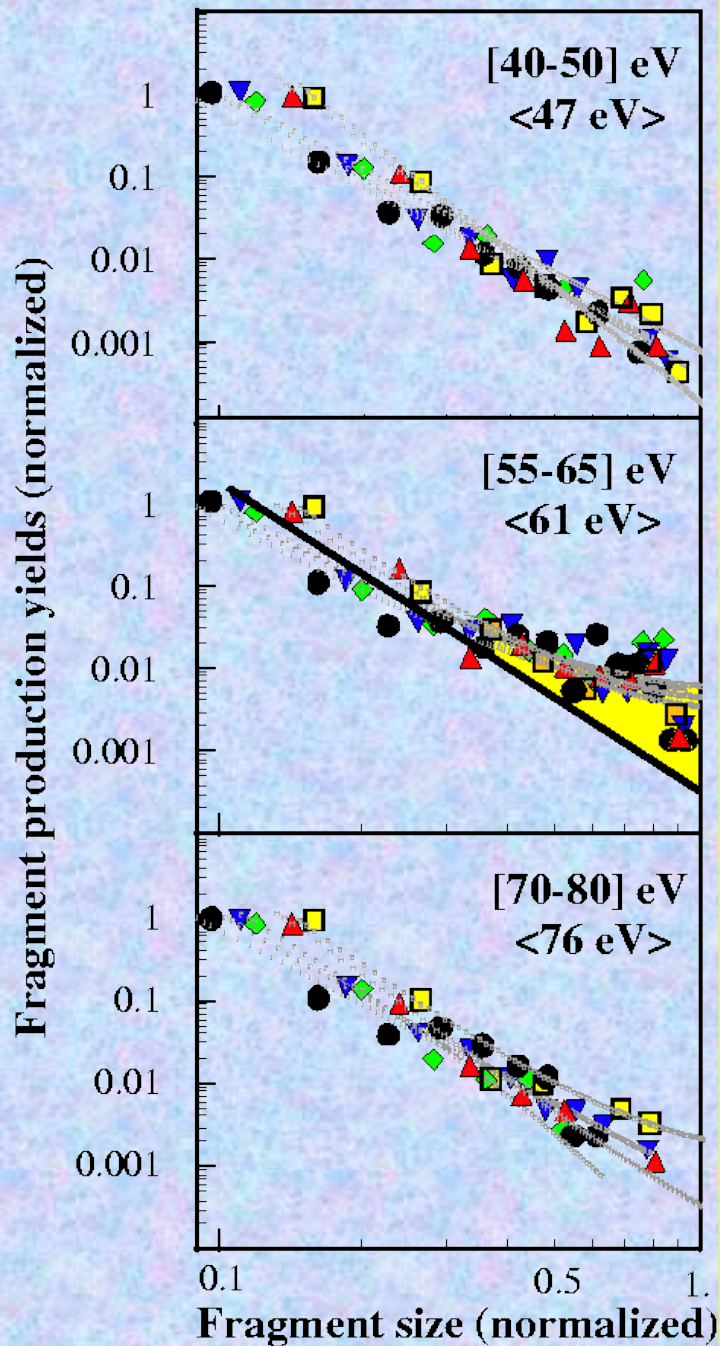
Fragmentation of hydrogen clusters



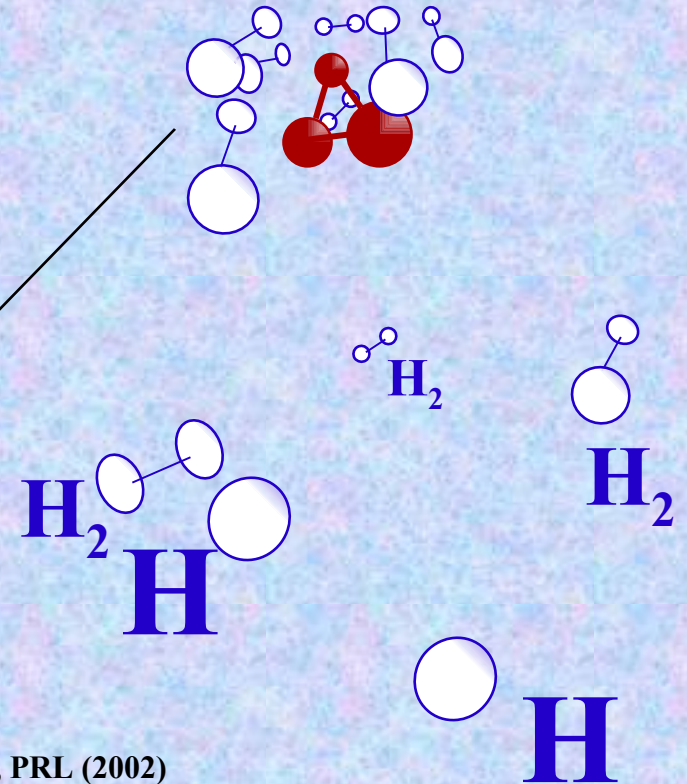
Fragmentation of hydrogen clusters



Fragmentation of hydrogen clusters



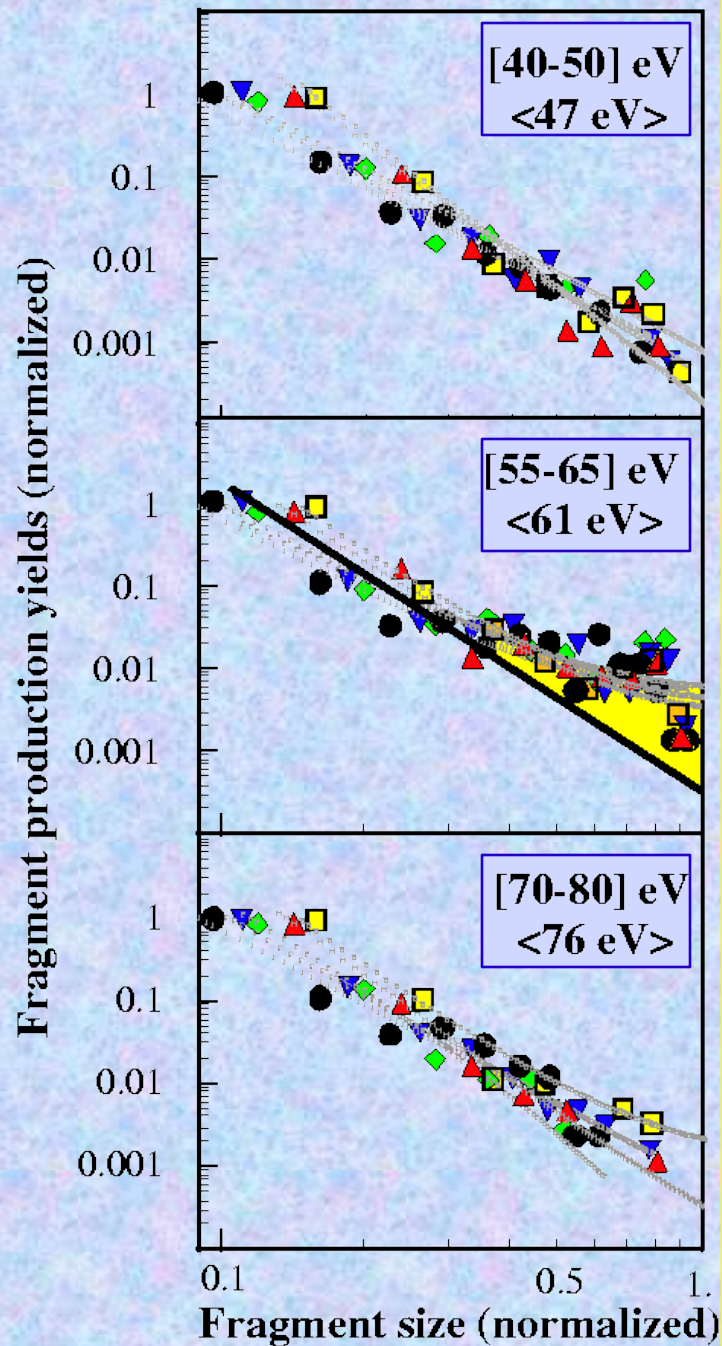
He



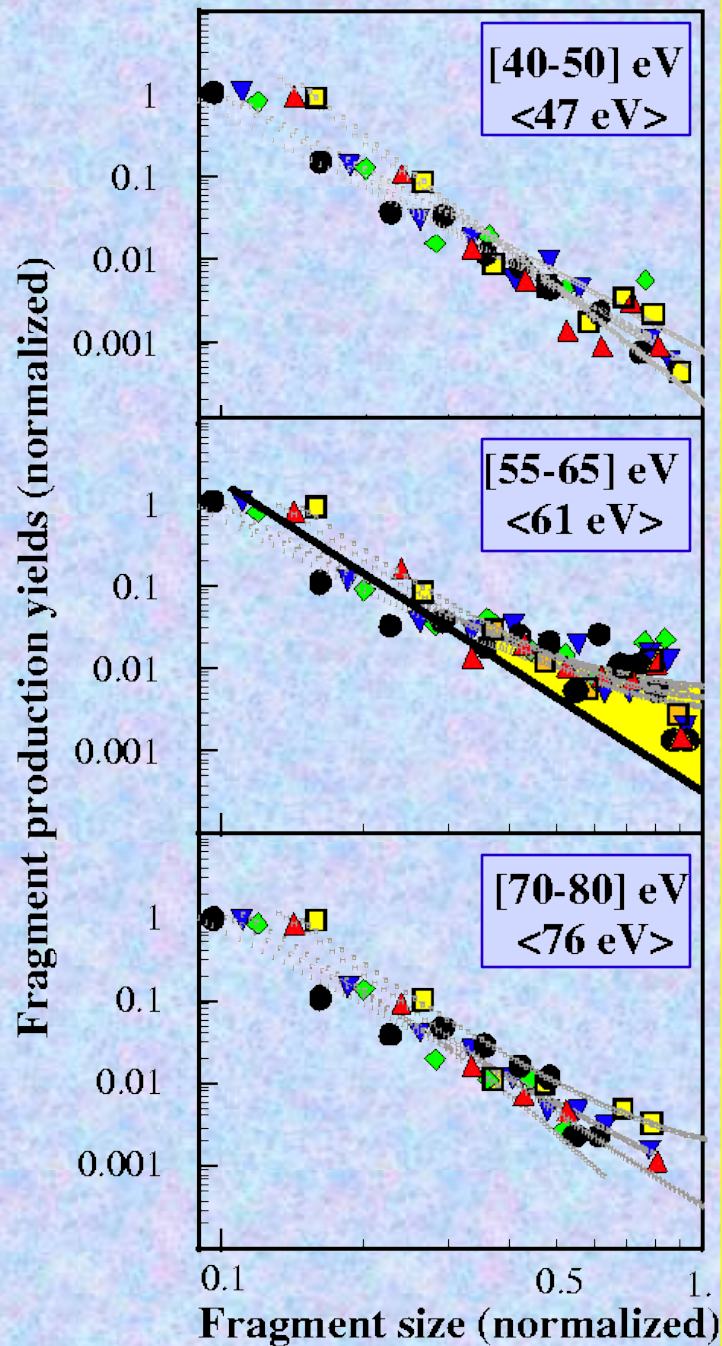
Gobet et al, PRL (2002)

Fragmentation of hydrogen clusters

■ Partition => energy

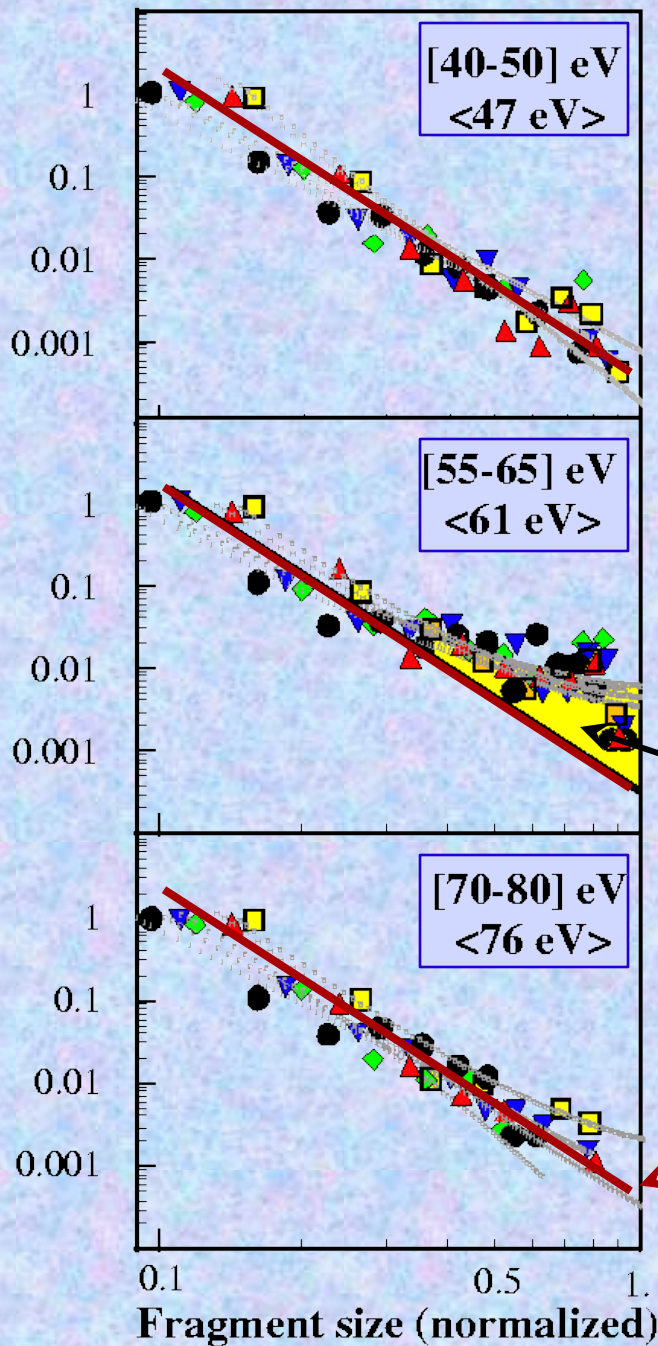


Fragmentation of hydrogen clusters



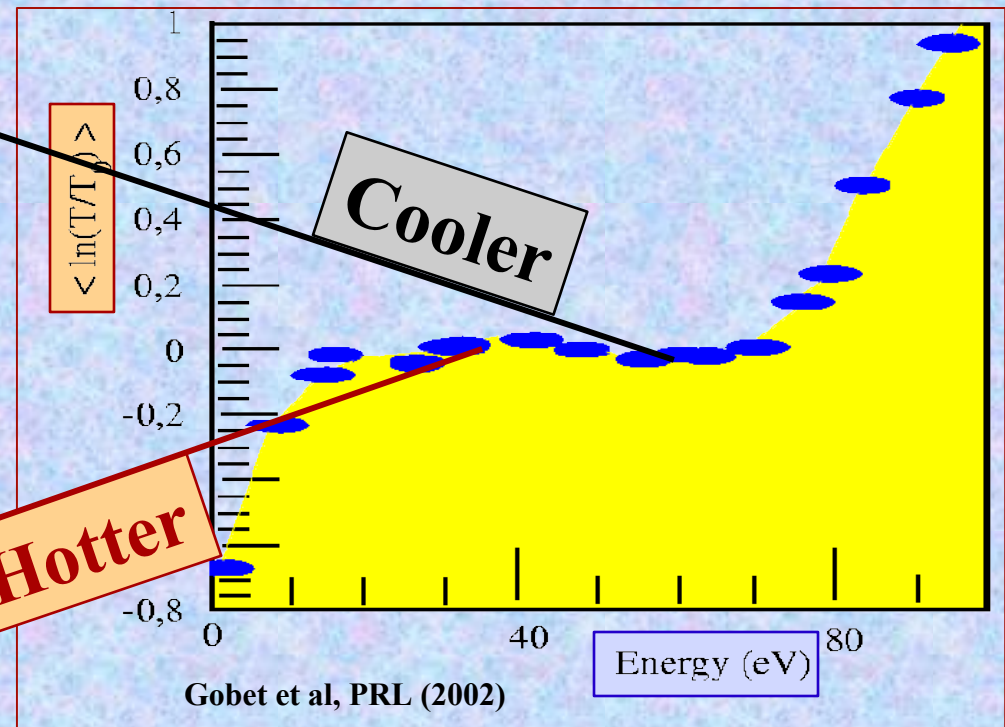
Fragmentation of hydrogen clusters

Fragment production yields (normalized)



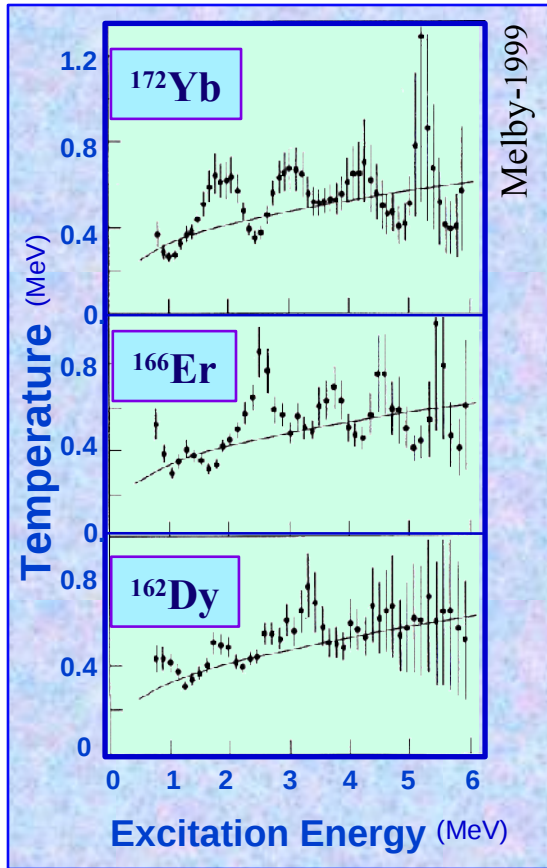
■ Partition => energy

■ Largest => Temperature



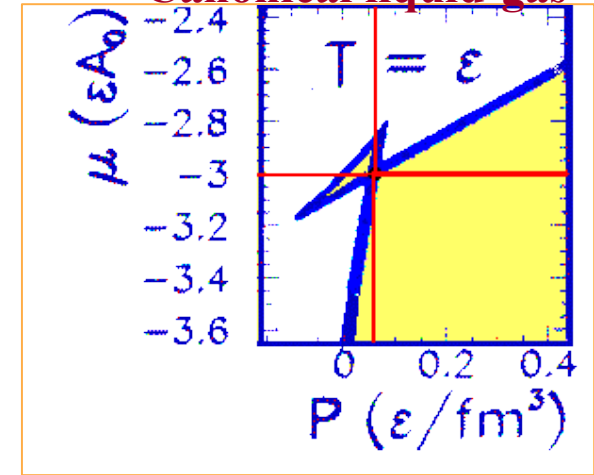
Abnormal curvatures

Nuclear superfluidity



Convex entropy

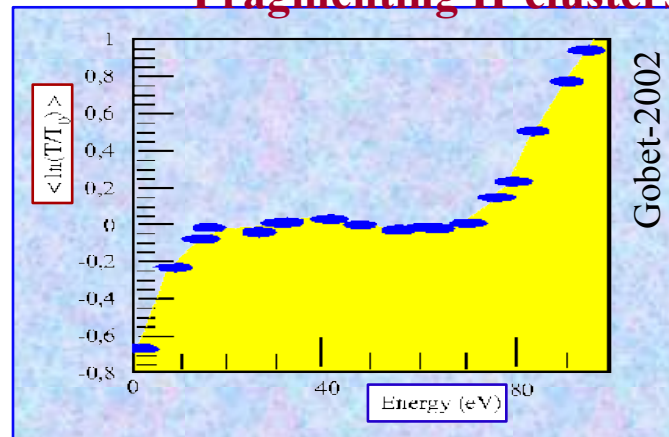
Canonical liquid-gas



Compressibility < 0
Susceptibility < 0

F. Gulminelli, Ph. Ch. PRL(1998)

Fragmenting H-clusters



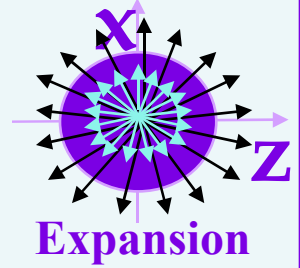
T decreasing with E

-Appendix - I -

Phase transition under flows

- Exact theory for radial flow
 - ✦ Isobar ensemble required
 - ✦ Phase transition not affected
- Example of oriented flow (transparency)
 - ✦ Exact thermodynamics
 - ✦ Effects on partitions

Radial flow at “equilibrium”



- **Equilibrium = Max S under constrains**
- **Additional constrains: radial flow $\langle p_r(r) \rangle$**

◆ **Additional Lagrange multiplier $\gamma(r)$**

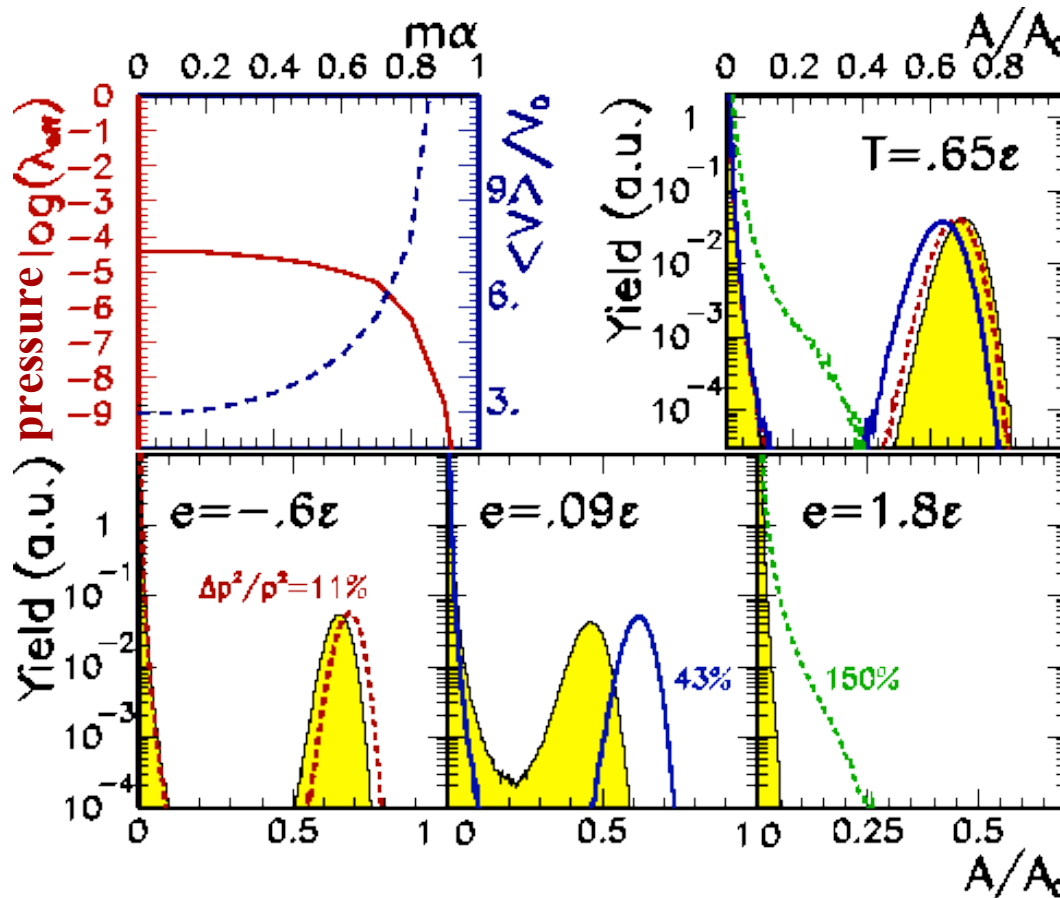
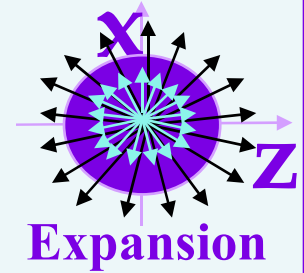
◆ **Self similar expansion $\langle p_r(r) \rangle = m \propto r \Rightarrow \gamma(r) = \beta \propto r$**

Thermal distribution in the
moving frame

Negative pressure
Isobar ensemble

- **Does not affect thermo is flow subtracted**

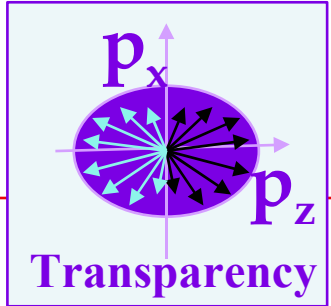
Radial flow at “equilibrium”



- ◆ Does not affect r partitioning
- ◆ Only reduces the pressure
- ◆ Shifts p distribution
- ◆ Changes fragment distribution:
fragmentation less effective

■ Does not affect thermo is flow subtracted

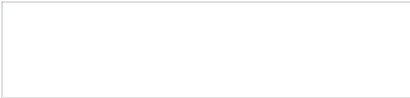
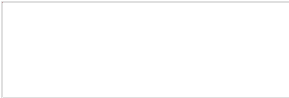
Transparency at “equilibrium”



- Equilibrium = Max S under constrains
- Additional constrains: memory flow $\langle \tau p_z \rangle$ $\begin{cases} \tau = -1 \text{ target} \\ \tau = 1 \text{ projectile} \end{cases}$

- ◆ Additional Lagrange multiplier α



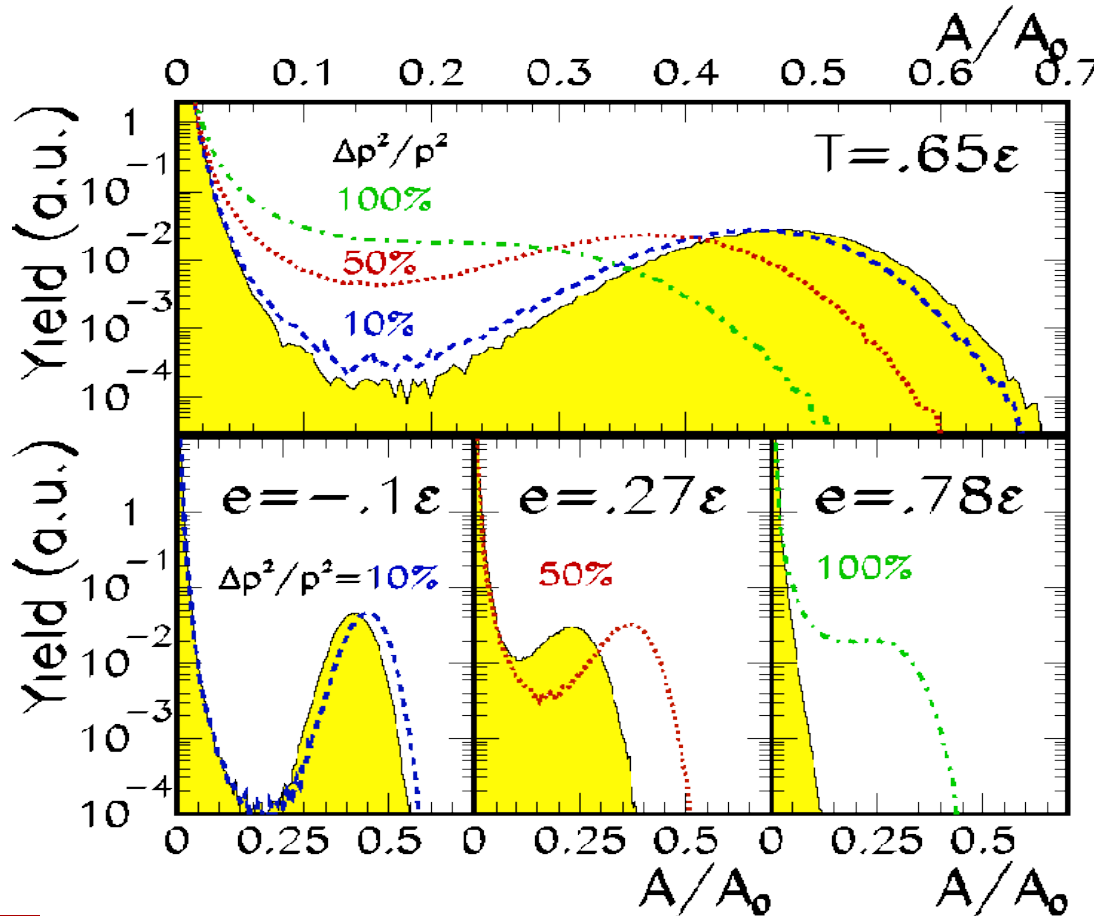
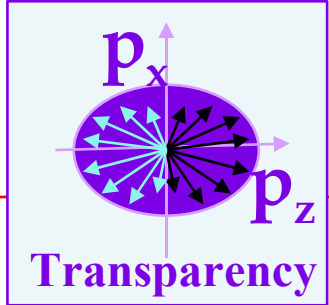
- ◆ EOS  leads to $\langle \tau p_z \rangle = A p_0$ with 



Thermal distribution
in moving frame

- Affects p space and fragments,
- Does not affect the thermo if E_{flow} subtracted

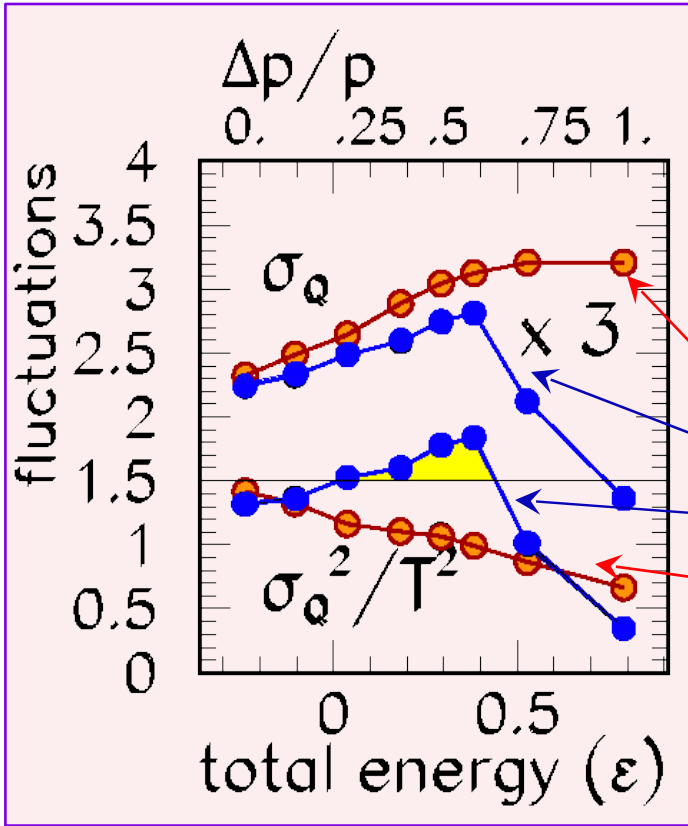
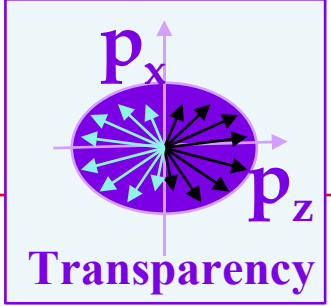
Transparency at “equilibrium”



- ◆ Does not affect r partitioning
- ◆ Shifts p distribution
- ◆ Changes fragment distribution: fragmentation less effective

■ Affects p space and fragments

Transparency at “equilibrium”



- Fragmentation less effective
- If E_{flow} not subtracted
 - ◆ Affect $\langle E \rangle$
 - ◆ Affect T from $\langle E_k \rangle$
 - ◆ Ex: Fluctuations of Q values
 - ◆ All thermal (from $E = -2\varepsilon$)
 - ◆ All relative motion
- Up to 10-15% no effect

- **Affects p space and fragments**

- Appendix -II-

Thermo with Coulomb

- Coulomb reduces coexistence and $C < 0$ region
- Statistical treatment of Coulomb
 - ♦ (E_N, V_C) a common phase diagram for charged and uncharged systems

Coulomb reduces L-G transition

■ Reduces coexistence

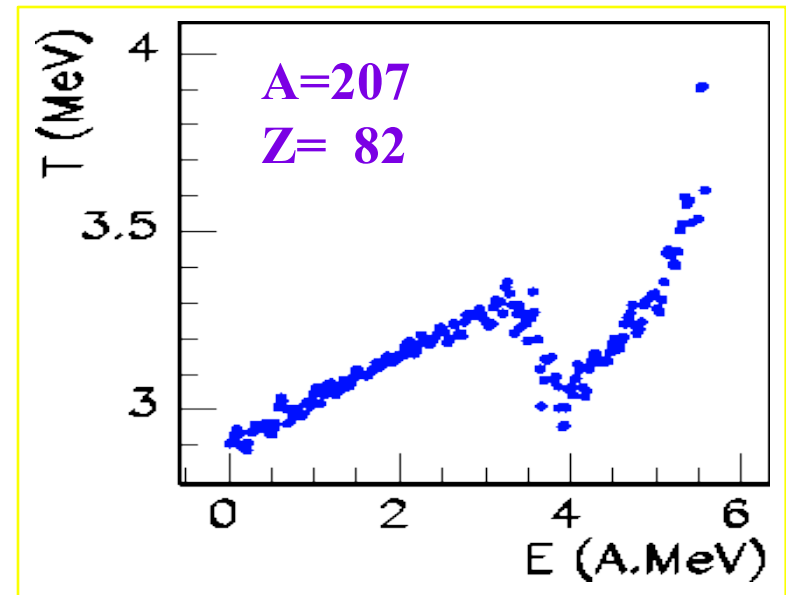
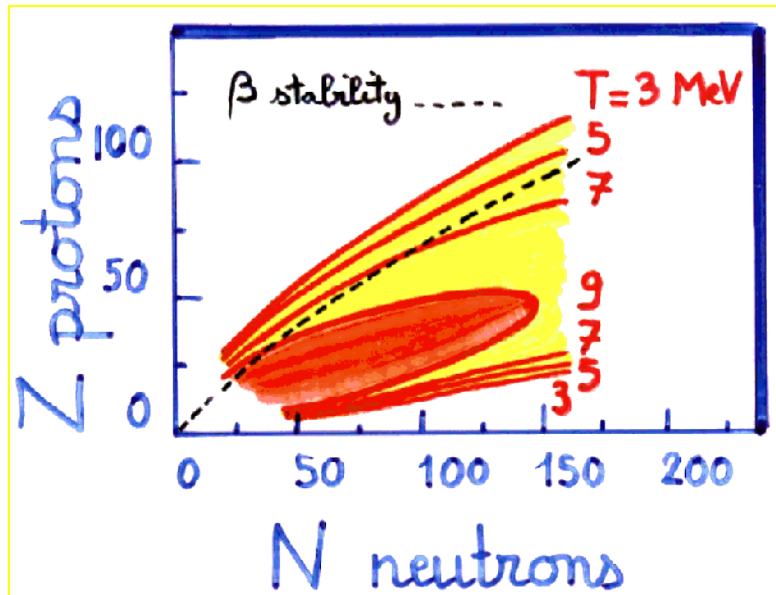
◆ Bonche-Levit-Vautherin

Nucl. Phys. A427 (1984) 278

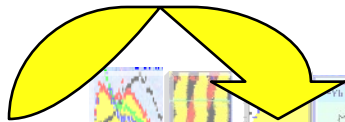
■ Reduces $C < 0$

◆ Lattice-Gas

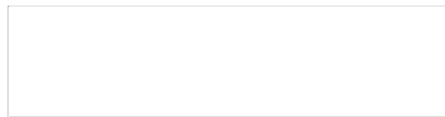
◆ OK up to heavy nuclei



Gulminelli, PC, Comment to PRC66 (2002) 041601



Statistical treatment of Non saturating forces interactions



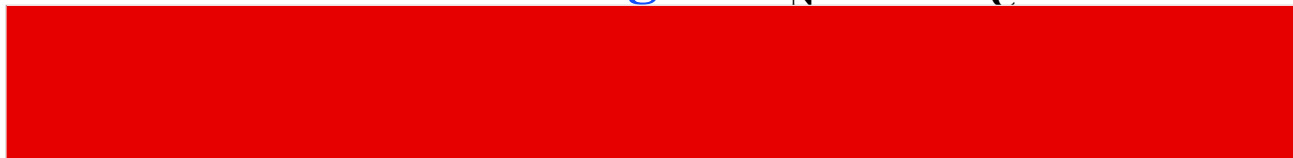
- Effective charge $q = 1$ charged system -
- Effective charge $q = 0$ uncharged system -



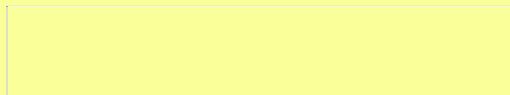
A unique framework: $e^{-\beta E} = e^{-\beta_N E_N} e^{-\beta_C V_C}$

♦ Introduce two temperatures $\beta_N = \beta$ and $\beta_C = q^2 \beta$

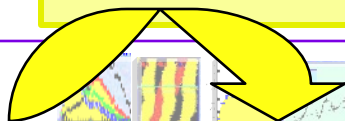
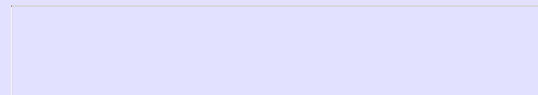
♦ $O \square \square \square \square \square$ two energies E_N and V_C



♦ $\beta_C = 0$ Uncharged

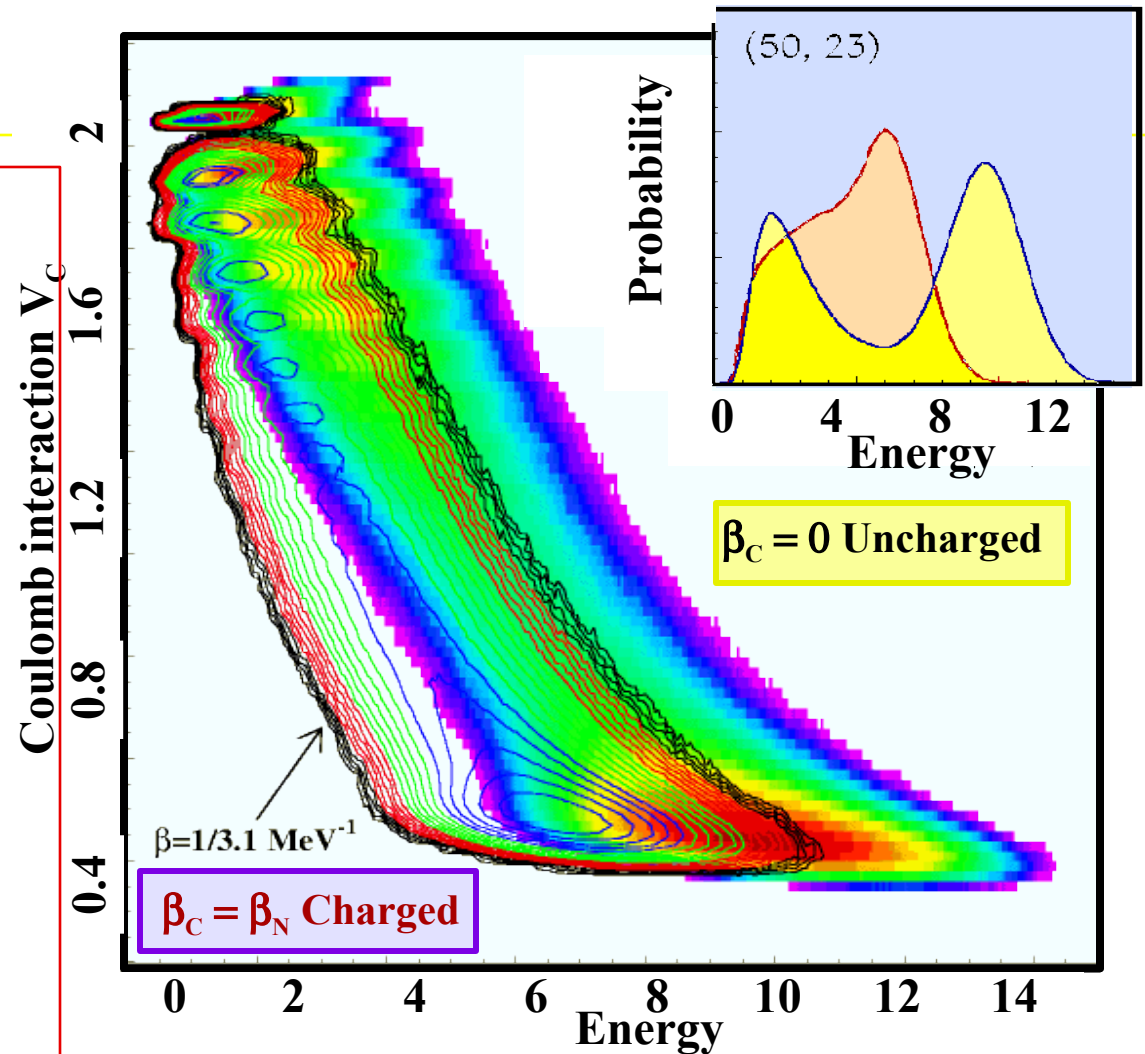


$\beta_C = \beta_N$ Charged

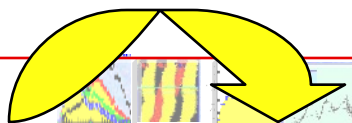


With and without Coulomb a unique $S(E_N, V_C)$

- (E_N, V_C) a unique phase diagram
- Coulomb reduces E bimodality
 - ◆ different weight
 - ◆ rotation of E axis
 - $E = E_N + V_C$, charged
 - $E = E_N$, uncharged



Gulminelli, PC, Raduta, Raduta, submitted to PRL



1

2

3

4

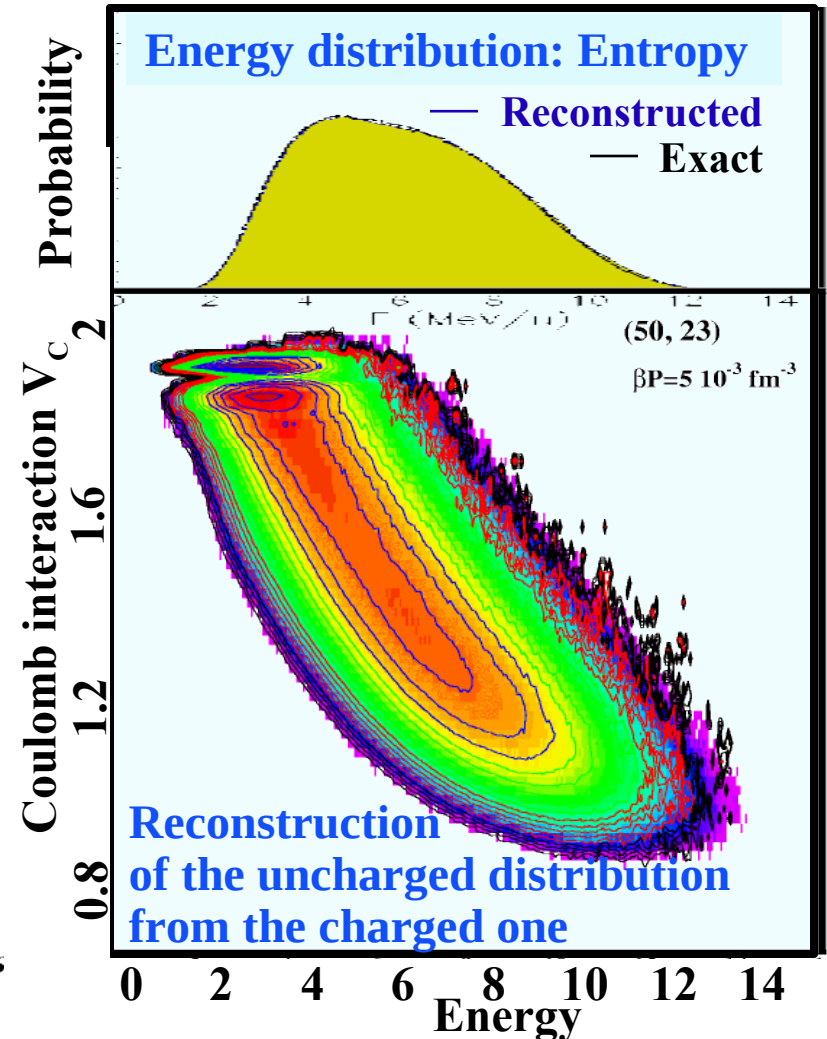
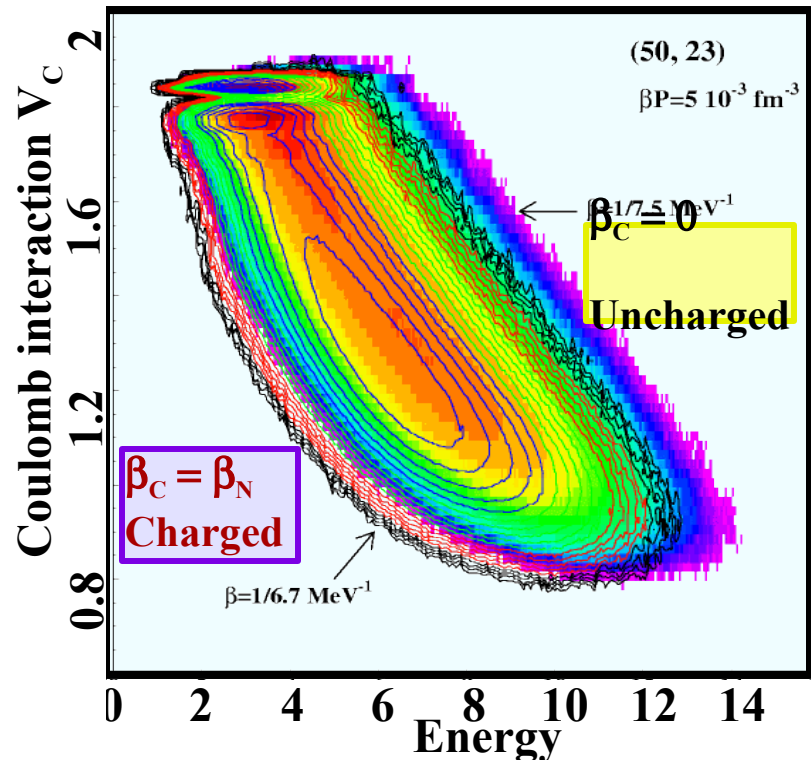
C



Play

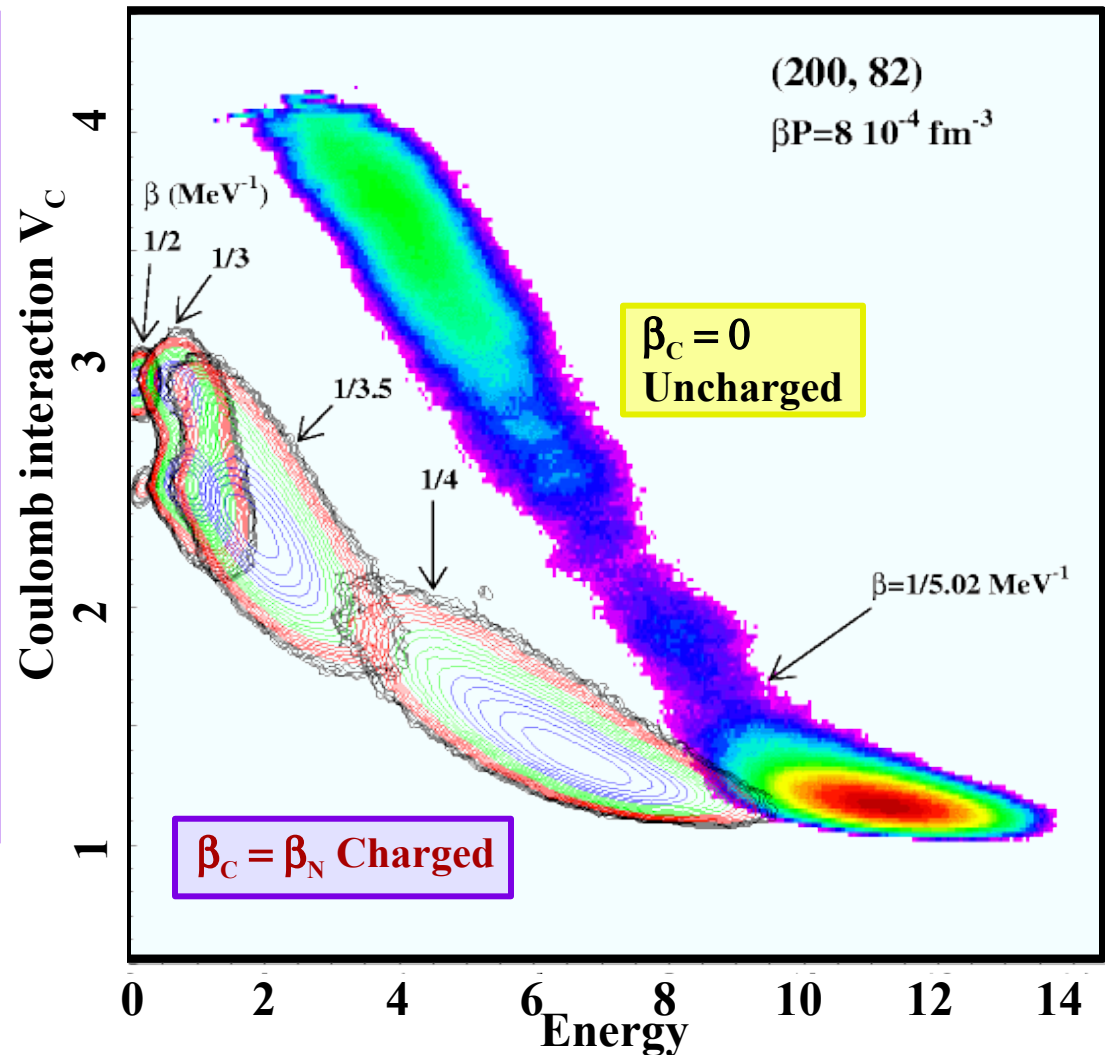
Correction of Coulomb effects

■ If events overlap



Partitions may differ for heavy nuclei

- Channels are different (cf fission)
- Re-weighting impossible
- However, a unique phase diagram (E_N V_C)



Gulminelli, PC, Raduta, Raduta, submitted to PRL

-Appendix - III -

Equilibrium in finite systems

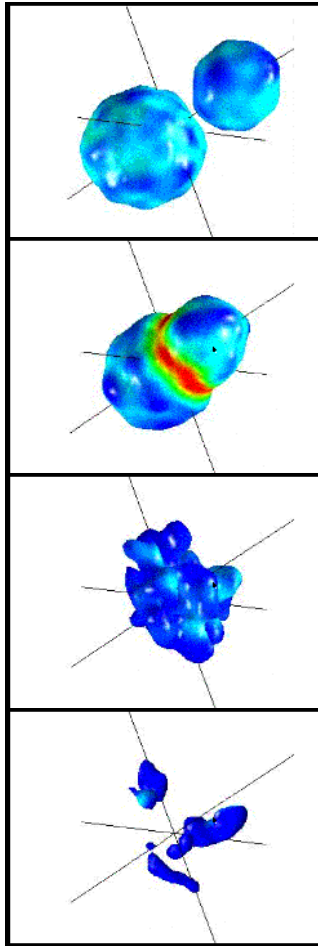
■ Macroscopic

- ♦ One realization (event) at equilibrium

■ Microscopic

- ♦ Statistical ensemble and information
- ♦ Ergodicity versus mixing dynamics
- ♦ Multiplicity of equilibria
- ♦ What is temperature?

Equilibrium : What Does It Mean?



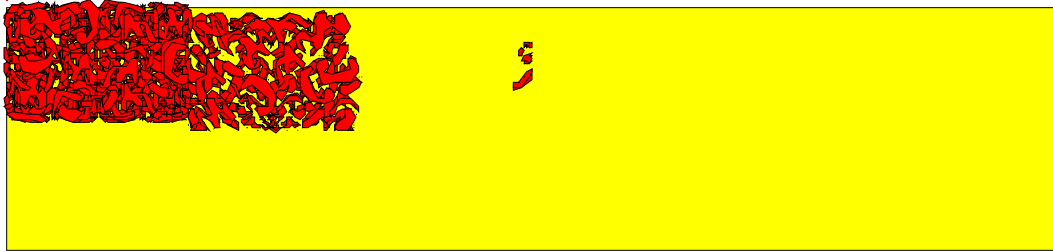
■ Finite systems

- ◆ In time
- ◆ In space

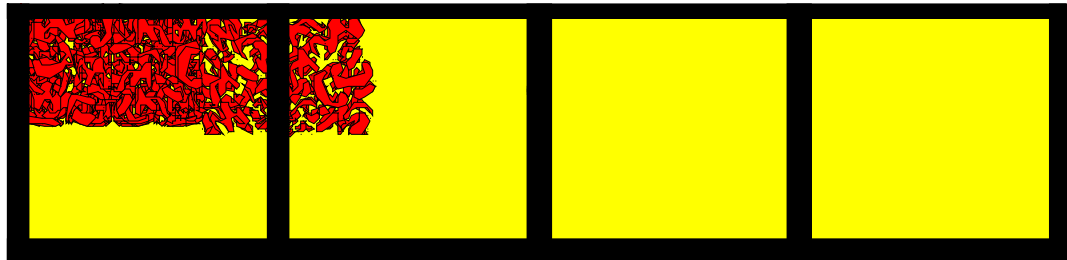
- Isolated open systems

- ◆ No reservoir or bath
- ◆ No container

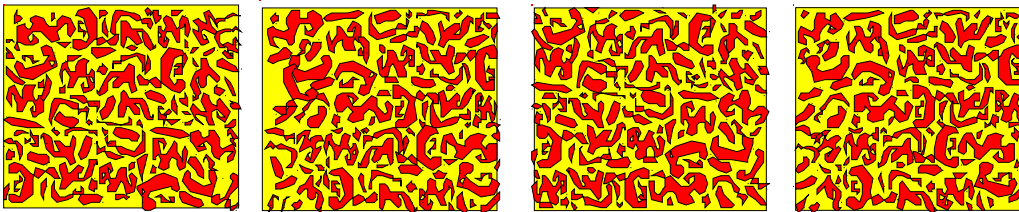
Equilibrium : What Does It Mean?



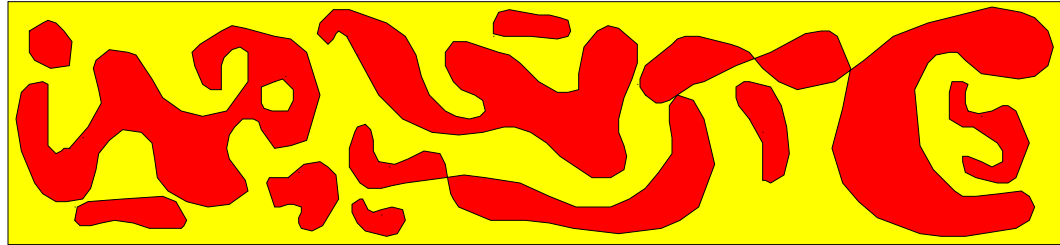
- Thermodynamics : infinite system



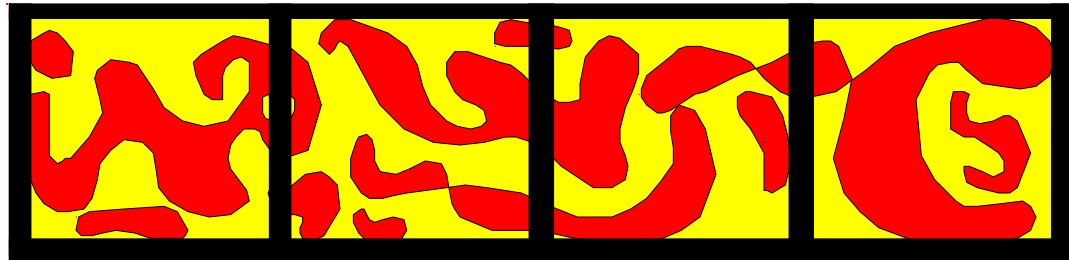
- One ∞ system = ensemble of ∞ sub-systems



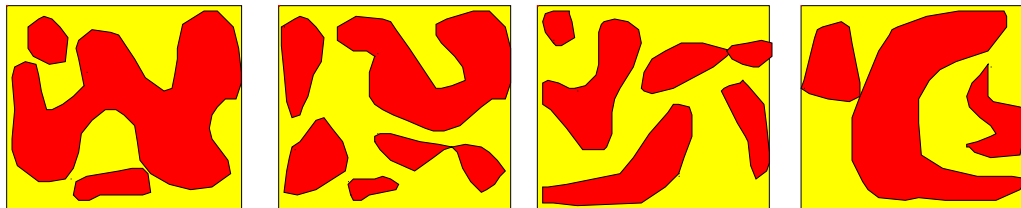
Equilibrium : What Does It Mean?



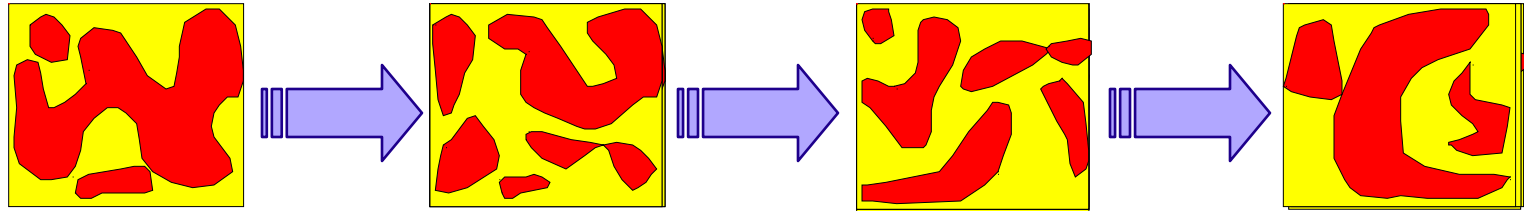
Finite system



Cannot be cut in sub-systems

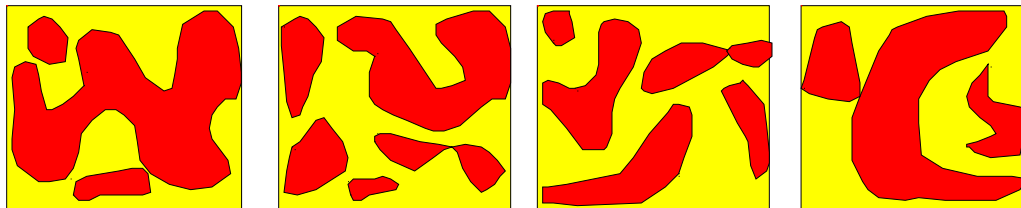


Equilibrium : What Does It Mean?

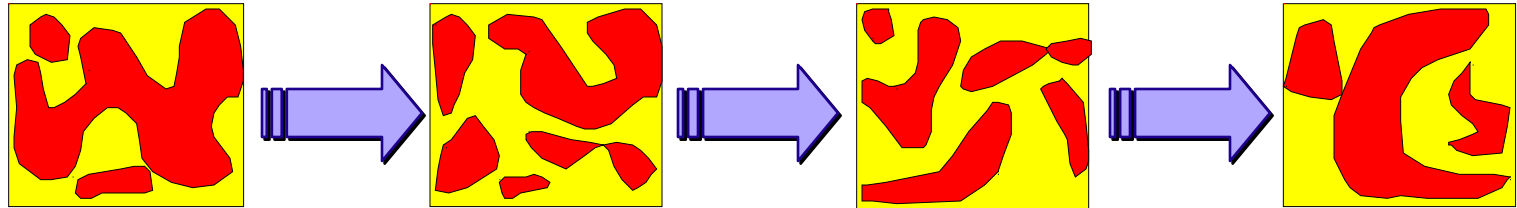


- One small system in time if it is ergodic
Is a statistical ensemble

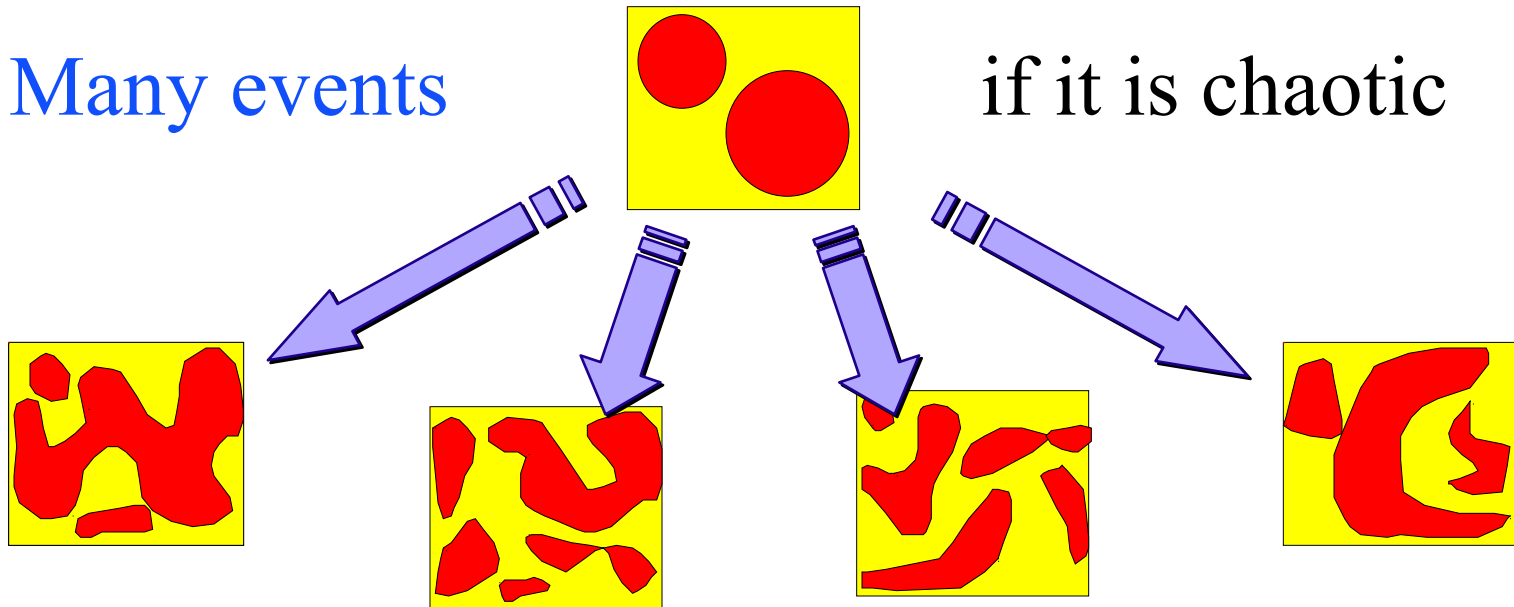
- Cannot be cut in sub-systems



Equilibrium : What Does It Mean?

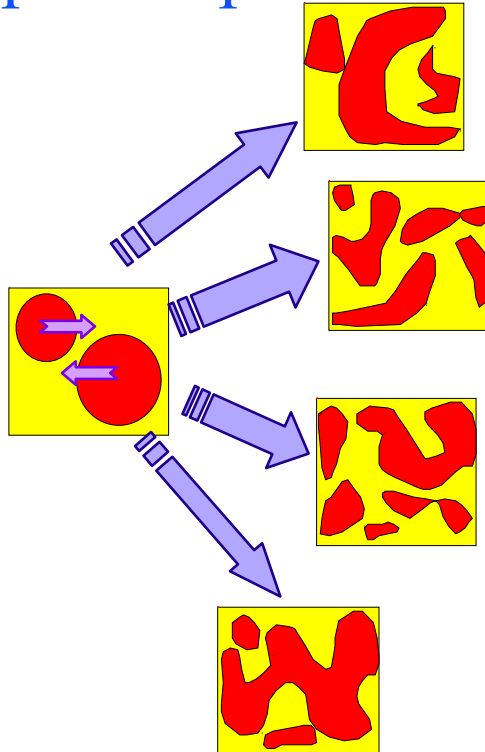


- One small system in time if it is ergodic
- Many events if it is chaotic



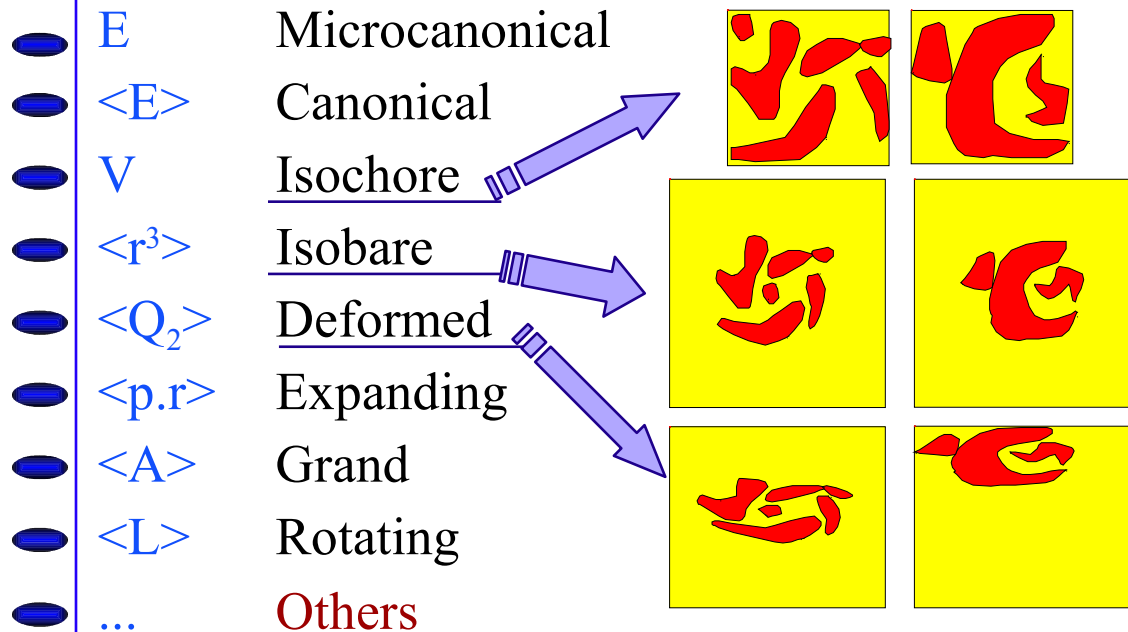
Statistics from minimum information

- “Chaos” populates phase space



- Dynamics: global variables
- Statistics: population

♦ Many different ensembles



What is temperature ?



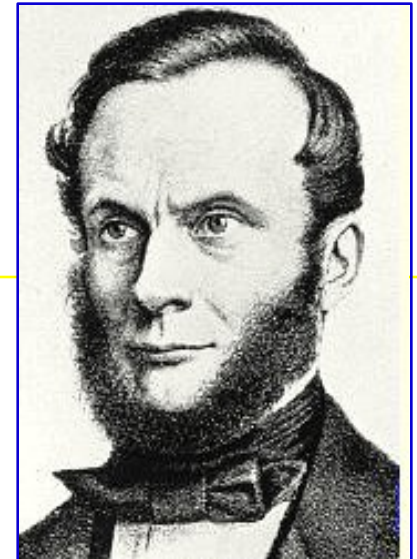
L. Boltzmann

■ W equiprobable microstates



Entropy = disorder (-information)

■ T is the entropy increase



R. Clausius

What is temperature ?

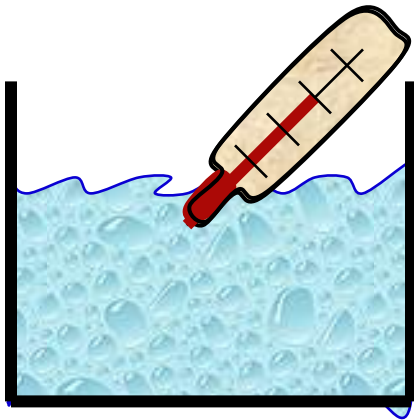
■ The microcanonical temperature

$$S = k \log W$$

$$\delta S = T^{-1} \delta E$$

What is temperature ?

- **It is what thermometers measure.**



$$\mathbf{E} = \mathbf{E}_{\text{thermometer}} + \mathbf{E}_{\text{system}}$$

- ## ■ The microcanonical temperature

$$S = k \log W$$

$$\delta S = T^{-1} \delta E$$

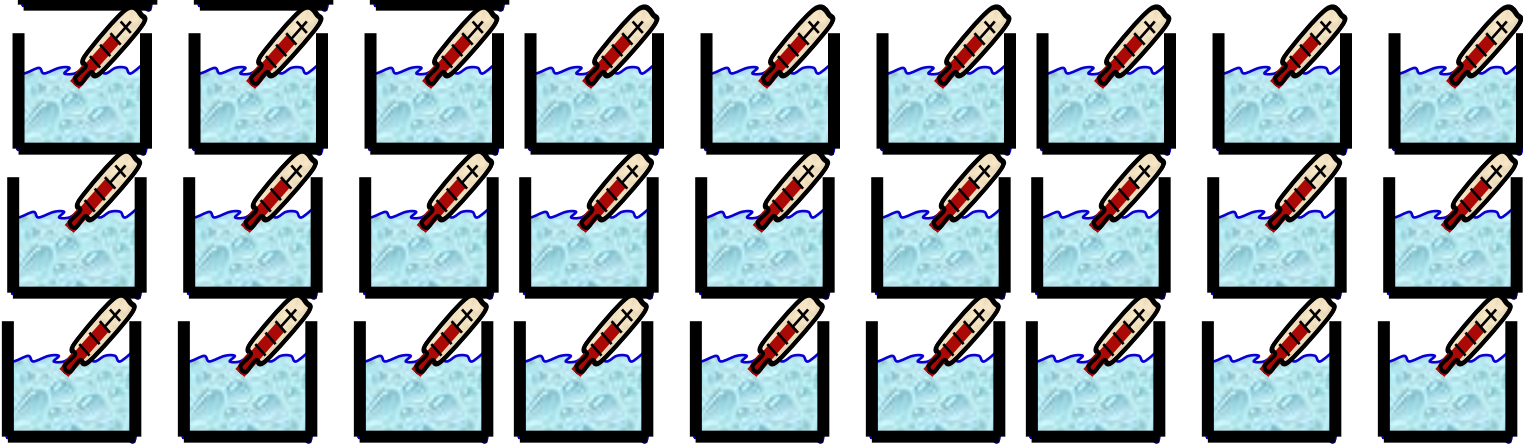
What is temperature ?

- It is what thermometers measure.



$$E = E_{\text{thermometer}} + E_{\text{system}}$$

Distribution of microstates



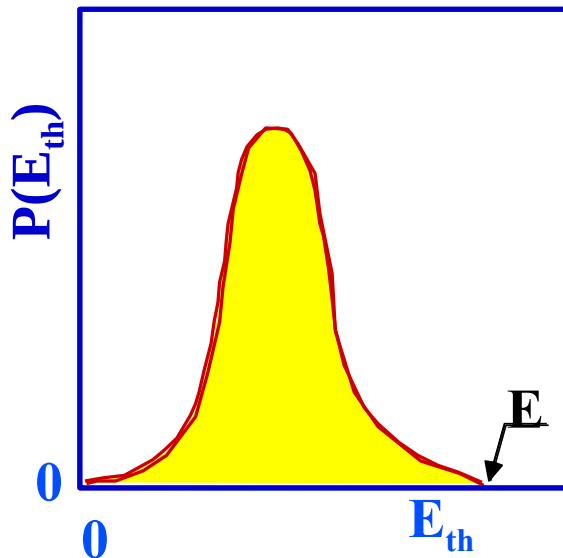
- The microcanonical temperature

$$S = k \log W$$

$$\delta S = T^{-1} \delta E$$

What is temperature ?

- It is what thermometers measure.



$$E = E_{th} + E_{sys}$$

Distribution of microstates

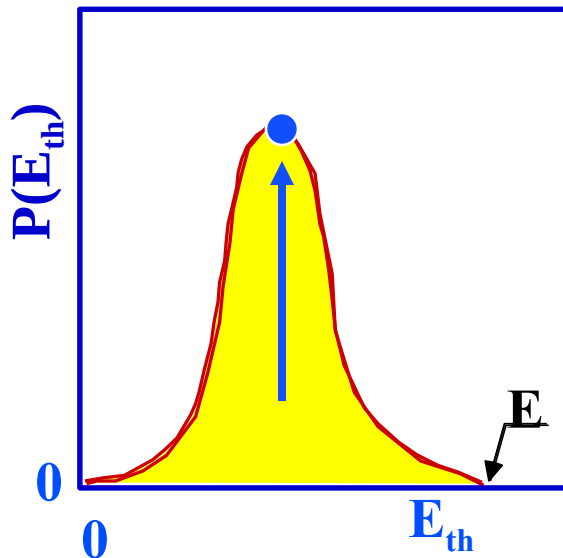
- The microcanonical temperature

$$S = k \log W$$

$$\delta S = T^{-1} \delta E$$

What is temperature ?

- It is what thermometers measure.



$$E = E_{th} + E_{sys}$$

Equiprobable microstates

$$P(E_{th}) \propto W_{th}(E_{th}) * W_{sys}(E - E_{th})$$
$$\max P \Rightarrow \delta \log W_{th} - \delta \log W_{sys} = 0$$

$$\max P \Rightarrow \delta S_{th} - \delta S_{sys} = 0$$

Most probable partition: $T_{th} = T_{sys}$

- The microcanonical temperature

$$S = k \log W$$

$$\delta S = T^{-1} \delta E$$

-Appendix -IV-

$C < 0$ in Liquid gas transition

- Negative heat capacity in fluctuating volume ensemble
 - ◆ Difference between C_p and C_v
- Difference between C and $E(T)$
 - ◆ Role of the thermo transformation
- Negative heat capacity in statistical models and Channel opening

- a -

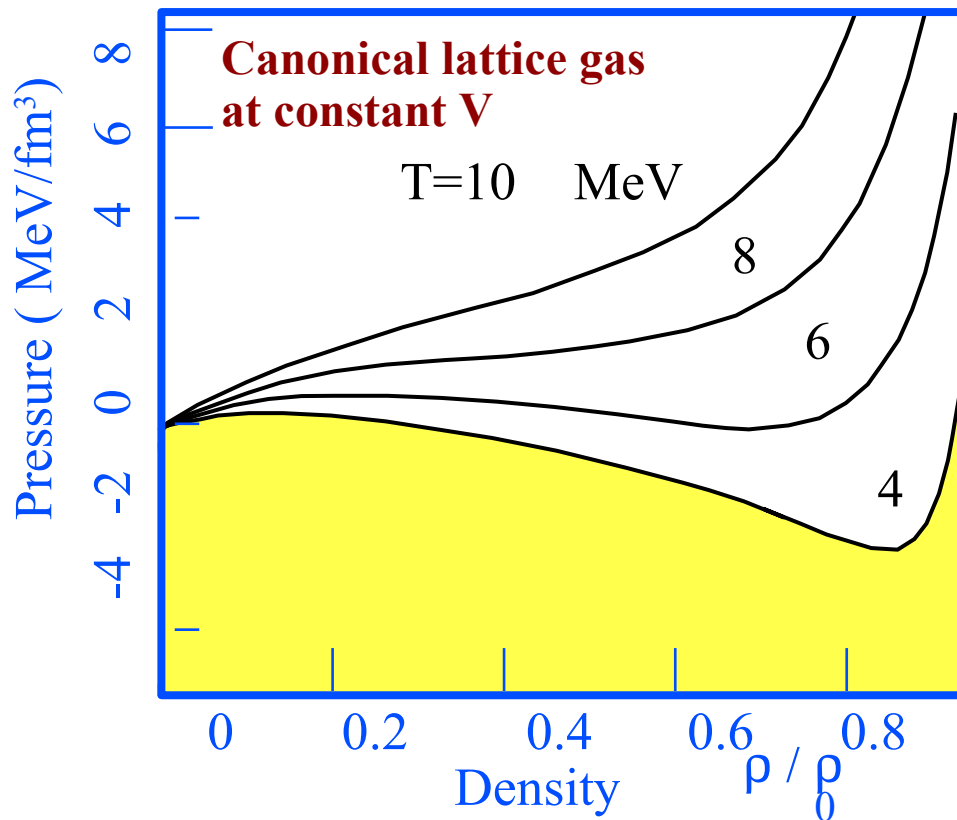
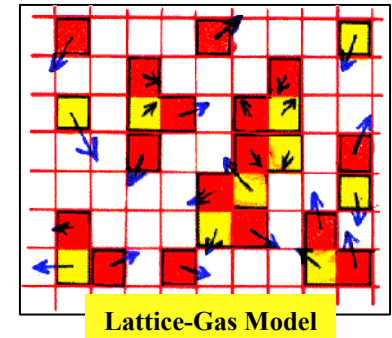
Role of volume fluctuations

Fixed volume $C_v > 0$ but $K < 0$

Open systems $C_p < 0$

Volume: order parameter

L-G in a box: $V=\text{cst}$

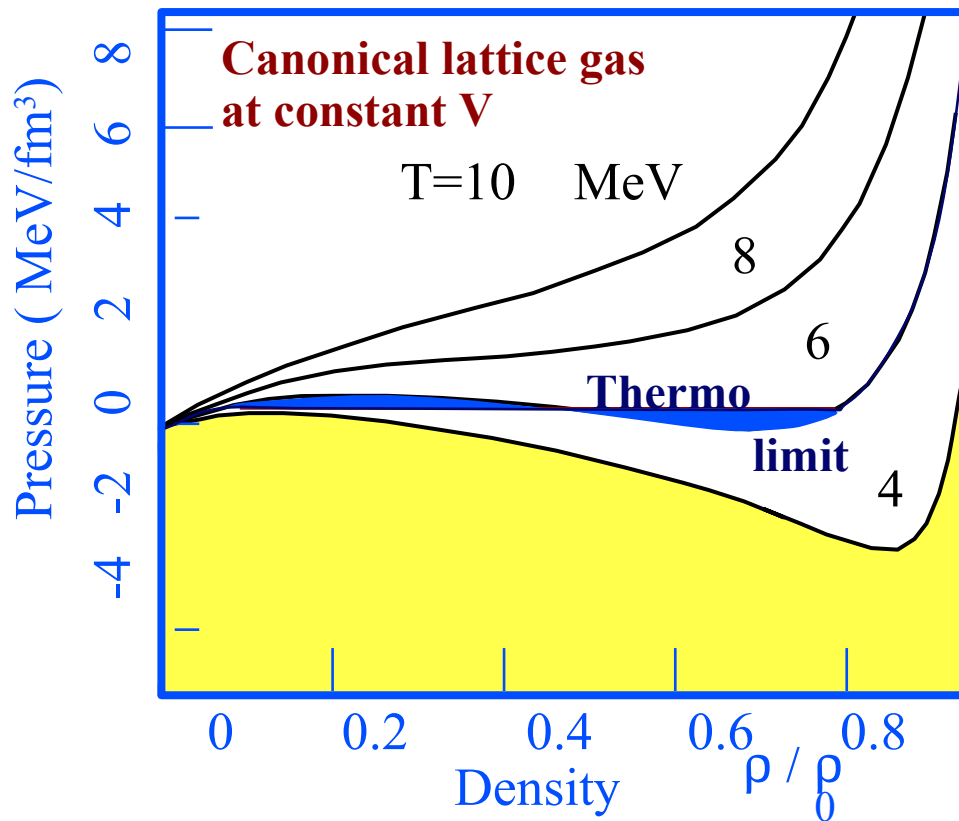
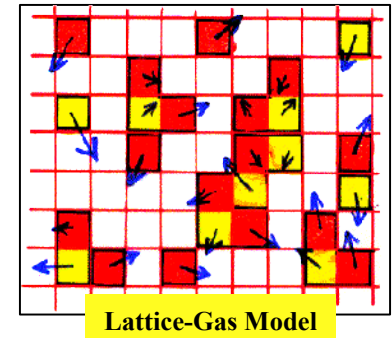


■ Negative compressibility

Gulminelli & PC PRL 82(1999)1402

Volume: order parameter

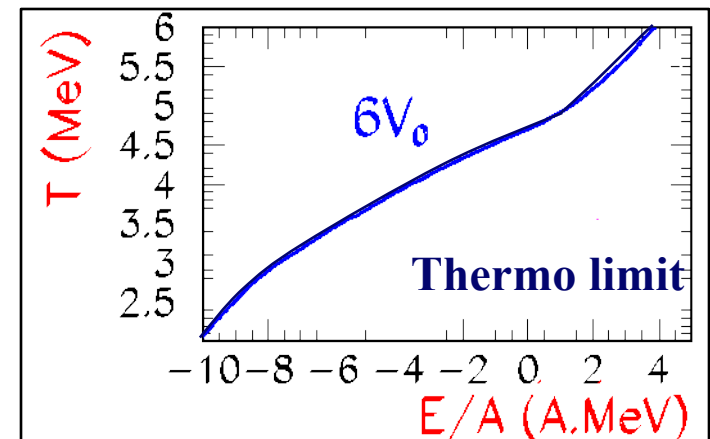
L-G in a box: $V=\text{cst}$



■ Negative compressibility

Gulminelli & PC PRL 82(1999)1402

■ $C_V > 0$



See specific discussion for large systems

Pleimling and Hueller, J.Stat.Phys.104 (2001) 971

Binder, Physica A 319 (2002) 99.

Gulminelli et al., cond-mat/0302177, Phys. Rev. E.

Open systems (no box)

Fluctuating volume

PC, Duflot & Gulminelli PRL 85(2000)3587



Isobar
ensemble

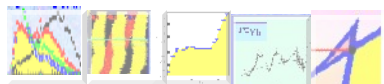
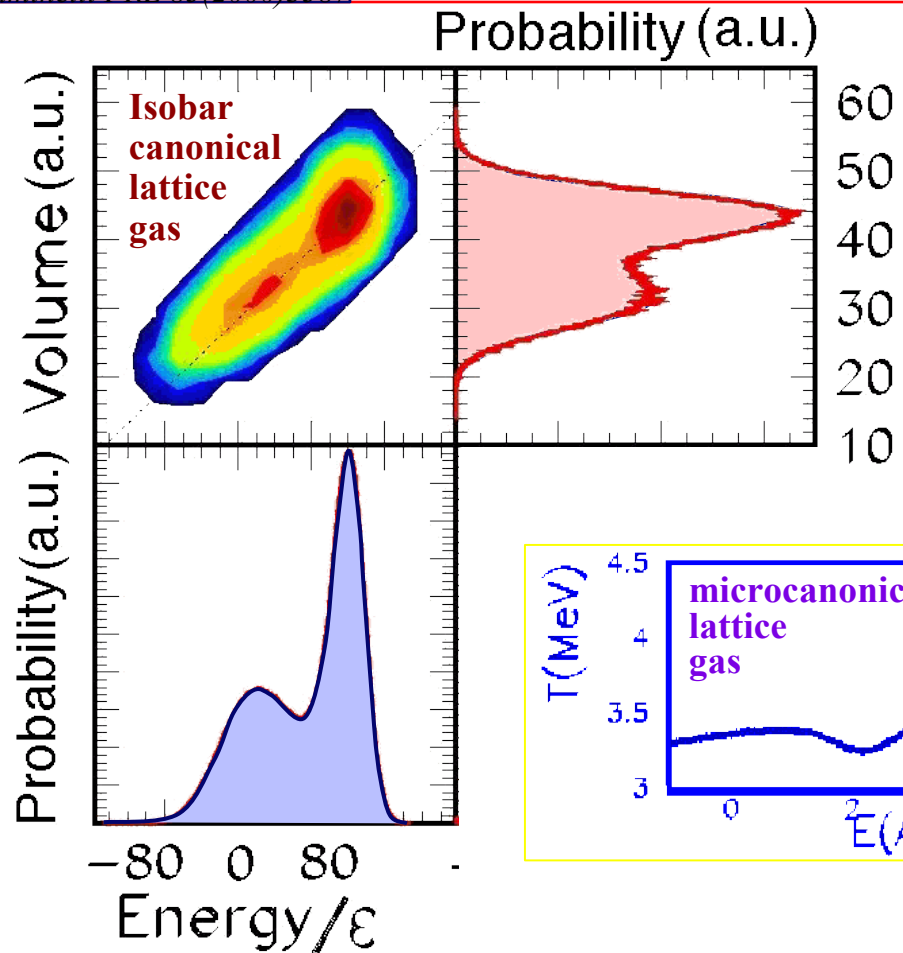
$$P^{(i)} \propto \exp(-\lambda V^{(i)})$$

■ Bimodal $P(V)$

◆ Negative compressibility

■ Bimodal $P(E)$

◆ Negative heat capacity $C_p < 0$



1

2

3

4

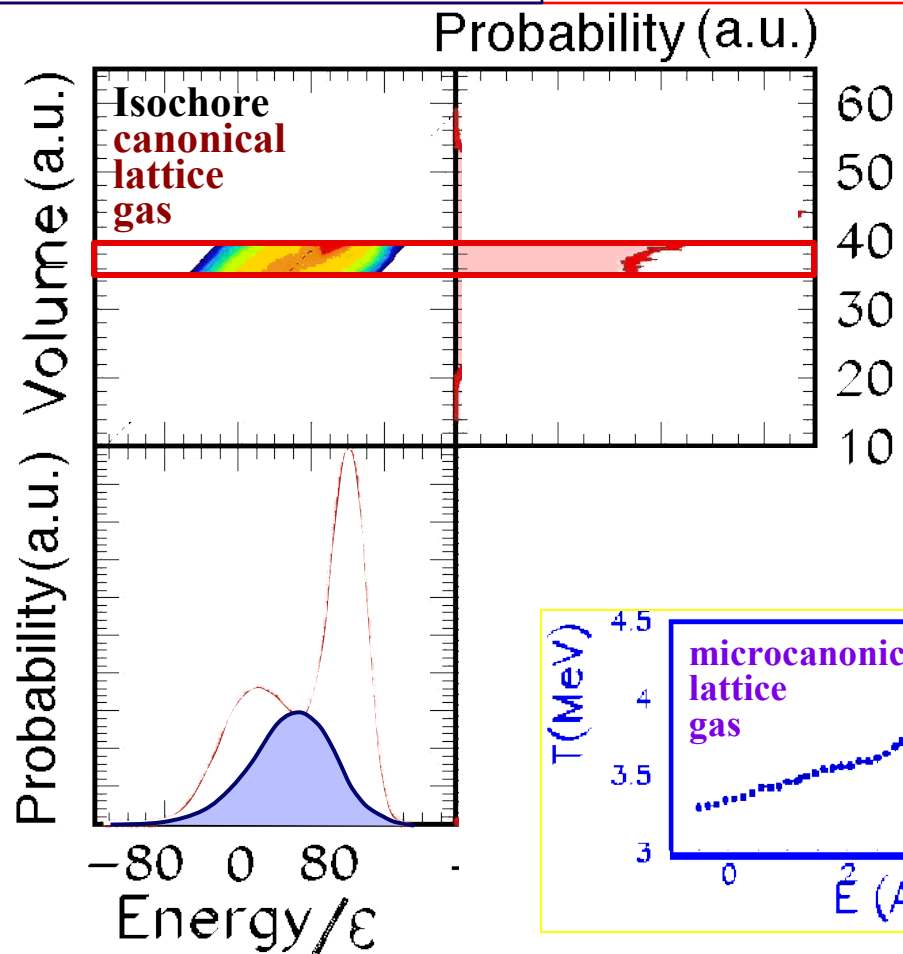
C



Constrain on V

i.e. on order parameter

- Suppresses bimodal $P(E)$
- ◆ No negative heat capacity $C_V > 0$

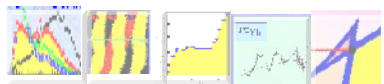


See specific discussion for large systems

Pleimling and Hueller, *J.Stat.Phys.* 104 (2001) 971

Binder, *Physica A* 319 (2002) 99.

Gulminelli et al., cond-mat/0302177, *Phys. Rev. E*.



1

2

3

4

C



Play

- b -

Pressure dependent EOS

Caloric Curves not EOS

Fluctuations measure EOS

Volume dependent $T(E)$

■ 2-D EOS

$E \longrightarrow T$

$V \longrightarrow P$

Temperature

Energy

Pressure



Volume dependent $T(E)$

■ 2-D EOS

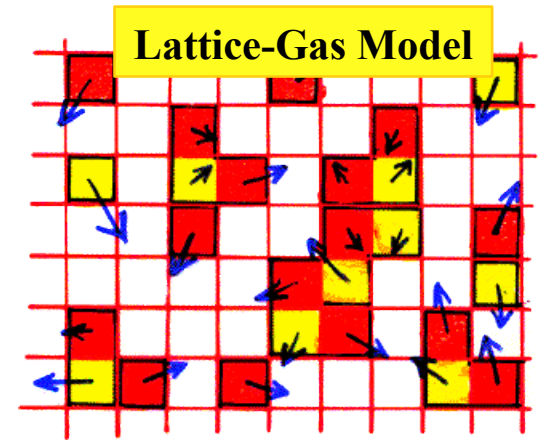
$E \longrightarrow T$

$V \longrightarrow P$

Temperature

Energy

Pressure



Volume dependent Energy $E(T, V)$

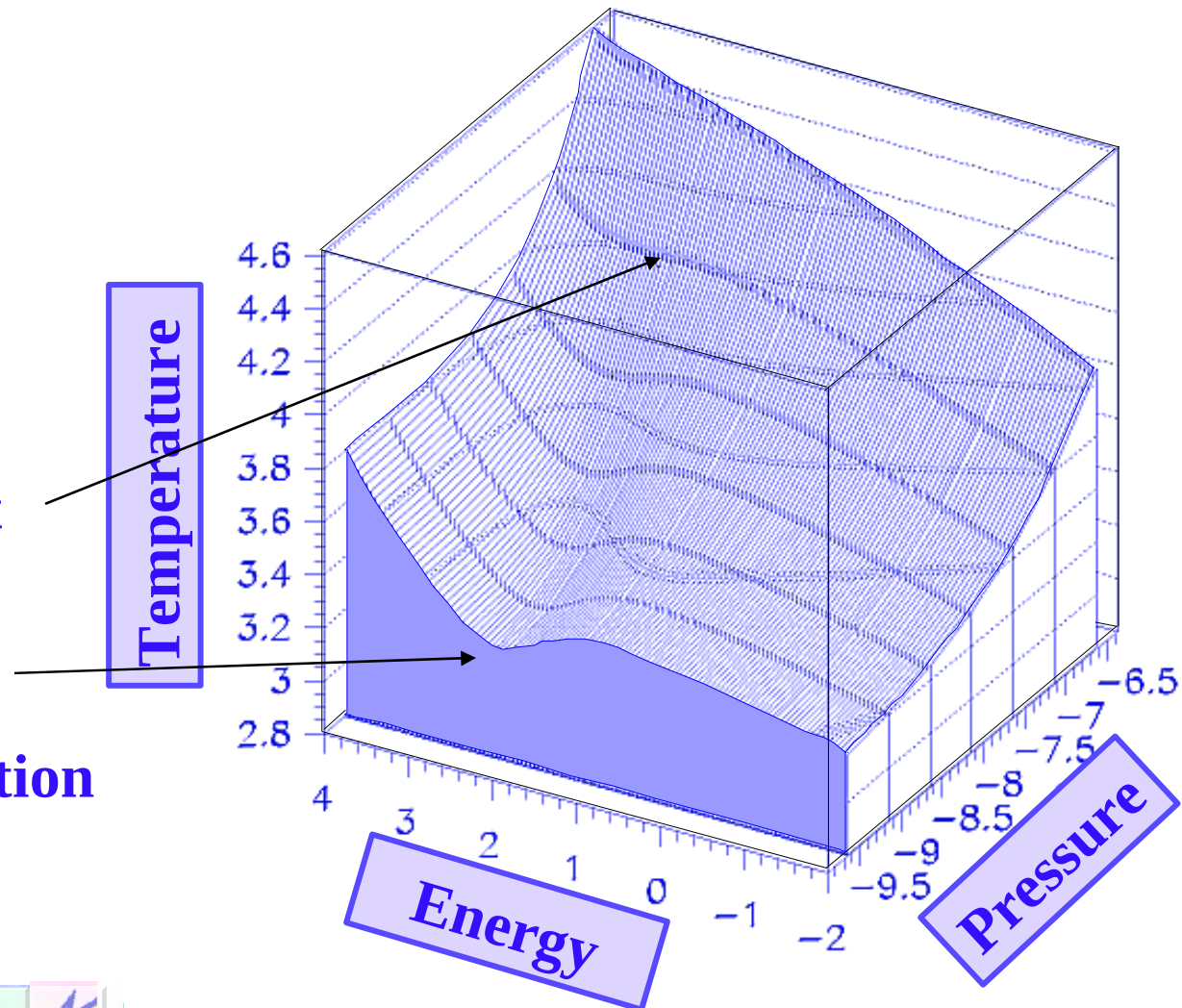
2-D EOS

$E \longrightarrow T$

$V \longrightarrow P$

■ Critical point

■ First order
Liquid gas
Phase Transition



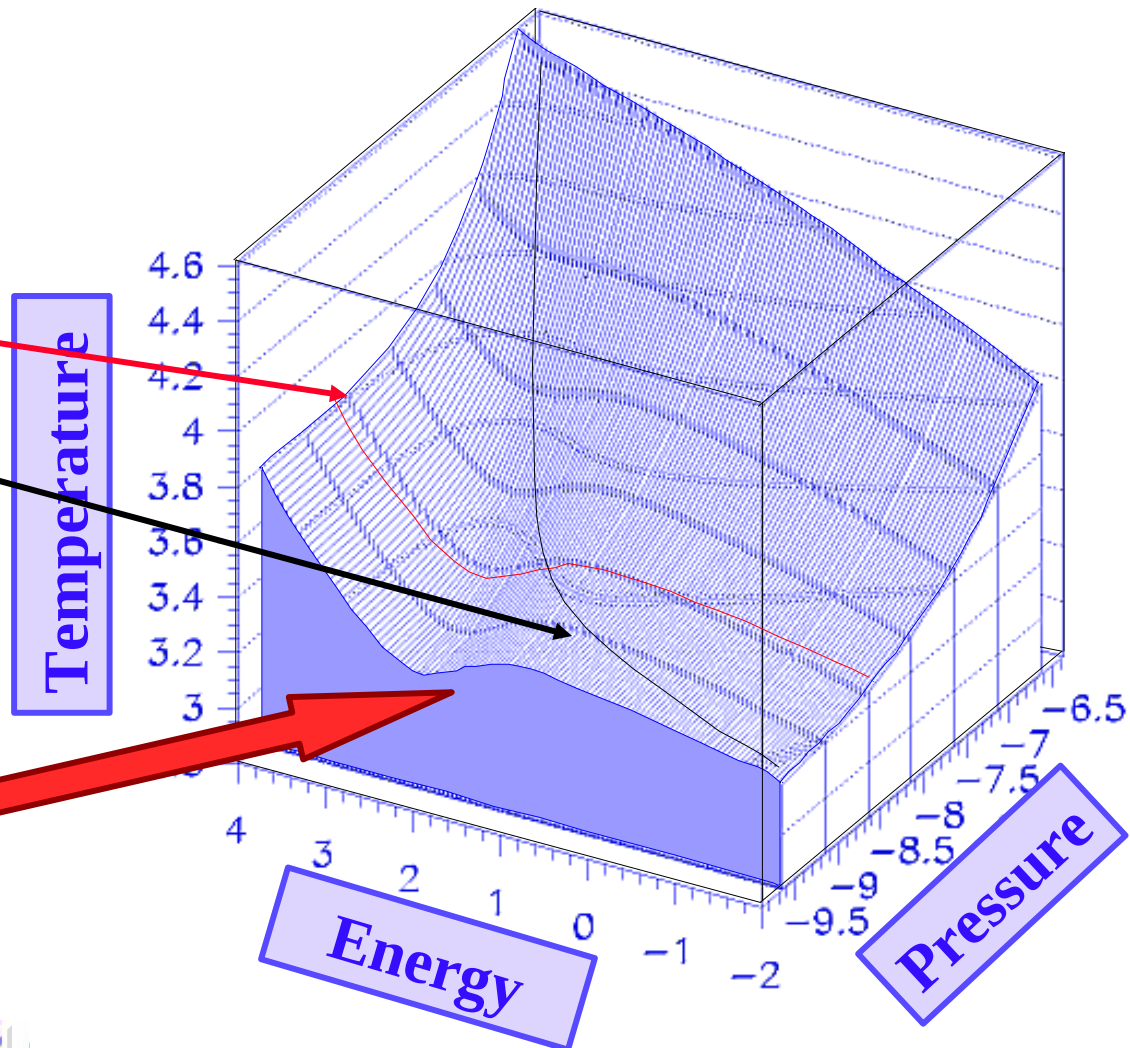
Volume dependent

Transformation dependent Caloric Curve

- $P = \text{cst}$

- $\langle V \rangle = \text{cst}$

Negative Heat Capacity



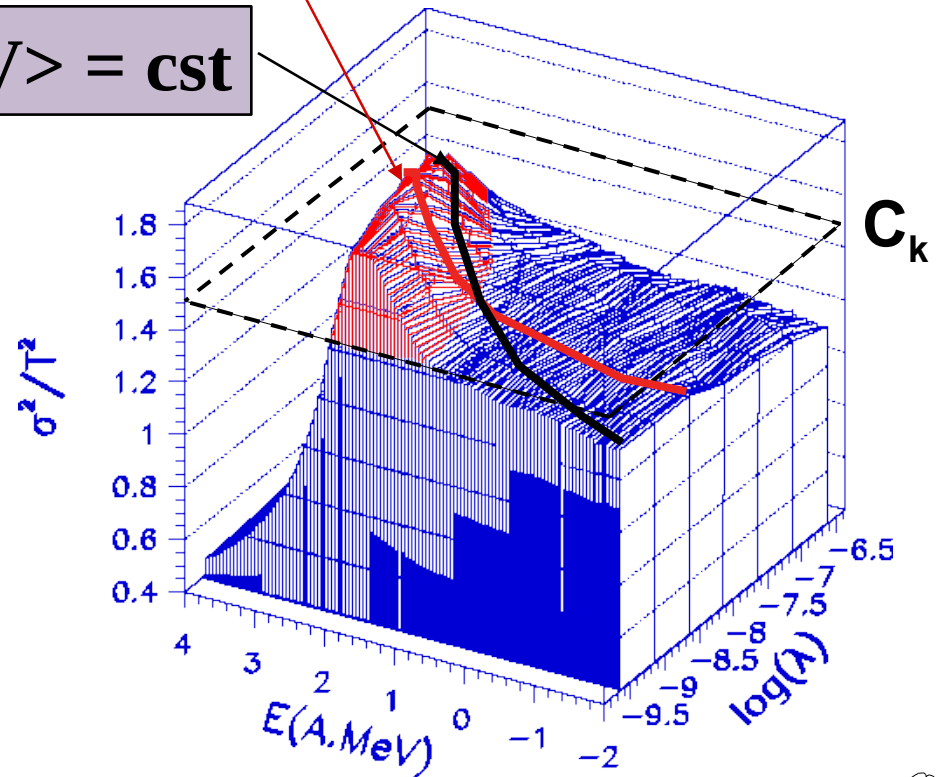
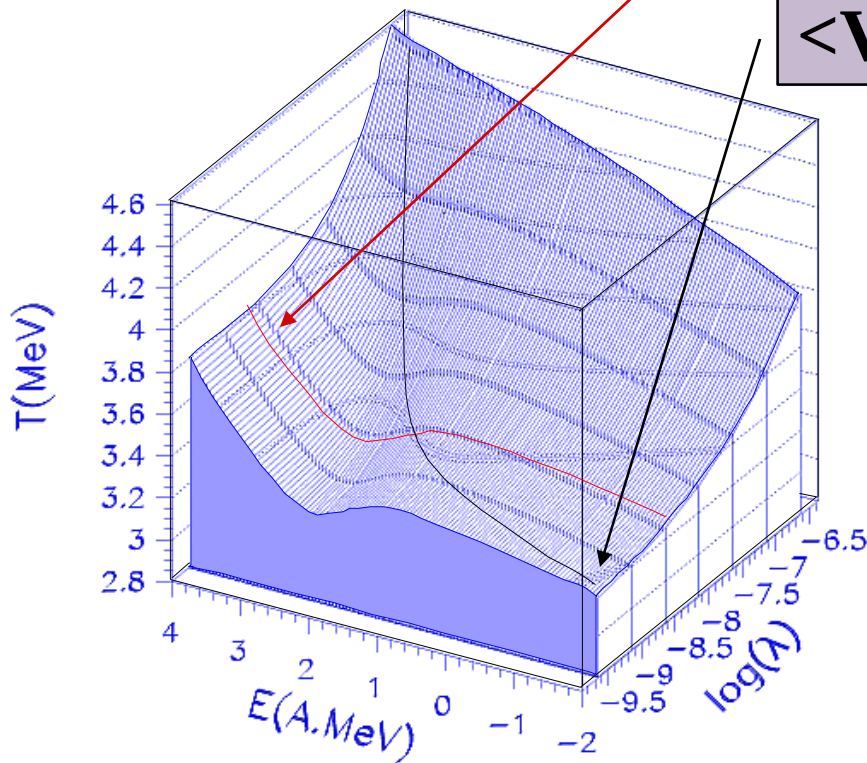
Volume dependent EOS

Temperature

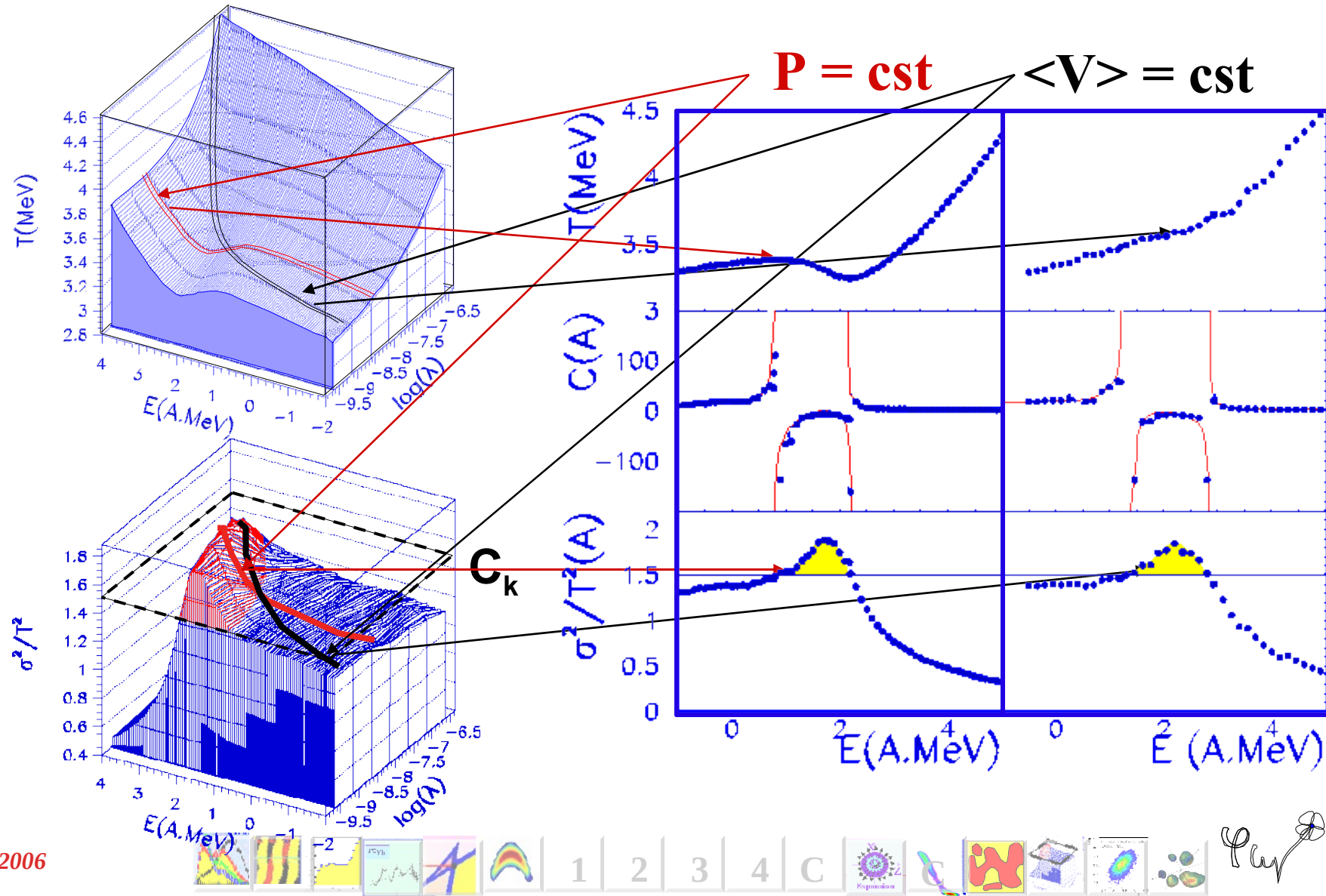
$$P = \text{cst}$$

Fluctuations

$$\langle V \rangle = \text{cst}$$

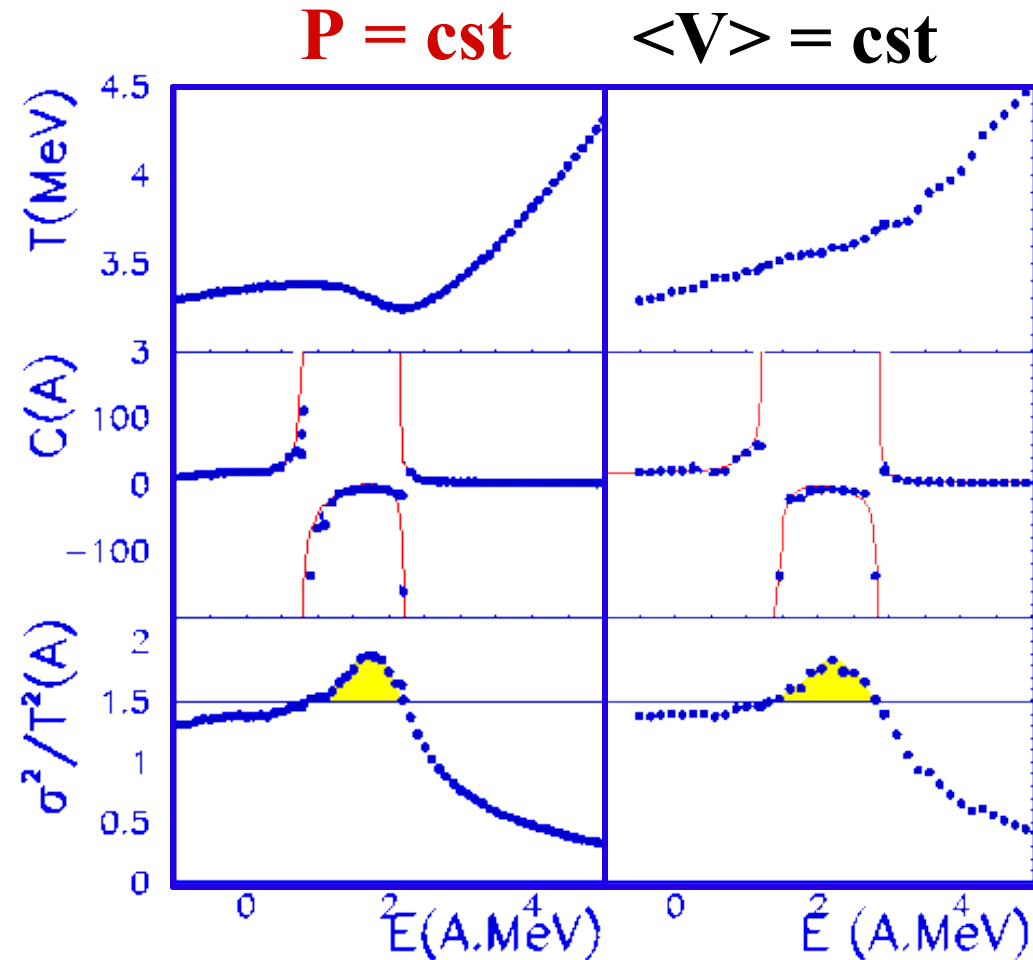


Volume dependent EOS



Volume dependent EOS

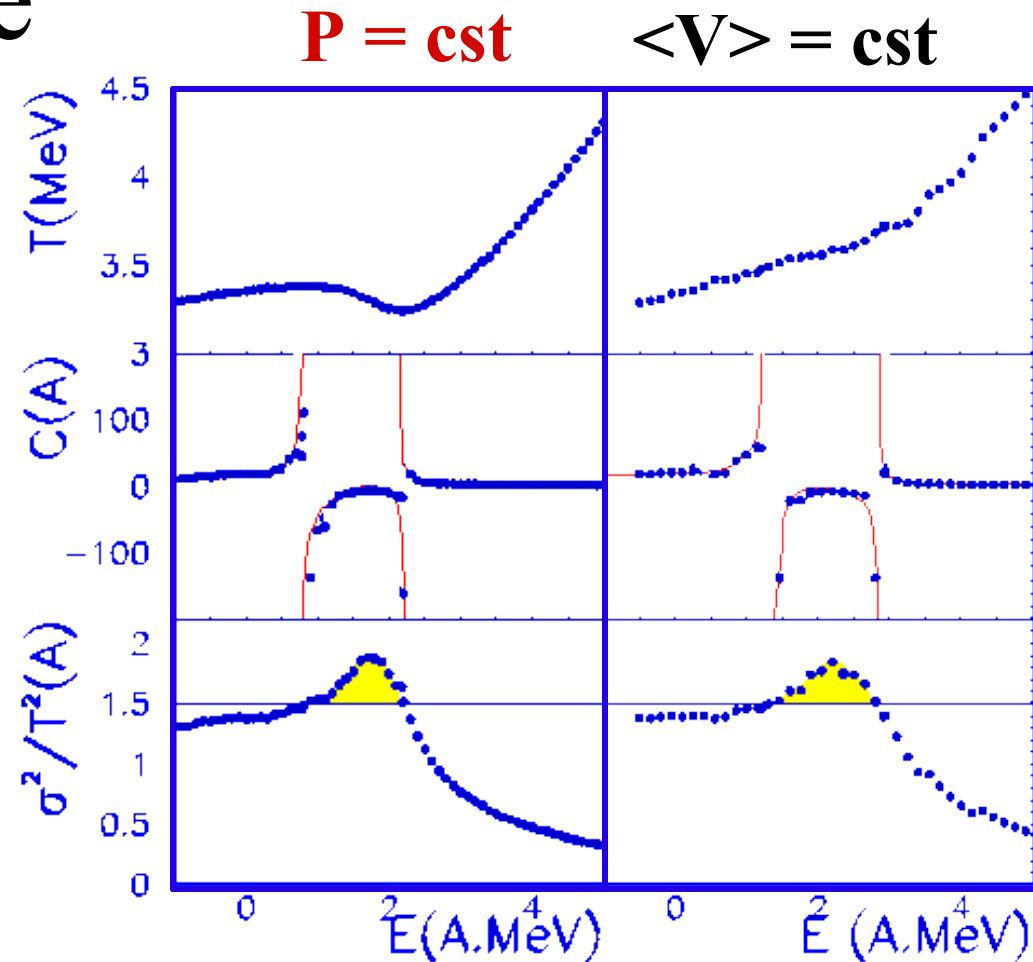
- The Caloric curve depends on the transformation
- C is not dT/dE
- σ^2 measures EOS Abnormal in Liquid-Gas



Caloric curve are not EOS

Fluctuations are

- The Caloric curve depends on the transformation
- C is not dT/dE
- σ^2 measures EOS
- σ^2 abnormal in Liquid-Gas



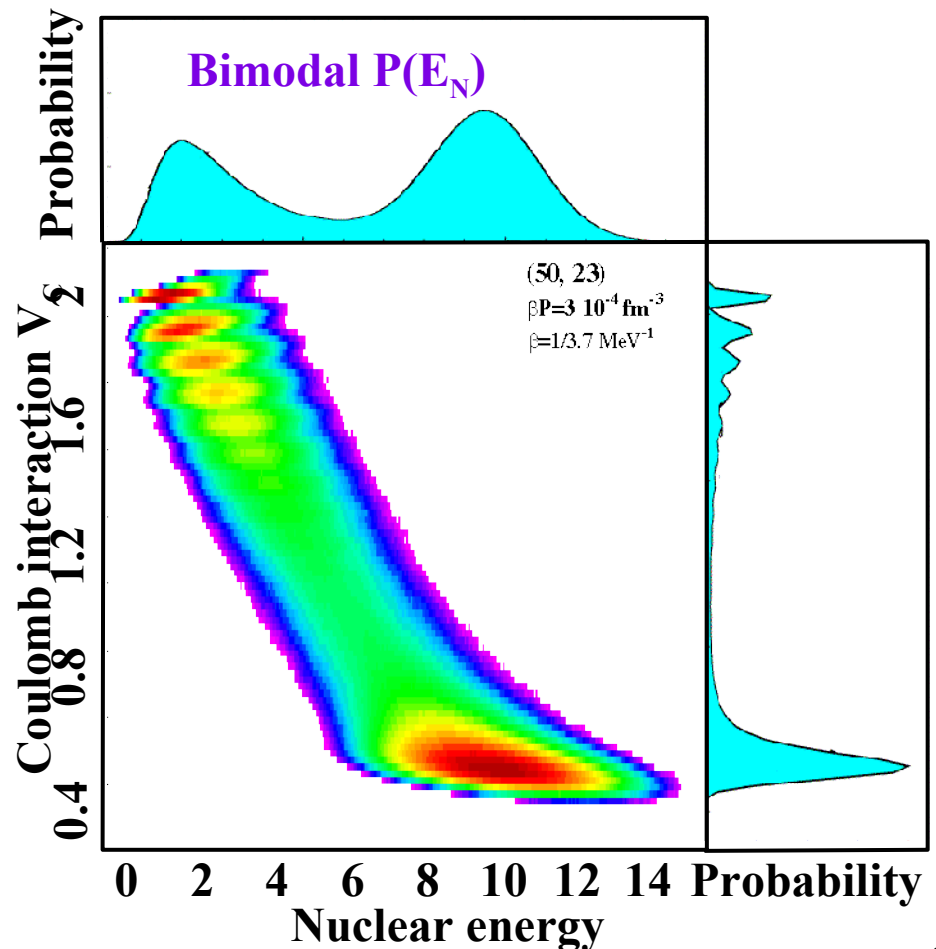
Negative curvatures

Ex: Statistical models

C<0 in statistical models

Liquid-gas and channel opening

-
- ◆ $q = 0 \Rightarrow$
no-Coulomb
- Many channel opening
- One Liquid-Gas transition (V_c related to V)



-Appendix -V-

How to progress

- Correcting errors in σ reconstruction
 - ◆ Iterative procedure
- C from other observables
 - ◆ Using correlated observables

Correction of experimental errors

■ Heat capacity from fluctuations



■ σ_{can} can be “filtered”



(apparatus and procedure)



Correction of experimental errors

Heat capacity from fluctuations



■ σ_{can} can be “filtered”



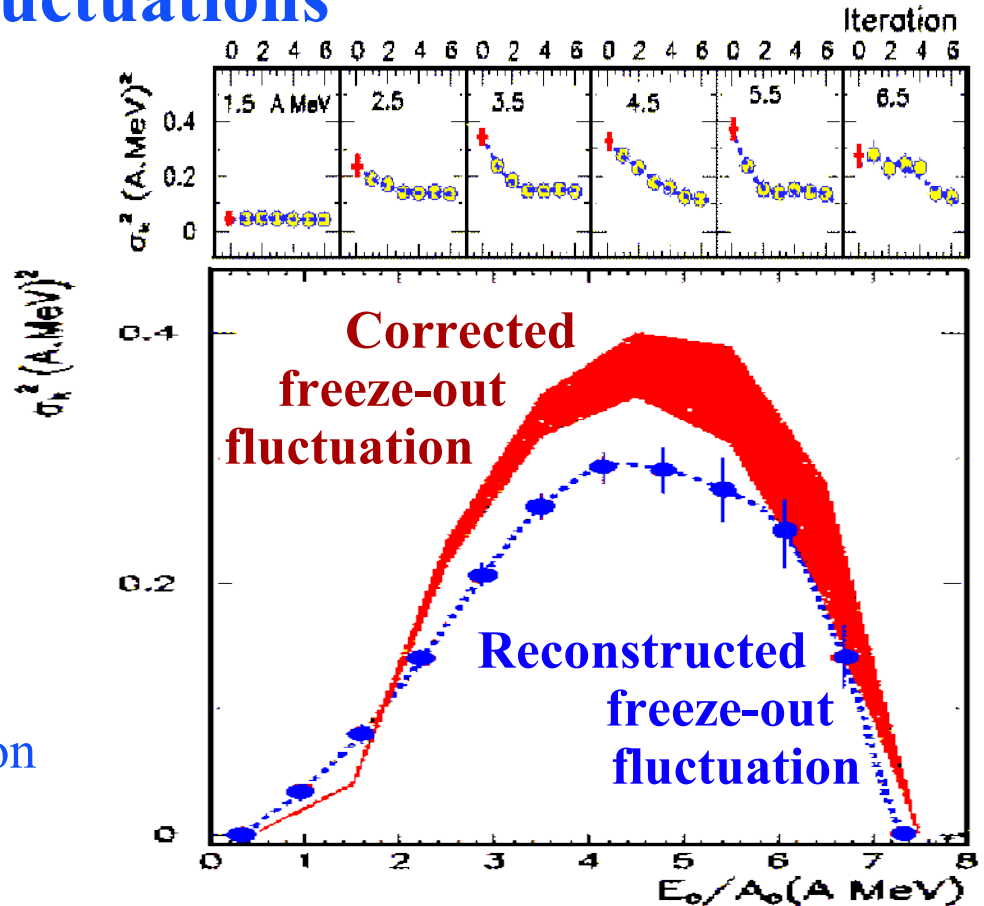
(apparatus and procedure)

■ σ_k can be corrected



Iterate the procedure

- ◆ Freeze-out reconstruction
- ◆ Decay toward detector



Heat capacity from any fluctuation

■ From kinetic energy



Heat capacity from any fluctuation

■ From kinetic energy

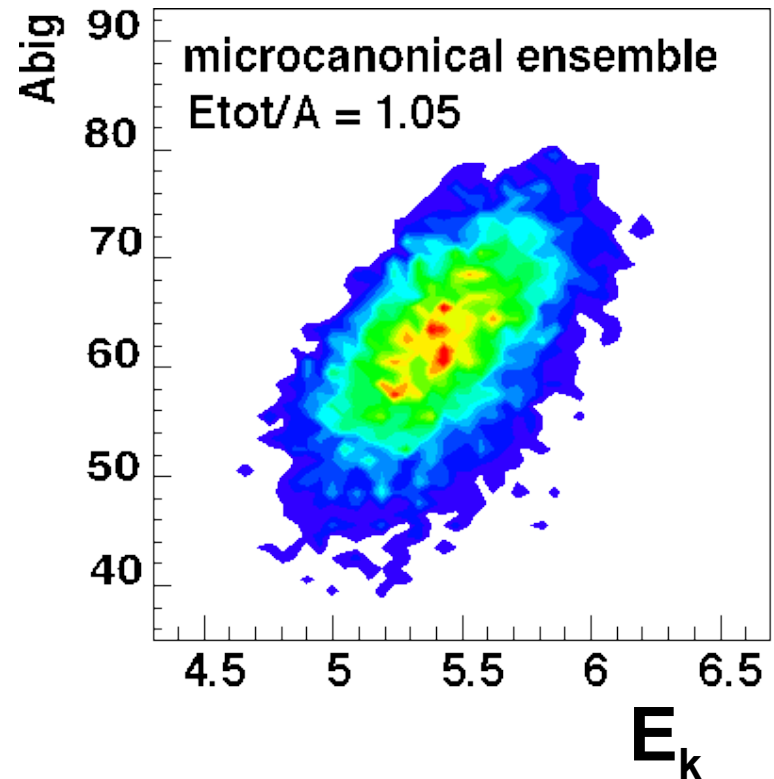


■ From any correlated observable (Ex. A_{big})

◆
$$\frac{X_K}{X} = 1 - \frac{\sigma_A^2}{\sigma_{A0}^2} \frac{1}{\rho^2}$$

◆ $\rho = \sigma_{kA} / \sigma_k \sigma_A$ correlation

◆ σ_{A0} “canonical” fluctuations



-Appendix -VI-

Test of freeze-out reconstruction

■ Volume

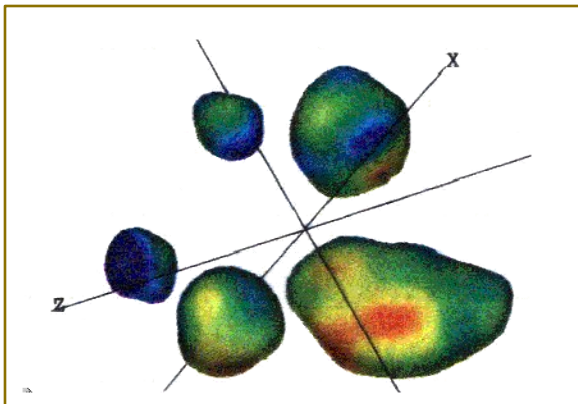
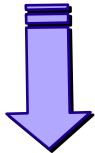
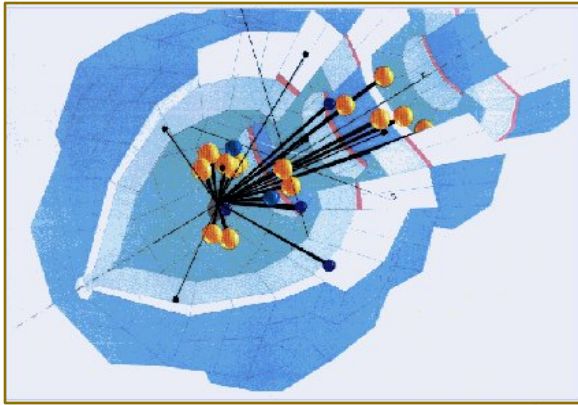
- ◆ Coulomb boost

■ Temperature

- ◆ Isotope ratios
- ◆ Particle-fragment correlations

■ Alternative extraction

Multifragmentation experiment



Sort events in energy
(Calorimetry)

Reconstruct a freeze-out
partition

1- Primary fragments: m_i

2- Freeze-out volume: E_{coul}

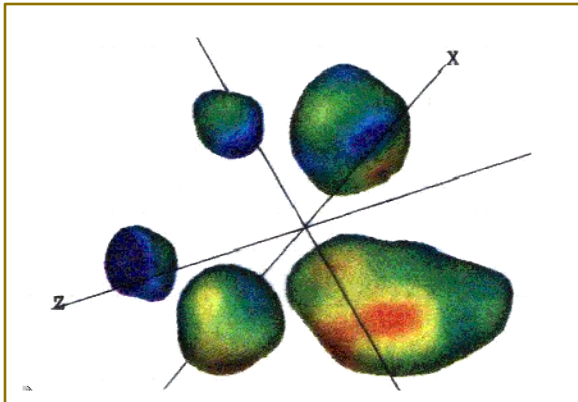
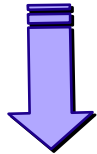
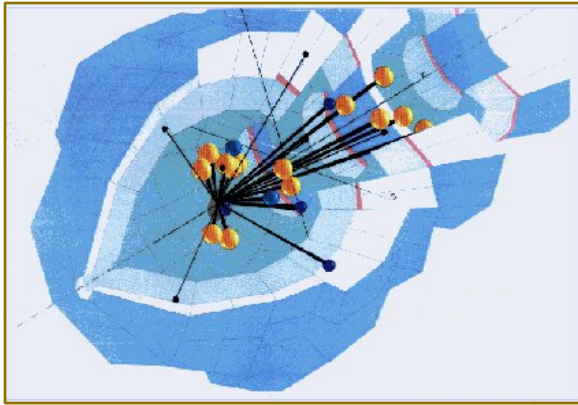
- $E_2 = \sum_i m_i + E_{\text{coul}}$

- $E_1 = E^* - E_2 \Rightarrow \langle E_1 \rangle, \sigma_1$

3- Kinetic EOS: T, C_1

- $\langle E_1 \rangle = \langle \sum_i a_i \rangle T^2 + 3/2 \langle M-1 \rangle T$

Multifragmentation experiment



Sort events in energy
(Calorimetry)

Reconstruct a freeze-out
partition

1- Primary fragments: m_i

2- Freeze-out volume: E_{coul}

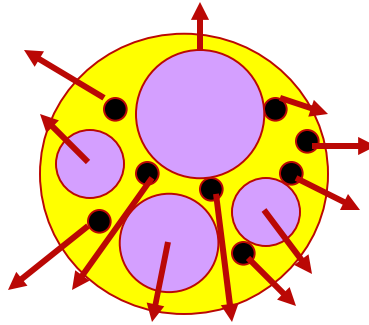
- $E_2 = \sum_i m_i + E_{\text{coul}}$

- $E_1 = E^* - E_2 \Rightarrow \langle E_1 \rangle, \sigma_1$

3- Kinetic EOS: T, C_1

- $\langle E_1 \rangle = \frac{\langle \sum_i a_i \rangle}{E^*} T^2 + 3/2 \langle M-1 \rangle T$

Volume

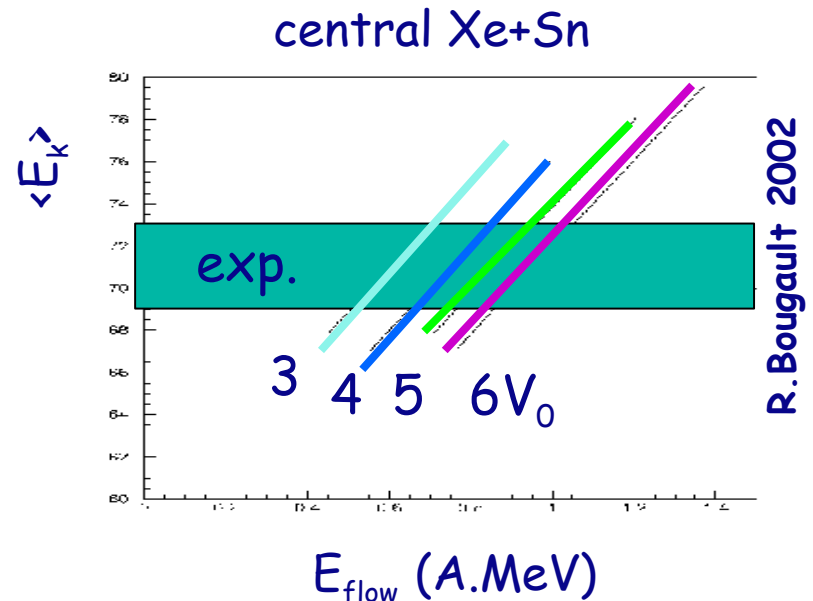
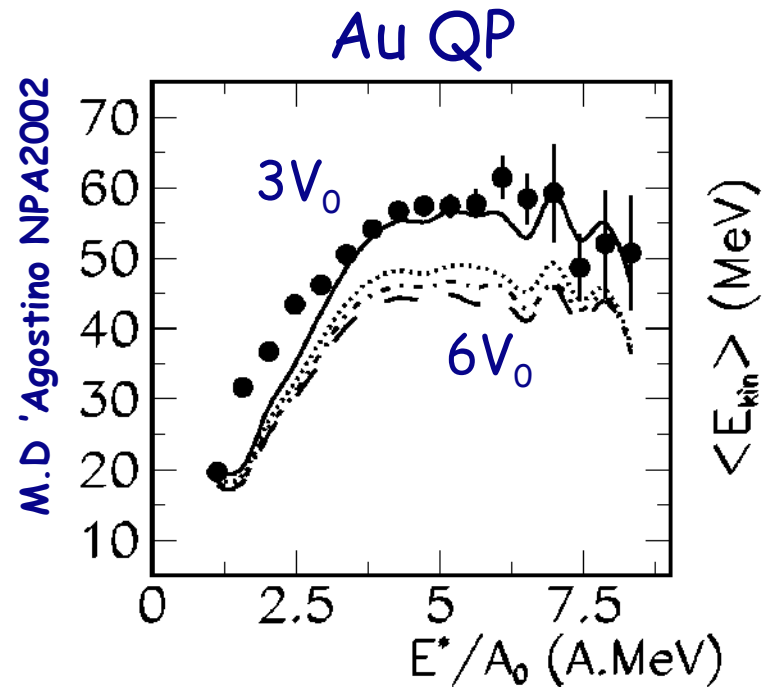
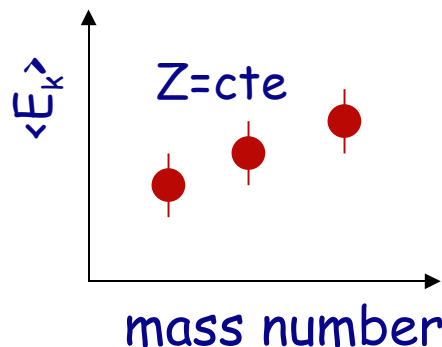


■ From Coulomb boost

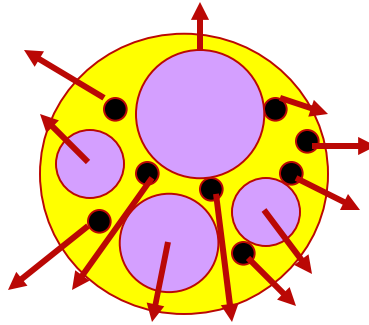
- ◆ C Weakly sensitive to V

■ Ambiguity with expansion

- ◆ Need isotopic resolution
- ◆ Tiny influence on C



Volume

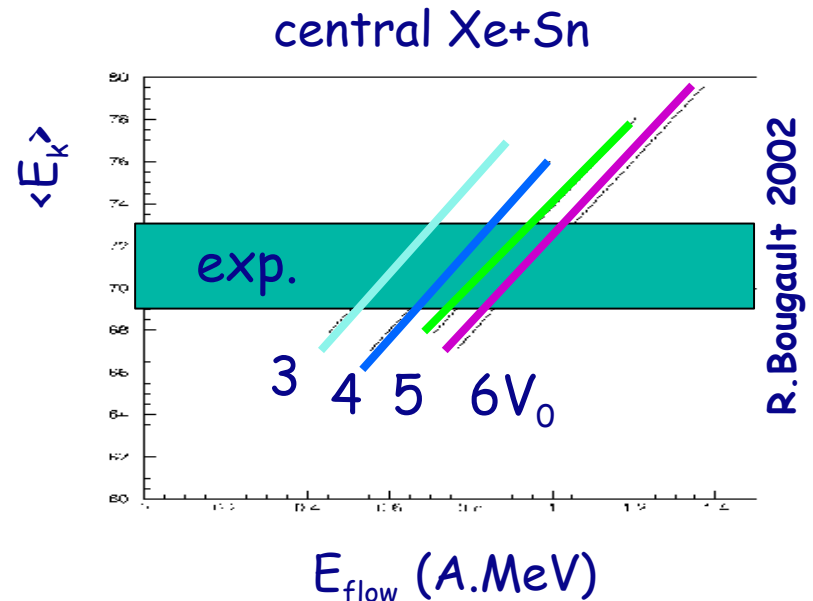
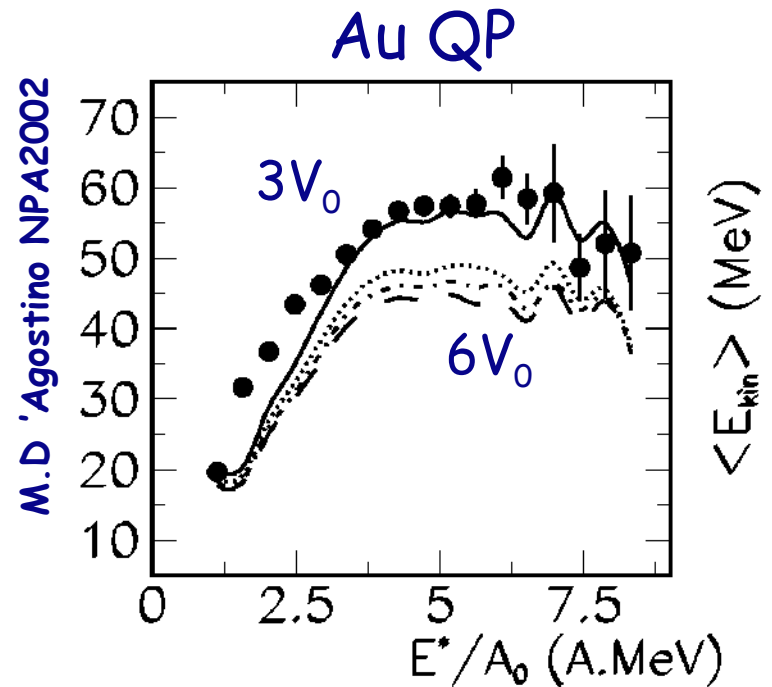
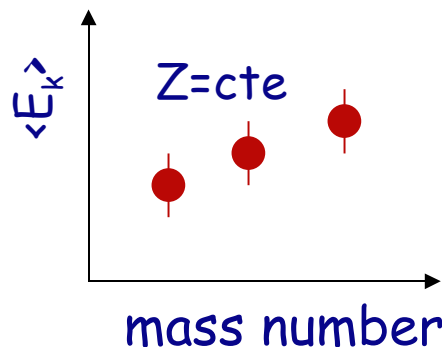


■ From Coulomb boost

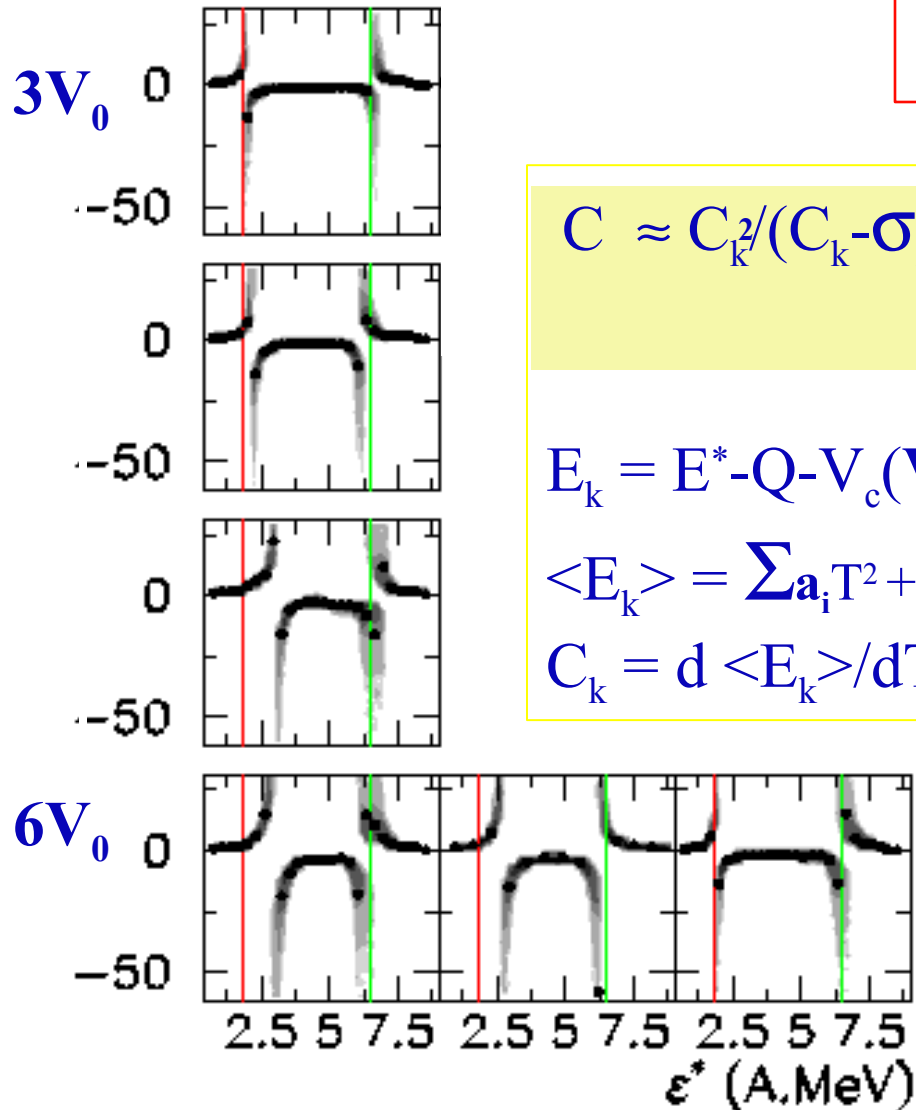
- ◆ C Weakly sensitive to V

■ Ambiguity with expansion

- ◆ Need isotopic resolution
- ◆ Tiny influence on C



Influence on C



a=12 → a=8

$$C \approx C_k^2 / (C_k - \sigma_k^2 / T^2)$$

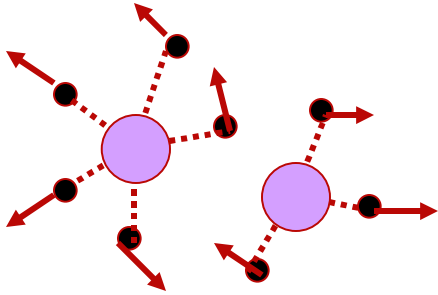
$$E_k = E^* - Q - V_c(V)$$

$$\langle E_k \rangle = \sum a_i T^2 + 3/2 (\langle m \rangle - 1) T$$

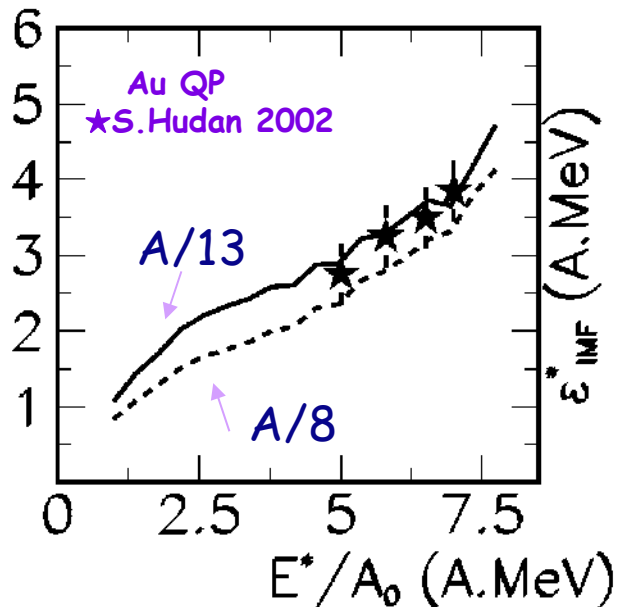
$$C_k = d \langle E_k \rangle / dT \quad C_k < 1.5 A_{\text{tot}}$$

Au+Au 35 A.MeV
MULTICS data

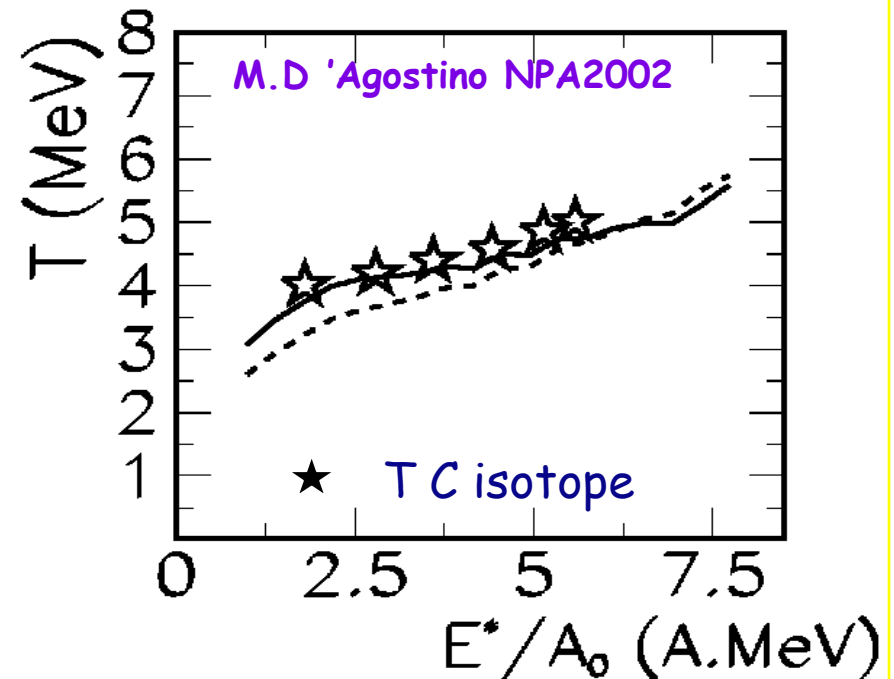
Evaporation and temperature



Correlation



$$\langle E_k \rangle = \sum a_i(T) T^2 + 3/2(\langle m \rangle - 1) T$$



Consistency of T

Heat capacity from the translational energy

$$T = \left\langle \frac{E_{tr}}{(3/2)m - 1} \right\rangle$$

$$\frac{1}{C} = 1 - T^2 \left\langle \frac{((3/2)m - 1)((3/2)m - 2)}{E_{tr}^2} \right\rangle$$

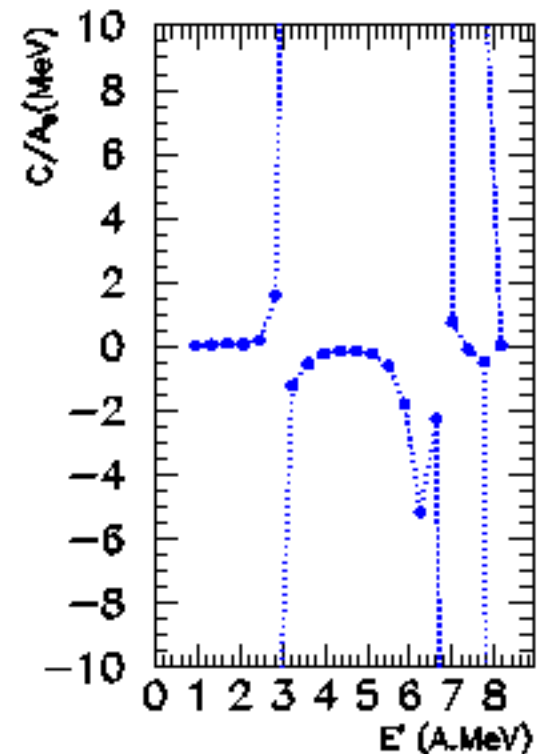
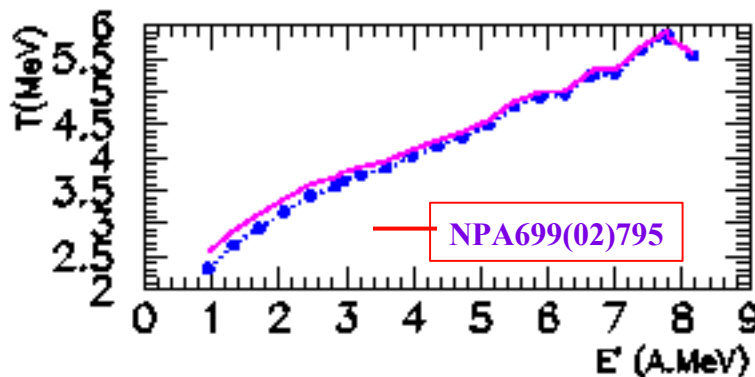
E.M.Pearson 1985, A.Raduta 2002

Coulomb trajectories

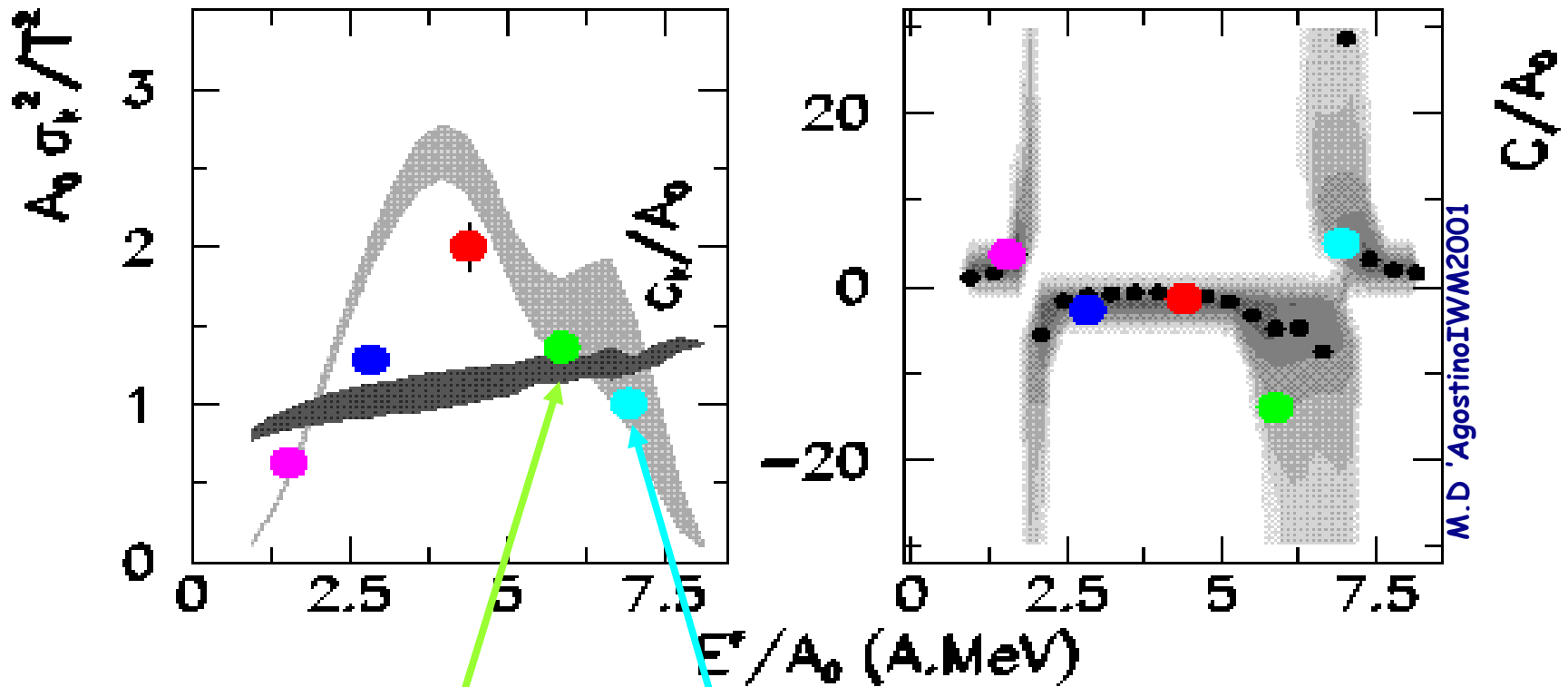
calorimetry

LCP-IMF correlations

Multics
Au+Au 35A.MeV



Comparison central/peripheral



Au+Au 35 $E_F=1$

Au+Au 35 $E_F=0$

Up to ≈ 35 A.MeV the flow ambiguity
is a small effect